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Prenylated Flavonoids from the Roots of Tephrosia rhodesica

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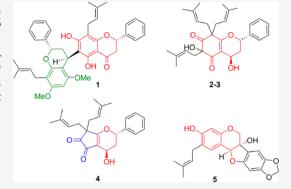
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ABSTRACT: Five new compounds—rhodimer (1), rhodiflavan A (2), rhodiflavan B (3), rhodiflavan C (4), and rhodacarpin (5)—along with 16 known secondary metabolites, were isolated from the $CH_2Cl_2-CH_3OH$ (1:1) extract of the roots of *Tephrosia rhodesica*. They were identified by NMR spectroscopic, mass spectrometric, X-ray crystallographic, and ECD spectroscopic analyses. The crude extract and the isolated compounds 2–5, 9, 15, and 21 showed activity (100% at 10 μ g and $IC_{50} = 5-15 \mu$ M) against the chloroquine-sensitive (3D7) strain of *Plasmodium falciparum*.



Tephrosia (Leguminosae) is a tropical and subtropical genus consisting of more than 400 species, 30 of which are native to Kenya. These have provided a variety of new flavonoids, including flavanones, flavones, chalcones, pterocarpans, and rotenoids, most of them prenylated and showing promising antiplasmodial, antifeedant, antileishmanial, estrogenic, antitumor, and antimicrobial activities. As part of our ongoing work on the genus Tephrosia, we now report the first phytochemical investigation of the roots of Tephrosia rhodesica, which showed antiplasmodial activity. The investigation of the root extract led to the isolation of five new compounds (1–5) and 16 known compounds (6–21). Herein, we describe their isolation, characterization, and activity against the chloroquine-sensitive (3D7) strain of Plasmodium falciparum.

RESULTS AND DISCUSSION

Applying silica gel gravity column chromatography, followed by Sephadex LH-20 gel filtration and preparative reverse-phase HPLC, 21 secondary metabolites were isolated from the roots of *T. rhodesica*. The structures of the isolated secondary metabolites were determined by NMR spectroscopic, mass spectrometric, and single-crystal X-ray diffractometric analyses. In addition to the five new secondary metabolites (1–5), the known tephrowatsin B (6), tephrinone (7), glabranin (8), uercetol B (9), mackiain (10), 6a-hydroxymaackiain (11), spisatin (12), tephrosin (13), fortenone (14), forthydroxyrotenone (15), forthydroxyrotenone (16), hiddecarpin (17), shydroxy-2-methoxy-8,9-methylenedioxypterocarpene (18), soliquirtigenin (19), popinitol (20), and tephrowatsin A (21) were identified by comparison of their observed and reported spectroscopic data (see pages \$22–\$34 in the Supporting Information).

Compound 1 was isolated as white crystals from CH₂Cl₂-CH₃OH (1:1) solution. Its molecular formula was deduced as $C_{42}H_{44}O_7$ based on HREIMS ([M]⁺ m/z 660.3095, calcd 660.3087) and NMR data analyses (see Table 1, as well as Figures S1–S6 in the Supporting Information). The ¹H NMR signals at $\delta_{\rm H}$ 5.43 (H-2), 5.06 (H-2'''), 2.84 and 3.07 (H-3a-b), 2.18 and 2.31 (H-3'''), and 4.66 (H-4'''), along with the ¹³C NMR signals at δ_{C} 78.6 (C-2), 75.4 (C-2'''), 43.7 (C-3), 36.9 (C-3'''), 196.3 (C-4), and 26.8 (C-4''') were consistent with a flavanone-flavan dimer core structure^{22,23} that was corroborated by the UV absorptions of λ_{max} = 230, 290, and 350 nm. The NMR data (Table 1) indicated the presence of two sets of unsubstituted aromatic rings, two sets of prenyl groups, two hydroxy, and two methoxy groups. This is consistent with a 5hydroxyflavanone moiety possessing an unsubstituted ring B, along with a ring A substituted with prenyl at C-8 ($\delta_{
m C}$ 108.1) and two hydroxy groups, $\delta_{\rm H}$ 12.61 (HO-5) and 6.80 (HO-7) at C-5 ($\delta_{\rm C}$ 159.3) and C-7 ($\delta_{\rm C}$ 162.8). Accordingly, C-6 ($\delta_{\rm C}$ 109.8) connects the flavanone to the flavan moiety. Whereas the placement of a hydroxy group at C-7 was based on biogenetic considerations,²⁴ and on the chemical shift of the HO-5 hydrogen ($\delta_{\rm H}$ 12.61), suggesting its involvement in an intramolecular hydrogen bond; the placement of the prenyl group at C-8 ($\delta_{\rm C}$ 108.1) rather than at C-6 ($\delta_{\rm C}$ 109.8) was

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Table 1. NMR Spectroscopic Data (800 MHz, CDCl₃) for Rhodimer (1)

position	$\delta_{ extsf{C}, ext{ type}}$	$\delta_{ m H}$ (J, in Hz)	HMBC
2	78.6, CH	5.43, dd (2.9, 13.1)	C-4, C-1', C-2'/6', C-8a
3	43.7, CH ₂	2.84, dd (2.9, 17.1) 3.07, dd (13.1, 17.1)	C-2, C-4, C-1', C-4, C-4a, C-1'
4	196.3, C=O		
4a	100.4, C		
5	159.3, C-O		
6	109.8, C		
7	162.8, C-O		
8	108.1, C		
8a	157.6, C-O		
1'	139.1, C		
2'/6'	125.9, CH	7.47, m	C-2, C-2'/6', C-3'/5'
3'/5'	128.7, CH	7.43, m	C-1', C-3'/5'
4'	128.5, CH	7.39, m	C-2'/6'
1''	21.8, CH ₂	3.18, br t	C-7, C-8, C-8a, C-3'', C-2''
2''	122.5, CH	5.15, br t	C-1'', C-4'', C-5''
3''	131.6, C	3.13, 51 (01,01,03
4''	17.8, CH ₃	1.57, s	C-3'', C-2'', C-5'
5''	25.8, CH ₃	1.65, s	C-3'', C-2'', C-4'
OH-5	20.0, 0113	12.61, s	C-5, C-6, C-4a, C-4, C-7
OH-7		6.80, s	C-6, C-8, C-8a, C-7, C-4a
2'''	75.4, CH	5.06, dd (2.0, 11.5)	C-4''', C-1'''', C-2''''/6'''', C-8a
3'''	36.9, CH ₂	2.18, dt (2.0, 6.0, 14.0) 2.31, dt (2.0, 11.5, 14.0)	C-1'''', C-4''', C-6, C-4''', C-4'''a, C-6
3 4'''	26.8, CH	4.66, dd (2.0, 6.0)	C-2''', C-3''', C-4'''a, C-5''', C-8'''a, C-5, C-6, C-
4′′′a	102.8, C	4.00, dd (2.0, 0.0)	C-2 , C-3 , C-4 a, C-3 , C-6 a, C-3, C-0, C-
4 a 5'''			
5 6'''	157.5, C-O	(17	C-4''', C-4'''a, C-5''', C-7''', C-8''', C-1''''' (w)
o 7'''	88.4, CH	6.17, s	C-4 , C-4 a, C-5 , C-7 , C-8 , C-1 (w)
8'''	158.6, C-O		
	110.5, C		
8′′′a	154.3, C-O		
1''''	141.4, C		
2''''/6''''	126.0, CH	7.38, m	C-2''', C-2''''/6'''', C-3''''/5''''
3''''/5''''	128.3, CH	7.34, m	C-3''''/5'''', C-1''''
4''''	127.6, CH	7.29, m	C-2''''/6''''
1''''	21.9, CH ₂	3.35, br t	C-8''', C-8a''', C-7''', C-3''''', C-2'''''
2''''	123.2, CH	5.24, br t	C-1'''', C-4'''', C-5''''
3''''	130.8, C		
4''''	17.8, CH ₃	1.66, s	C-2''''', C-3'''''', C-5'''''
5''''	25.9, CH ₃	1.68, s	C-2'''', C-3''''', C-4'''''
OMe-5'''	55.9, C-O	3.73, s	C-5'''
OMe-7'''	55.8, C-O	3.87, s	C-7'''

based on the HMBC correlation of CH $_2$ -1'' ($\delta_{\rm H}$ 3.18) with C-7 ($\delta_{\rm C}$ 162.8), C-8 ($\delta_{\rm C}$ 108.1), C-8a ($\delta_{\rm C}$ 157.6), C-3'' ($\delta_{\rm C}$ 131.6), C-2'' ($\delta_{\rm C}$ 122.5), and of H-2 ($\delta_{\rm H}$ 5.43) with C-4 ($\delta_{\rm C}$ 196.3), C-1' ($\delta_{\rm C}$ 139.1), C-2'/C-6' ($\delta_{\rm C}$ 125.9), C-8a ($\delta_{\rm C}$

157.6). The identity of the other half of the molecule was established as a 5,7-dimethoxy-8-prenylflavan moiety based on the NMR data (Table 1). Hence, its ¹H NMR spectrum revealed the presence of an unsubstituted aromatic ring (ring

4'''')], two methoxy groups at $\delta_{\rm H}$ 3.73 (OMe-5''') and 3.87 (OMe-7''') and a prenyl side chain $[\delta_H 3.35 (H-1''''), 5.24]$ (H-2''''), 1.66 (H-4''''), and 1.68 (H-5'''')], and a singlet at $\delta_{\rm H}$ 6.17 (H-6'''). The placement of the two methoxy groups was established based on the HMBC correlation of H-6 $^{\prime\prime\prime}$ ($\delta_{\rm H}$ 6.17) with C-4''' ($\delta_{\rm C}$ 102.8), C-5''' ($\delta_{\rm C}$ 157.5), and C-7''' ($\delta_{\rm C}$ 158.6). The prenyl group was positioned at C-8''' ($\delta_{\rm C}$ 110.5) based on the HMBC cross-peaks of H_2 -1'''' (δ_H 3.35) to C-8''' ($\delta_{\rm C}$ 110.5), C-8a''' ($\delta_{\rm C}$ 154.3), and C-7''' ($\delta_{\rm C}$ 158.6). Methoxy groups were placed at C-5 $^{\prime\prime\prime}$ ($\delta_{\rm C}$ 157.5) and C-7 $^{\prime\prime\prime}$ $(\delta_C 158.6)$ upon biogenetic considerations, ²⁴ as well as based on the HMBC correlations of MeO-5''' ($\delta_{\rm H}$ 3.73) to C-5''' ($\delta_{\rm C}$ 157.5), and of MeO-7''' ($\delta_{\rm H}$ 3.87) to C-7''' ($\delta_{\rm C}$ 158.6). This assignment was corroborated by the NOEs of MeO-5''' $(\delta_{\rm H}~3.73)$ and MeO-7''' $(\delta_{\rm H}~3.87)$ with H-6''' $(\delta_{\rm H}~6.17)$. The C-6 ($\delta_{\rm C}$ 109.8) to C-4''' ($\delta_{\rm C}$ 26.8) linkage of the flavanone and the flavan moieties was confirmed by the HMBC correlations of H-4''' ($\delta_{\rm H}$ 4.66) with C-5 ($\delta_{\rm C}$ 159.3), C-6 ($\delta_{\rm C}$ 109.8), and C-7 ($\delta_{\rm C}$ 162.8), and of H₂-3''' ($\delta_{\rm H}$ 2.18 and 2.31) with C-6 ($\delta_{\rm C}$ 109.8). Because of its three stereocenters at C-2, C-2", and C-4''', 1 has eight possible stereoisomers. The large ${}^{3}J_{2,3} = 13.1$ Hz indicated H-2 to be axial, and thereby ring B to be equatorial. Based on the available literature on the flavanones of this genus, C-2 is expected to be S-configured. 1,25 Similarly, the large ${}^{3}J_{2''',3'''}=11.5$ Hz suggests H-2''' to be axial, and accordingly ring B of the flavan moiety to be equatorially oriented. Furthermore, based on the literature, 25 C-2" is also expected to be S-configured as in the flavanones and flavans of this genus. The ${}^{3}J_{3a''',4'''}=2.0$ suggests the axial-equatorial or equatorial-equatorial orientation of the involved protons, whereas ${}^{3}J_{3b''',4'''} = 6.0$ Hz gauche orientation, suggesting H-4''' to be axially oriented and, consequently, the flavan moiety to occupy the more favorable equatorial orientation with C-4''' being R-configured.²⁵ The NOE correlation of H-2''' ($\delta_{\rm H}$ 5.06) and HO-7 ($\delta_{\rm H}$ 6.80) reveals the flavanone and H-2''' to be oriented in the same direction. This absolute configuration (2S,2"'S,4"'R), proposed on the basis of biogenetic considerations and NMR analyses was subsequently confirmed by single-crystal X-ray crystallography (Figure 1). Thus, based on the above spectroscopic data, this new compound, rhodimer (1), was characterized as (2S,2"'S,4R)-5,7-dihydroxy-5''',7'''-dimethoxy-8,8'''-bis(3-methylbut-2-en-1-yl)-2',2'''-diphenyl-[4''',6-bichroman]-4-one.

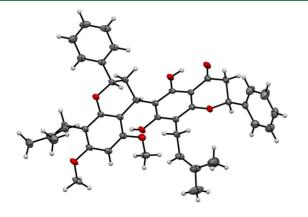


Figure 1. Single crystal X-ray structure of compound 1 with the thermal displacement parameters at 40% probability level.

Compound 2 was isolated as an oily paste and was assigned the molecular formula $C_{30}H_{38}O_5$, based on HRESIMS ([M + H]⁺ m/z 479.2493, calcd 479.2753) and NMR data analyses (see Table 2, as well as Figures S9-S14 in the Supporting Information). The 1 H NMR signals at $\delta_{\rm H}$ 5.22 (H-2), 1.80 (H- 3_{ax}), 2.31 (H- 3_{eq}), and 4.99 (H-4), along with the $^{13}\mathrm{C}$ NMR signals at $\delta_{\rm C}$ 76.1 (C-2), 37.0 (C-3), and 57.4 (C-4), suggested it to be a flavan-4-ol derivative, which was corroborated by the UV absorptions $\lambda_{\text{max}} = 240$ and 290 nm. The characteristic flavan-4-ol ring C of 2 was confirmed by the HMBC correlations of H-4 ($\delta_{\rm H}$ 4.99) with C-2 ($\delta_{\rm C}$ 76.1), C-3 ($\delta_{\rm C}$ 37.0), C-5 ($\delta_{\rm C}$ 194.7), C-4a ($\delta_{\rm C}$ 114.9), and C-8a ($\delta_{\rm C}$ 171.0), and by the COSY correlations of H-2 ($\delta_{\rm H}$ 5.22) with CH₂-3 ($\delta_{\rm H}$ 2.31 and 1.80), and of the latter with H-4 ($\delta_{\rm H}$ 4.99). The six methyl (H₃-4", H₃-4", H₃-4", H₃-5", H₃-5", and H₃-5'''), along with the three methine signals (H-2", H-2", and H-2'''') were diagnostic for three prenyl groups. The NMR signals of H-2'/6', H-3'/5', and H-4' (Table 2) further suggested ring B to be unsubstituted and, hence, all three prenyl groups to be connected to ring A. This ring is oxygenated at C-5 ($\delta_{\rm C}$ 194.7) and C-7 ($\delta_{\rm C}$ 206.4), with their placement having been derived from biogenetic considerations. An additional oxygenation is presented at C-6 (δ_C 83.3), which uncharacteristically is an sp³ hybridized carbon. Prenyl substitution at this carbon was confirmed by the HMBC correlation of H-2'''' ($\delta_{\rm H}$ 4.99) with C-6 ($\delta_{\rm C}$ 83.3). The location of the other two prenyl groups at C-8 ($\delta_{\rm C}$ 58.3), also sp³ hybridized, was established from the ³*J* HMBC correlations of H-2'' ($\delta_{\rm H}$ 5.03, m) and H-2''' ($\delta_{\rm H}$ 4.66, m) with C-8 ($\delta_{\rm C}$ 58.3). Moreover, CH₂–1 $^{\prime\prime}$ ($\delta_{\rm H}$ 2.60 and 2.81) and CH₂–1 $^{\prime\prime}$ ($\delta_{\rm H}$ 2.46 and 2.77) showed HMBC cross-peaks with C-7 ($\delta_{\rm C}$ 206.4) and C-8a ($\delta_{\rm C}$ 171.0), whereas CH₂-1'''' ($\delta_{\rm H}$ 2.71 and 2.25) with C-5 ($\delta_{\rm C}$ 194.7), confirming that ring A is fully substituted. Based on the above data, compound 2 was characterized as 4,6-dihydroxy-6,8,8-tris(3-methylbut-2-en-1yl)-2-phenyl-2,3,4,8-tetrahydro-5*H*-chromene-5,7(6*H*)-dione. Its configurational assignment is discussed below, together with that of the structurally closely related compound 3.

Compound 3 was isolated as an oily paste and was assigned the molecular formula $\rm C_{30}H_{38}O_5$ based on HRESIMS ([M + H]⁺ m/z 479.2493, calcd 479.2753) and NMR data (see Table 2, as well as Figures S16–S21 in the Supporting Information). The $^{\rm 1}H$ NMR signals at $\delta_{\rm H}$ 5.24 (H-2), 2.00 and 2.30 (CH₂–3), and $\delta_{\rm H}$ 4.55 (H-4) and $^{\rm 13}{\rm C}$ NMR signals at $\delta_{\rm C}$ 76.2 (C-2), 36.5 (C-3), and 59.5 (C-4) suggested 3 to be a flavan derivative with identical core structure, 4,6-dihydroxy-6,8,8-tris(3-methylbut-2-en-1-yl)-2-phenyl-2,3,4,8-tetrahydro-5*H*-chromene-5,7(6*H*)-dione to compound 2. The highly similar NMR data of 2 and 3 (Table 2) indicated these compounds to be stereoisomers.

Compounds 2 ($[\alpha]_D$ +66.1) and 3 ($[\alpha]_D$ +17.5) are both dextrorotatory (for details, see the Experimental Section) and, hence, are diastereomers (as enantiomers would be expected to rotate plan-polarized light in opposite directions). The large coupling constant ${}^3J_{2,3ax}$ (12.6 Hz in both 2 and 3) is consistent with H-2 being axial and ring B of both compounds to occupy the more favorable equatorial orientation. The ${}^3J_{3ax,4}$ (4.2 Hz for 2, and 3.7 Hz for 3) and ${}^3J_{3eq,4}$ (1.9 Hz in 2, 2.3 Hz in 3) are small, suggesting H-4 to be equatorial and hence HO-4 to be axial in both compounds, according to Pouget et al. This is further corroborated by the 3.4 Å H-2–H-4 interatomic distance of 3, estimated based on the NOE cross-peak intensities using H-3a/b as an internal reference (1.8 Å; see

Table 2. NMR Spectroscopic Data (800 MHz, CDCl₃) for Rhodiflavan A (2) and Rhodiflavan B (3)

	2			3		
position	δ_{C} , type	$\delta_{ ext{H, }}$ m $(J ext{ in Hz})$	HMBC	$\delta_{\rm C}$, type	$\delta_{\rm H,}$ m (J in Hz)	НМВС
2	76.1, CH	5.22, dd (12.6, 2.3)	C-3, C-4, C-1', C-2', C-6'	76.2, CH	5.24, dd (12.6, 2.3)	C-3, C-4, C-1', C-2', C-6'
3	37.0, CH ₂	2.31, ddd (14.6, 2.3, 1.9)	C-4, C-4a	36.5, CH ₂	2.30, dt (14.6, 2.3, 2.0)	C-4, C-4a
		1.80, ddd (14.6, 12.6, 4.2)	C-1', C-2		2.00, ddd (14.6, 12.6, 3.6)	C-2, C-1'
4	57.4, CH	4.99, m	C-2, C-3, C-5, C-4a, C-8a	59.5, CH	4.55, dd (2.0, 3.6)	C-2, C-4a, C-8a, C-5
4a	114.9, C			114.2, C		
5	194.7, C-O			197.0, C-O		
6	83.3, С-ОН			83.4, С–ОН		
7	206.4, C=O			207.2, C=O		
8	58.3, C			58.3, C		
8a	171.0, C-O			171.0, C-O		
1'	138.9, C			139.2, C		
2', 6'	125.9, CH	7.40, m	C-2, C- 2', C-4', C-6'	125.9, CH	7.40, m	C-2, C-4', C-2', C-6'
3', 5'	128.7, CH	7.43, m	C-1', C-3', C-5'	128.7, CH	7.43, m	C-1', C-3', C-5'
4′	128.5, CH	7.39, m	C-1', C-2', C-6'	128.5, CH	7.39, m	C-1', C-2', C-6'
1′′	35.0, CH ₂	2.81, dd (14.2, 9.2)	C-7, C-8, C-8a, C-2", C-3", C-1"	34.5, CH ₂	2.89, dd (14.2, 9.8)	C-7, C-8, C-8a, C-2'', C-3'', C-1'''
		2.60, dd (6.3, 14.2)	C-7, C-8, C-8a, C-2", C-3", C-1"		2.59, m	C-7, C-8, C-8a, C-2'', C-3'', C-1'''
1'''	38.5, CH ₂	2.77, m	C-7, C-8, C-8a, C-2''', C-3''', C-1''	38.5, CH ₂	2.69, dd (13.5, 7.3)	C-7, C-8, C-8a, C-2''', C-3''', C-1''
		2.46, dd (13.7, 8.2)	C-7, C-8, C-8a, C-2''', C-3''', C-1''		2.43, dd (13.5, 8.7)	C-7, C-8, C-8a, C-2''', C-3''', C-1''
1''''	38.5, CH ₂	2.71, dd (14.9, 6.3)	C-6, C-5, C-2''''	38.2, CH ₂	2.63, dd (14.7, 8.7)	C-6, C-5, C-2''''
		2.25, dd (15.0, 5.7)	C-6, C-5, C-2''''		2.18, m	C-6, C-5, C-2''''
2′′	118.8, CH	5.03, m	C-1'', C-4'', C-5''	118.8, CH	5.13, m	C-8, C-1'', C-4'', C-5''
2′′′	117.4, CH	4.66, m	C-8, C-1''', C-4''', C-5'''	117.4, CH	4.71, m	C-8, C-1''', C-4''', C-5'''
2''''	115.6, CH	4.99, m	C-6, C-4'''', C-5''''	115.5, CH	4.97, m	C-6, C-4'''', C-5''''
3''	136.0, C			136.0, C		
3′′′	136.9, C			136.9, C		
3''''	136.9, C			136.9, C		
4′′	26.0, CH ₃	1.64, s	C-1'', C-2'', C-5''	18.0, CH ₃	1.61, s	C-1'', C-2'', C-5''
4′′′	18.1, CH ₃	1.52, s	C-1''', C-2''', C-5'''	17.8, CH ₃	1.52, s	C-1''', C-2''', C-5'''
4''''	26.0, CH ₃	1.71, s	C-2'''', C-3'''', C-5''''	25.9, CH ₃	1.71, s	C-2'''', C-3'''', C-5''''
5''	18.1, CH ₃	1.59, s	C-1'', C-2'', C-4'''	26.0, CH ₃	1.68, s	C-1'', C-2'', C-4'''
5′′′	25.8, CH ₃	1.55, s	C-1'', C-2'', C-4'''	25.7, CH ₃	1.56, s	C-1'', C-2'', C-4'''
5''''	17.9, CH ₃	1.55, s	C-2'''', C-3'''', C-4''''	17.9, CH ₃	1.54, s	C-2'''', C-3'''',C-4''''

Figure S23 in the Supporting Information). This distance was computed to 2.3 Å for the (2S*,4S*) diastereomer, and 3.7 Å for the (2S*,4R*) configured C-ring, using Boltzmann averaging of the H-2-H-3 distances of the output conformers of Monte Carlo conformational searches (see Experimental Section for details; this distance was not measured for 2, because of signal overlap). Based on biogenetic considerations discussed for compound 1 and the NMR data, compounds 2 and 3 are C-6 epimers and possess a (2S,4R) absolute configuration. In order to determine their absolute configuration, their electronic circular dichroism (ECD) spectra (Figure 2) were recorded. The observed Cotton effects supported the ORD-based suggestion of the diastereomeric relationship of 2 and 3 as enantiomers are expected to show mirror-image ECD spectra. The comparison of the experimental spectra with those reported for flavan-4-ols¹¹ did not allow configurational assignment, because of the dissimilarity of rings A of 2 and 3, compared to those of the reported compounds. Consequently, the Boltzmann-weighted ECD spectra of the possible diastereomers were calculated using

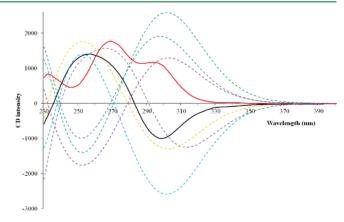


Figure 2. ECD spectra of compounds **2** (black) and **3** (red) along with the calculated spectra of the stereoisomers [(2S,4R,6S) yellow, (2S,4R,6R) blue, (2S,4S,6R) green, (2S,4S,6S) lilac, (2R,4S,6R) pink, (2R,4S,6S) light blue]. See details of the ECD calculations in the Experimental Section.

Table 3. ¹H and ¹³C NMR Spectroscopic Data (800 MHz, CDCl₂) for Rhodiflavan C (4)

		, 3/	()
position	δ_{C} type	$\delta_{ m H_{\scriptscriptstyle J}}$ m (J in Hz)	НМВС
2	78.8, CH	5.45, dd (12.3, 2.2)	C-3, C-4, C-1', C-2'/C-6'
3	37.0, CH ₂	2.31, dt (14.8, 2.3)	C-4, C-4a
		2.02, ddd (14.9, 12.3, 3.5)	C-2, C-1'
4	55.6, CH	4.86, m	C-2, C-5, C-4a, C-7a
4a	125.9, C		
5	184.5, C=O		
6	203.4 C=O		
7	52.9, C,		
7a	188.6, C-O		
1'	137.9, C		
2', 6'	126.3, CH	7.40, m	C-2, C-2', C-4', C-6'
3', 5'	128.9, CH	7.46, m	C-1', C-2', C-3', C-5', C-6'
4′	129.1, CH	7.43, m	C-2', C-6', C-3', C-5'
1''	32.8, CH ₂	2.51, dd (14.3, 7.2)	C-5, C-6, C-7, C-7a, C-2", C-3", C-1""
		2.42, dd (14.3, 7.9)	C-5, C-6, C-7, C-7a, C-2", C-3", C-1""
1'''	32.6, CH ₂	2.56, dd (14.2, 7.2)	C-5, C-6, C-7, C-7a, C-2", C-3", C-1"
		2.44, dd (14.2, 7.9)	C-5, C-6, C-7, C-7a, C-1", C-2", C-3"
2''	116.9, CH	4.86, m	C-8, C-1", C-4", C-5"
2'''	116.9, CH	4.86, m	C-8, C-1''', C-4''', C-5'''
3''	136.6, C		
3'''	136.8, C		
4''	25.9, CH ₃	1.61, s	C-1'', C-2'', C-5''
4'''	25.9, CH ₃	1.61, s	C-1'', C-2'', C-4'''
5''	17.9, CH ₃	1.57, s	C-1''', C-2''', C-5'''
5'''	17.8, CH ₃	1.56, s	C-1'', C-2'', C-4'''

Table 4. ¹H and ¹³C NMR Spectroscopic Data (800 MHz, CDCl₃) of Compound 5

position	δ_{C} , type	$\delta_{ ext{H}}$, m $(J ext{ in Hz})$	HMBC
1	131.8, CH	7.20, s	C-3, C-11a, C-4a, C-1'
2	121.8, C		
3	155.8, C-O		
4	104.0, CH	6.40	C-2, C-3, C-11b, C-4a
4a	154.1, C-O		
6	69.6, CH ₂	3.95, d (11.5)	C-4a, C-6b, C-11a
		4.15, d (11.5)	C-4a, C-6a, C-6b, C-11a
6a	77.2, C–H		
6b	119.0, C		
7	103.0, CH	6.80, s	C-6a, C-8, C-9, C-10a
8	142.4, C-O		
9	149.7, C-O		
10	94.2, CH	6.81, s	C-8, C-9, C-6b, C-10a
10a	149.7, C		
11a	84.8, C-H	5.25, s	C-1, C-4a, C-6, C-6a, C-11b, C-10a
11b	112.1, C		
OCH ₂ O-8, 9	101.5, CH ₂	5.91, d (1.4) 5.94, d (1.4)	C-8, C-9
1'	29.2, CH ₂	3.32, d (7.2)	C-1, C-3, C-2', C-3'
2'	121.8, CH	5.31, m	C-4', C-5', C-1'
3'	135.8, C		
4'	17.9, CH ₃	1.78, s	C-2', C-3', C-5'
5'	25.8, CH ₃	1.78, s	C-2', C-3', C-4'

the TD-DFT method (see Experimental Section for details), and compared to those obtained experimentally for $\mathbf{2}$ and $\mathbf{3}$ (Figure 2). The experimental ECD spectrum of $\mathbf{2}$ showed a negative Cotton effect at ca. 300 nm, positive at 260, and negative at 230 nm, best matched with the calculated spectrum of the (2S,4R,6S) isomer. Nonetheless, the mirror imaged spectrum of the (2S,4R,6S) isomer, corresponding to a (2R,4S,6S) configuration, shows similar Cotton effects. Furthermore, the calculated spectra of the (2S,4R,6S) and

(2*S*,4*S*,6*S*) isomers are remarkably similar, and the same goes for the other two calculated isomers. None of these show agreement with the experimentally obtained ECD spectrum of 3. Based on the above spectroscopic evidence, the new compounds rhodiflavan A (2) and B (3) are characterized as diastereomers of 4,6-dihydroxy-6,8,8-tris(3-methylbut-2-en-1-yl)-2-phenyl-2,3,4,8-tetrahydro-5*H*-chromene-5,7(6*H*)-dione, with $(2S^*,4R^*,6S^*)$ and $(2S^*,4R^*,6R^*)$ relative configurations, respectively.

Compound 4 was isolated as an oily paste. Its molecular formula was deduced as $C_{24}H_{28}O_4$ based on HREIMS (m/z[M]+ 381.2062, calcd 381.2066) and NMR data (see Table 3, as well as Figures S23-S28 in the Supporting Information) analyses. The ¹H NMR signals at $\delta_{\rm H}$ 5.45 (H-2), 2.02 (H-3_{ax}), 2.31 (H-3_{eq}), and 4.86 (H-4), the ¹³C NMR signals at δ_C 78.8 (C-2), 37.0 (C-3), and 55.6 (C-4), along with the UV λ_{max} at 230 and 270 nm suggested compound 4 to have an analogous ring C to those of 2 and 3. Its 1 H NMR signals at $\delta_{\rm H}$ 7.4 (H-2'/6'), 7.46 (H-3'/5'), and 7.43 (H-4') indicated ring B to be unsubstituted. The lack of aromatic protons for ring A suggested it to be fully substituted. As rings B and C were fully assigned, the carbonyl groups at $\delta_{\rm C}$ 184.5 (C-5) and 203.4 (C-6), and the two prenyl groups (see Table 3 for signals 1''5'', and 1'''-5'') could only be related to ring A. The HMBC correlations of H-2'' ($\delta_{\rm H}$ 4.86) and H-2''' ($\delta_{\rm H}$ 4.86) with C-7 $(\delta_{\rm C}$ 52.9) revealed the two prenyl groups to be both attached to C-7. Both methylene groups CH_2-2'' (δ_H 4.86) and CH_2-1 2'''' ($\delta_{\rm H}$ 4.86) showed HMBC correlation to $\delta_{\rm C}$ 188.6 (C-7a), $\delta_{\rm C}$ 203.4 (C-6), and $\delta_{\rm C}$ 52.9 (C-7), indicating the placement of the two carbonyl groups at C-5 and C-6, which, being a fivemembered ring, is unique for the genus Tephrosia. The difference in the chemical shifts of C-5 ($\delta_{\rm C}$ 184.5) and C-6 ($\delta_{\rm C}$ 203.4) is rationalized by the strong intramolecular hydrogen bond of HO-4 and the oxygen of the C-5 carbonyl group, and C-5 has strong conjugation with C-7a through the 4a (7a) double bond. The observed mass and NMR data (vide supra), the UV spectrum only showing a benzenoid band, and the HMBC of H-4 ($\delta_{\rm H}$ 4.86) with C-5 ($\delta_{\rm C}$ 184.5) corroborate the proposed structure. It is optically active ($[\alpha]_D$ +90 (c 0.001, CH₂Cl₂), and its ECD spectrum (see Figure S30 in the Supporting Information) shows a positive Cotton effect at λ 300-340 nm and a negative one at 275 nm. Based on the similarity of the NMR data (δ , J, Table 3) of H-2, H-3, and H-4 (ring C) of 4 to those of compounds 2 and 3 (Table 2), compound 4 was assigned the (2S*,4R*) relative configuration. Therefore, based on the above spectroscopic data, this unique compound, rhodiflavan C (4), was characterized as (2S*,4R*)-4-hydroxy-7,7-bis(3-methylbut-2-en-1-yl)-2-phenyl-2,3,4,7-tetrahydrocyclopenta[b]pyran-5,6-dione.

Compound 5 was isolated as an amorphous solid and was assigned the molecular formula C₂₁H₂₀O₆, based on HRESIMS $([M + H]^{+} m/z \text{ obs } 369.1367) \text{ and NMR data (see Table 4, as})$ well as Figures S31-S36 in the Supporting Information) analyses. It showed characteristic UV (λ_{max} 230, 190, 310, and 350 nm), ${}^{1}\text{H}$ NMR [δ_{H} 3.95 and 4.15 (CH₂-6), 5.25 (H-11a)], and ¹³C NMR data [$\delta_{\rm C}$ 69.6 (C-6), 77.2 (C-6a), 84.8 (C-11a) for a 6a-hydroxypterocarpan skeleton. The NMR data further indicated the presence of a prenyl group [$\delta_{\rm H}$ 3.32 (H₂-1'), 5.31 (H-2'), 1.78 (H₃-4' and H₃-5'), and $\delta_{\rm C}$ 29.2 (C-1') 121.8 (C-2'), 135.8 (C-3'), 17.9 (C-4'), 25.8 (C-5')] connected to C-2, as revealed by HMBC cross-peaks of H-1 $(\delta_{\rm H}$ 7.20) with C-1' $(\delta_{\rm C}$ 29.2), and of CH₂-1' $(\delta_{\rm H}$ 3.32) with C-1 ($\delta_{\rm C}$ 131.8) and C-3 ($\delta_{\rm C}$ 155.8). The position of the prenyl group was confirmed by the NOE correlation between H-1 ($\delta_{
m H}$ 7.20) and CH₂-1' ($\delta_{\rm H}$ 3.32). Hence, ring A is substituted with a prenyl group at C-2 and a hydroxy group at C-3. Its C-1 and C-4 are unsubstituted as evidenced by the two singlets in its ¹H NMR data. In ring D, the NMR data (Table 4) indicated the presence of an 8, 9-methylenedioxy group ($\delta_{\rm H}$ 5.91 and 5.94; $\delta_{\rm C}$ 101.5) and two *p*-substituted aromatic protons at $\delta_{\rm H}$ 6.80 (H-7) and $\delta_{\rm H}$ 6.81 (H-10). The NMR spectra of this compound showed resemblance to those of previously published pterocarpans.^{7,27–29} The absolute configuration at the B/C-ring junction was determined as (6aS,12aS), based on the ECD spectrum (see Figure S34 in the Supporting Information), which exhibited a positive and negative Cotton effect at 309 nm and 241 nm, respectively.³⁰ Based on the above spectroscopic data, this new compound, rhodacarpin (5), was characterized as (6aS,12aS)-2-(3-methylbut-2-en-1-yl)-6H-[1,3]dioxolo[4',5':5,6]benzofuro[3,2-c]chromene-3,6a-(12aH)-diol.

The crude extract of the roots of *T. rhodesica* and the isolated compounds were tested for antiplasmodial activity using a previously established protocol. The crude extract gave 100% growth inhibition of the chloroquine-sensitive (3D7) strain of *Plasmodium falciparum* at 10 μ g/mL concentration, whereas the isolated compounds 2, 3, 4, 5, 9, and 15 showed moderate activities (Table 5). Among the tested compounds, 3 showed the highest activity (IC₅₀ = 5.7 \pm 1.9 μ M). 32

Table 5. In Vitro Antiplasmodial Activities (IC₅₀) of Isolated Compounds against 3D7 Strains of *P. falciparum*^a

sample	IC ₅₀ , μΜ (Pf 3D7)	LD ₅₀ (HEK-293)
T. rhodesica (roots) crude extract	100% active at 10 μ g/mL	
Rhodiflavan A (2)	7.3 ± 1.8	71% at 40 μM
Rhodiflavan B (3)	5.7 ± 1.9	34% at 40 μM
Rhodiflavan C (4)	7.0 ± 2.4	101% at 40 μM
Rhodacarpin (5)	10.2 ± 0.2	27% at 40 μM
Quercetol B (9)	7.4 ± 0.3	87% at 40 μM
6-Hydroxyrotenone (15)	8.6 ± 2.6	$10.9 \mu M$
Tephrowatsin A (21)	14.5 ± 0.7	
Chloroquine	0.0047	
Artesunate	0.00067	

^aCompounds were tested independently either two or three times against Pf3D7. All compounds were tested in a single experiment against HEK 293 mammalian cells (LD₅₀).

In conclusion, five new compounds—rhodimer (1), rhodiflavan A (2), rhodiflavan B (3), rhodiflavan C (4), and rhodacarpin (5)—were isolated from the roots of *Tephrosia rhodesica*, along with 16 known natural products. Rohodimer is an unusual flavanone-flavan dimer, whereas rhodiflavan C has a five-membered A-ring, which is unprecedented for this genus. The crude root extract and several of its isolated constituents showed activities against the chloroquine-sensitive (3D7) strain of *Plasmodium falciparum*.

■ EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were measured on a PerkinElmer 341-LC system, whereas ECD experiments were performed on a Jasco Model J-715 spectropolarimeter. UV spectra were recorded on a Specord S600 (Analytik Jena AG) spectrophotometer. Melting points were obtained on a Büchi Melting Point B-545 Switzerland apparatus; NMR spectra were acquired on a Bruker Avance III HD 800 MHz NMR spectrometer equipped with a TCI cryogenic probe and were processed with the MestReNova 10.0 software, using the solvent residual signal (CDCl₃ $\delta_{\rm H}$ 7.26; $\delta_{\rm C}$ 77.16) as chemical shift reference. LC-ESIMS data were obtained on a Micromass GC-TOF micro mass spectrometer (Micromass, Wythenshawe, Waters, Inc., U.K.), using direct inlet, and 70 eV ionization voltage. TLC was performed on Merck precoated silica gel 60 F254 plates. Column chromatography was run on silica gel 60 (70-230 mesh). Gel filtration was done on Sephadex LH-20. Preparative HPLC was performed on a Waters 600E instrument using the

Chromulan (Pikron, Ltd.) software and an RP C_8 Kromasil (250 mm \times 55 mm) column eluting with repeated CH_3OH-H_2O (5 to 95) gradients. Single-crystal X-ray data of 1 were collected at 120 K on an Agilent SuperNova dual wavelength diffractometer with a microfocus X-ray source and multilayer optics monochromatized Cu K α (λ = 1.54184 Å) radiation.

Plant Material. The roots of *Tephrosia rhodesica* were collected in April 2015 from Kilungu Hills in Makueni County, Kenya. The plant specimen was identified by Mr. Patrick C. Mutiso of the School of Biological Sciences, the University of Nairobi. A voucher specimen (Mutiso-842/April 2015) was deposited at the University Herbarium of the University of Nairobi, Kenya.

Extraction and Isolation. The air-dried roots (2 kg) of T. rhodesica were ground and extracted with CH₂Cl₂-CH₃OH (1:1) (3 × 2 L) by percolation at room temperature to yield 70 g of a dark brown paste after evaporation of the solvent. A portion of the extract (37 g) was subjected to column chromatography over silica gel (400 g) eluting with a mixture of iso-hexane containing increasing amounts of EtOAc. The fraction eluted with 1% EtOAc in iso-hexane was purified by preparative HPLC (CH₃OH-H₂O gradient elution) to give tephrowatsin B (6, 20 mg).8 The fraction eluted with 3% EtOAc in iso-hexane was purified by column chromatography on a Sephadex LH-20 column (CH₂Cl₂-CH₃OH; 1:1) to give tephrinone (7, 200 mg)9 and glabranin (8, 100 mg).10 The fractions eluted with 5% EtOAc in iso-hexane were purified by column chromatography on a Sephadex LH-20 column (CH₂Cl₂-CH₃OH; 1:1), followed by recrystallization from CH₂Cl₂-CH₃OH (1:1) to afford rhodimer (1, 15 mg). The fractions eluted with 6% EtOAc in iso-hexane were subjected to column chromatography on a Sephadex LH-20 (CH₂Cl₂-CH₃OH (1:1)) system to give quercetol B (9, 300 mg). The fractions eluted with 7% EtOAc in iso-hexane were purified by column chromatography on a Sephadex LH-20 column (CH₂Cl₂–CH₃OH (1:1)) to give maackiain (**10**, 20 mg),¹² 6a-hydroxymaackiain (**11**, 10 mg),¹³ and pisatin (**12**, 5 mg).¹⁴ The fractions eluted with 8% EtOAc in iso-hexane were subjected to column chromatography on a silica gel (200 g) with 3% EtOAc in isohexane to give tephrowatsin A (21, 10 mg), with 5% EtOAc in isohexane to give tephrowatsin hexane to give rhodiflavan A (2, 50 mg), and with 10% EtOAc in isohexane to provide rhodiflavan B (3, 20 mg) and rhodiflavan C (4, 15 mg), which were further purified by preparative HPLC (CH₃OH-H₂O gradient elution). The fractions eluted with 9%-10% EtOAc in iso-hexane were combined and purified by column chromatography on a Sephadex LH-20 column (CH2Cl2-CH3OH; 1:1), and subsequently by preparative HPLC (CH₃OH-H₂O gradient elution) to give tephrosin (13, 10 mg),¹⁵ rotenone (14, 15 mg),¹⁵ 6hydroxyrotenone (15, 10 mg), ¹⁶ 12a-hydroxyrotenone (15, 10 mg), ¹⁷ rhodacarpin (5, 10 mg), hildecarpin (17, 10 mg), 18 and 3-hydroxy-2methoxy-8-9-methylenedioxypterocarpene (18, 15 mg). The fractions eluted with 12% EtOAc in iso-hexane gave isoliquirtigenin (19, 15 mg),²⁰ and those eluted with 20% EtOAc in iso-hexane gave D-pinitol (20, 900 mg).²

Rhodimer (1). White crystals (CH₂Cl₂-CH₃OH; 1:1); $[\alpha]_D^{10}$ +13.4 (c 0.001, CH₃OH); UV (CH₂Cl₂) λ_{max} (log ε) 230 (4.05), 290 (4.25), and 350 (4.30) nm; ECD (c 0.05, CH₃OH) λ_{max} ($\Delta \varepsilon$) 314 (10.16), 292 (-40.36), 249 (-23.33), 235 (50.08), 220 (-56.36), 212 (62.23); H and 13 C NMR, see Table 1; HREIMS m/z 660.3095 [M]⁺ (calcd for C₄₂H₄₄O₇, 660.3087).

Rhodiflavan A (2). Yellow oily paste; $[\alpha]_D^{20}$ +17.5 (*c* 0.001, CH₂Cl₂); UV (CH₂Cl₂) λ_{max} (log ε) 240 (3.05) and 290 (3.40) nm; ECD (*c* 0.05, CH₃OH) λ_{max} (Δε) 300 (-92.25), 257 (85.45), 243 (-119.45); ¹H and ¹³C NMR, see Table 2; HRESIMS m/z 479.2493 [M + H]⁺ (calcd for C₃₀H₃₈O₅, 479.2753).

Rhodiflavan B (3). Yellow oily paste; $[\alpha]_D^{20}$ +66.1 (c 0.001, CH₂Cl₂); UV (CH₂Cl₂) λ_{max} (log ε) 230 (3.08) and 280 (3.20) nm; ECD (c 0.05, CH₃OH) λ_{max} ($\Delta \varepsilon$) 301 (199.73), 257 (-130.29), 225 (65.0); ¹H and ¹³C NMR, see Table 2; HRESIMS m/z 479.2493 [M + H]⁺ (calcd for C₃₀H₃₈O₅, 479.2753).

Rhodiflavan C (4). Yellow oily paste. $[\alpha]_D^{20}$ +90.0 (*c* 0.001, CH₂Cl₂); UV (CH₂Cl₂) λ_{max} (log ε) 230 (3.20) and 270 (3.08) nm; ECD (*c* 0.05, CH₃OH) λ_{max} (Δε) 316 (89.0), 284 (-32.21), 245

(91.80), 226 (-26.26), 218 (70.0), 210 (80.5); 1 H and 13 C NMR, see Table 3; HREIMS m/z 381.2062 [M] $^{+}$ (calcd for $C_{24}H_{28}O_{4}$, 381.2066).

Rhodacarpin (*5*). White amorphous solid; UV (CH₂Cl₂) $\lambda_{\rm max}$ (log ε) 230 (3.04), 290 (3.45), 310 (3.42) and 350 (3.10) nm; ECD (c 0.05, CH₃OH) $\lambda_{\rm max}$ (Δ ε) 309 (12.0), 241 (-82.1), 220 (7.0), 205 (-80); 1 H and 13 C NMR, see Table 4; HRESIMS m/z 369.1367 [M + H]⁺ (calcd for C₂₁H₂₀O₆, 369.1338).

Plasmodium falciparum Culture. In vitro parasite culture of the *P. falciparum* strain 3D7 was maintained in RPMI with 10 mM Hepes (Life Technologies), 50 μ g/mL hypoxanthine (Sigma) and 5% human serum from male AB plasma and 2.5 mg/mL AlbuMAX II (Life Technologies). Human 0⁺ erythrocytes were obtained from the Australian Red Cross Blood Service (Agreement No. 13-04QLD-09; 17-06QLD-16). The parasites were maintained at 2%–8% parasitaemia (% P) at 5% hematocrit (% H), and incubated at 37 °C, 5% CO₂₁ 5% O₂₁ 90% N₂ and 95% humidity.

Plasmodium falciparum Growth Inhibition Assay. A well-established asexual *P. falciparum* imaging assay was used to determine parasite growth inhibition according to the procedure described by Duffy and Avery.³¹

Computation. A low-energy conformation library of postulated compounds 2 and 3 were generated using Macromodel as implemented in the Macromodel v12.1 Schrödinger suite by performing careful Monte Carlo conformational analysis using MMFF force fields, each with the GB/SA solvation models CHCl₃ and H₂O. Elimination of redundant conformations was performed by comparison of heavy atom coordinates applying an RMSD cutoff set to 2.0 Å. Next, all unique conformers were optimized at the B3LYP/6-311++G(3df,2pd) level of theory. The conformers having a Boltzmann weight, calculated using ΔH° , above 1% were selected for subsequent TD-DFT calculations at the same level of theory using 50 singlet excited states. To obtain the Boltzmann-weighted ECD spectrum, the individual spectra were line-broadened using a Gaussian band shape (σ = 0.3 eV). All (TD-)DFT calculations were performed using Gaussian 16, Revision A.03.33 The solvent was taken into account by using the integral equation formalism model (IEFPCM) as implemented in Gaussian 16 and the dielectric constant for methanol (ε = 32.613). To compensate for the typical underestimation of the transition energies, a 20 nm blue shift was applied on the computed ECD spectra for comparison with the experimental one.³

ASSOCIATED CONTENT

Supporting Information

CCDC 1987379 has been deposited with the Cambridge Crystallographic Data Centre. Copies of the data can be obtained, free of charge, on application to the Director, CCDC, 12 Union Road, Cambridge CB2 1EZ, U.K. (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk). The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jnatprod.0c00245.

Original MS and NMR spectra for all compounds, along with the corresponding NMReDATA³⁵ for the new compounds 1–5; the X-ray structure of 1 (details are freely available on Zenodo at https://10.5281/zenodo. 3679345); NMR and MS data for compounds (1–21) (PDF)

Crystallographic data for 1 (CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Chen, Y.; Yan, T.; Gao, C.; Cao, W.; Huang, R. Molecules 2014, 19, 1432–1458.
- (2) Muiva, L. M.; Yenesew, A.; Derese, S.; Heydenreich, M.; Peter, M. G.; Akala, H. M.; Eyase, F.; Waters, N. C.; Mutai, C.; Keriko, J. M.; Walsh, D. *Phytochem. Lett.* **2009**, *2*, 99–102.
- (3) Juma, W. P.; Akala, H. M.; Eyase, F. L.; Muiva, L. M.; Heydenreich, M.; Okalebo, F. A.; Gitu, P. M.; Peter, M. G.; Walsh, D. S.; Imbuga, M.; Yenesew, A. *Phytochem. Lett.* **2011**, *4*, 176–178.

- (4) Touqeer, S.; Saeed, M. A.; Ajaib, M. Phytopharmacology 2013, 4, 598-637.
- (5) Muiva-Mutisya, L.; Macharia, B.; Heydenreich, M.; Koch, A.; Akala, H. M.; Derese, S.; Omosa, L. K.; Yusuf, A. O.; Kamau, E.; Yenesew, A. *Phytochem. Lett.* **2014**, *10*, 179–183.
- (6) Atilaw, Y.; Muiva-Mutisya, L.; Ndakala, A.; Akala, H. M.; Yeda, R.; Wu, Y. J.; Coghi, P.; Wong, V. K. W.; Erdélyi, M.; Yenesew, A. *Molecules* **2017**, 22, 1514.
- (7) Atilaw, Y.; Duffy, S.; Heydenreich, M.; Muiva-Mutisya, L.; Avery, V. M.; Erdélyi, M.; Yenesew, A. *Molecules* **2017**, 22, 318.
- (8) Gomez, F.; Quijano, L.; Calderon, J. S.; Rodriquez, C.; Rios, T. *Phytochemistry* **1985**, *24*, 1057–1059.
- (9) Dagne, E.; Mammo, W.; Sterner, O. *Phytochemistry* **1992**, *31*, 3662–3663.
- (10) Céspedes, C. L.; Achnine, L.; Lotina-Hennsen, B.; Salazar, J. R.; Gómez-Garibay, F.; Calderón, J. S. *Pestic. Biochem. Physiol.* **2001**, *69*, 63–76.
- (11) Gomez-Garibay, F.; Quijano, L.; Calderon, J. S.; Morales, S.; Rios, T. *Phytochemistry* **1988**, *27*, 2971–2973.
- (12) Mizuno, M.; Tanaka, T.; Katsuragawa, M.; Saito, H.; Iinuma, M. J. Nat. Prod. **1990**, 53, 498–499.
- (13) Bilton, J. N.; Debnam, J. R.; Smith, I. M. *Phytochemistry* **1976**, 15, 1411–1412.
- (14) Kobayashi, A.; Akiyama, K.; Kawazu, K. *Phytochemistry* **1992**, 32, 77–78.
- (15) Stevenson, P. C.; Kite, G. C.; Lewis, G. P.; Forest, F.; Nyirenda, S. P.; Belmain, S. R.; Sileshi, G. W.; Veitch, N. C. *Phytochemistry* **2012**, *78*, 135–146.
- (16) Krupadanam, G. L. D.; Srimannarayana, G.; Rao, N. V. S. *Indian J. Chem., Sect. B* **1978**, *16B*, 770–772.
- (17) Oberholzer, M. E.; Rall, G. J. H.; Roux, D. G. Phytochemistry 1976, 15, 1283-1284.
- (18) Lwande, W.; Bentley, M. D.; Macfoy, C.; Lugemwa, F. N.; Hassanali, A.; Nyandat, E. *Phytochemistry* **1987**, 26, 2425–2426.
- (19) Lwande, W.; Bentley, M. D.; Hassanali, A. Int. J. Trop. Insect Sci. 1986, 7, 501–503.
- (20) Takahashi, T.; Takasuka, N.; Iigo, M.; Baba, M.; Nishino, H.; Tsuda, H.; Okuyama, T. *Cancer Sci.* **2004**, *95*, 448–453.
- (21) Raya-Gonzalez, D.; Pamatz-Bolanos, T.; del Rio-Torres, Rosa E. d.; Martinez-Munoz, R. E.; Ron-Echeverria, O.; Martinez-Pacheco, M. M. Z. Naturforsch., C: J. Biosci. 2008, 63, 922–924.
- (22) Yan, X.-T.; Li, W.; Sun, Y.-N.; Yang, S.-Y.; Lee, S.-H.; Chen, J.-B.; Jang, H.-D.; Kim, Y.-H. *Bioorg. Med. Chem. Lett.* **2014**, 24, 1397–1402
- (23) Asada, Y.; Sukemori, A.; Watanabe, T.; Malla, K. J.; Yoshikawa, T.; Li, W.; Kuang, X.; Koike, K.; Chen, C.-H.; Akiyama, T.; Qian, K.; Nakagawa-Goto, K.; Morris-Natschke, S. L.; Lu, Y.; Lee, K.-H. *J. Nat. Prod.* **2013**, *76*, 852–857.
- (24) Bogorad, L. Annu. Rev. Plant Physiol. 1958, 9, 417-448.
- (25) Slade, D.; Ferreira, D.; Marais, J. P. J. Phytochemistry 2005, 66, 2177–2215.
- (26) Pouget, C.; Fagnere, C.; Basly, J.-P.; Leveque, H.; Chulia, A.-J. *Tetrahedron* **2000**, *56*, 6047–6052.
- (27) Mai, H. D.; Nguyen, T. T.; Pham, V. C.; Litaudon, M.; Gueritte, F.; Tran, D. T.; Nguyen, V. H. *Planta Med.* **2010**, *76*, 1739–1742.
- (28) Soby, S.; Bates, R.; van Etten, H. Phytochemistry 1997, 45, 925–929.
- (29) Soby, S.; Caldera, S.; Bates, R.; VanEtten, H. *Phytochemistry* **1996**, 41, 759–765.
- (30) Goel, A.; Kumar, A.; Raghuvanshi, A. Chem. Rev. 2013, 113, 1614-1640.
- (31) Duffy, S.; Avery, V. M. Am. J. Trop. Med. Hyg. 2012, 86, 84-92.
- (32) Mishra, L. C.; Bhattacharya, A.; Bhasin, V. K. Acta Trop. 2009, 109, 194–198.
- (33) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko, B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J.

V.; Izmaylov, A. F.; Sonnenberg, J. L.; Williams Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.; Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Throssell, K.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. Gaussian 16, Revision C.01; Gaussian: Wallingford, CT, 2016.

- (34) Autschbach, J. Ab Initio Electronic Circular Dichroism and Optical Rotatory Dispersion: from Organic Molecules to Transition Metal Complexes. In *Comprehensive Chiroptical Spectroscopy: Instrumentation, Methodologies, and Theoretical Simulations*, Vol. 1; John Wiley & Sons, 2012; Chapter 21, pp 593–642.
- (35) Pupier, M.; Nuzillard, J. M.; Wist, J.; Schlorer, N. E.; Kuhn, S.; Erdelyi, M.; Steinbeck, C.; Williams, A. J.; Butts, C.; Claridge, T. D. W.; Mikhova, B.; Robien, W.; Dashti, H.; Eghbalnia, H. R.; Fares, C.; Adam, C.; Kessler, P.; Moriaud, F.; Elyashberg, M.; Argyropoulos, D.; Perez, M.; Giraudeau, P.; Gil, R. R.; Trevorrow, P.; Jeannerat, D. Magn. Reson. Chem. 2018, 56, 703–715.

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