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Azulene as a biphenyl mimetic in orexin/hypocretin receptor agonists

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ABSTRACT

Azulene is a rare ring structure in drugs, and we investigated whether it could be used as a biphenyl mimetic in known orexin receptor agonist Nag 26, which is binding to both orexin receptors OX_1 and OX_2 with preference towards OX_2 . The most potent azulene-based compound was identified as an OX_1 orexin receptor agonist (pEC₅₀ = 5.79 \pm 0.07, maximum response = 81 \pm 8% (s.e.m. of five independent experiments) of the maximum response to orexin-A in Ca²⁺ elevation assay). However, the azulene ring and the biphenyl scaffold are not identical in their spatial shape and electron distribution, and their derivatives may adopt different binding modes in the binding site.

1. Introduction

The orexin/hypocretin system, which consists of two orexin receptors (OX1 and OX2) and their endogenous peptide-ligands orexin-A and orexin-B,¹⁻² plays a part in many physiological functions, such as sleep-wake regulation, analgesia, stress response, motivational behaviour and addiction, and feeding and metabolic regulation.^{3–7} Its central role in sleep-wake regulation has raised interest in orexin receptor antagonists as a novel class of hypnotics, now already realized as two compounds in the clinical use.^{8–9} On the other hand, a study in a mouse model of narcolepsy has demonstrated that orexin receptor agonists may prove advantageous as a treatment for narcolepsy.¹⁰ Furthermore, OX₁ activation is linked to anti-proliferation in some colorectal cancer cell lines suggesting that OX₁ agonists could be beneficial for cancer treatment.^{11–15} Unlike antagonists, agonists have gained attention in drug discovery only recently, and a few series of potent small molecular orexin receptor agonists have been reported in the literature so far.^{16–21} The first series was published in 2015 with Nag 26 (1) as the most potent compound in it (Figure 1).^{16,22}

Nag 26, like several other potent orexin receptor agonists, has a biphenyl scaffold.^{16,19–21} As a consequence of replacing the amidesubstituted phenyl moiety of the biphenyl in Nag 26 with heteroaromatic five- or six-membered ring, or alkyl group, the activity towards both receptor subtypes has decreased or completely abrogated.¹⁹ The first reported orexin receptor small molecule agonists, including Nag 26, have been OX_2 -selective, ^{16–18} and not until 2021 a potent dual orexin receptor agonist, RTOXA-43 (2), a derivative of Nag 26, was discovered.¹⁹ **RTOXA-43** was obtained by modifying the dimethylcarbamoyl group of Nag 26 (Figure 1). Restriction of the bond rotation in Nag 26 using naphthalene in place of N-phenylethylenediamine moiety has produced a potent and OX₂-selective agonist 3.²⁰ In addition, replacement of the naphthalene moiety in 3 with tetralin has led to the discovery of compound 4, which is not only a potent OX₂ but also potent OX₁ agonist.²¹ The first potent OX₁-selective agonist **5**, based on compound 4 with modifications on the biphenyl part of the molecule, has been reported recently.²²

Azulene is a bicyclic aromatic hydrocarbon, which has not been intensively employed ring structure in drug molecules. We have

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previously studied the applicability of azulene structure in medicinal chemistry, and shown that azulene scaffold does not have any general liabilities to hinder its use in drug discovery.²⁴ Inspired by this fact, we have studied azulene-based compounds as orexin receptor ligands.^{25–26} These compounds indicate a benzoyl group in the 1-position and an ester group in the 6-position of the azulene scaffold as the most important functionalities for the observed weak orexin receptor agonism, and molecular modelling suggested that the hydrogen bond acceptors of these groups lie approximately at the same distances from each other as the amide and *N*-sulfonyl anilide groups in the biphenyl scaffold of **Nag 26** to further study the applicability of azulenes as biphenyl minetics in orexin receptor ligands. The effect of replacing the biphenyl moiety of **Nag 26** with a fused bicyclic aromatic structure on orexin receptor activity is reported for the first time.

2. Results and discussion

We synthesized azulene derivatives from two building blocks, A and B, which are suitable for sulfonamide coupling in the end of the synthesis (Figure 1). Building block A is a 1-sulfonate azulene with the 3-and 6-positions available for functionalization and building block B is the unmodified half of Nag 26.

6-Methylazulene $(7)^{27}$ or the 6-carboxyazulene²⁵-derived amide **8** were used as starting compounds for synthesis of building block A and the synthesis was started by introducing a trifluoroacetyl or a dimethylamide functionality into position 1 (Scheme 1b). The trifluoroacetyl group was introduced with trifluoroacetic anhydride (TFAA) in the presence of Et₃N giving **9** and **11** in 83% and 94% yields, respectively. The dimethylamide derivative **12** was obtained in 60% yield upon heating **8** and phosgene in toluene followed by addition of dimethylamine.²⁶ To obtain **10**, the temperature was lowered and reaction time shortened, as heating the reaction mixture led to an insoluble solid. The optimized reaction resulted in **10** in 76% yield.

Introduction of a sulfonic acid group to the non-functionalized five-

membered ring of azulene failed to give the monosubstituted product, as only the 1,3-disulfonic acid derivative and unreacted starting material were obtained. However, the sulfonic acid derivatives with dimethylamide or trifluoroacetyl group in the 3-position were obtained as sodium sulfonates in good yields. Altogether four different A building blocks, **13**, **14**, **15** and **16**, were synthesized.

The synthesis of the building block B (compound **20**) was started by heating 1-fluoro-3-nitrobenzene (**17**) in ethylenediamine, which was followed by the addition of toluoyl chloride to give **18** (Scheme 1c). In order to avoid selectivity issues in the coupling of the building blocks A and B, the amino group of **18** was protected with trifluoroacetamide using TFAA. The reduction of the nitro group of **19** gave **20** in 27% yield over three steps.

Building blocks A and B were combined by first converting the sulfonate groups of **13–16** to sulfonyl chlorides by heating them in SOCl₂ (Scheme 1a). Then, the reaction between the sulfonyl chlorides and compound **20** gave **21–24** in 31%, 44%, 38% and 18% yields, respectively. The compounds were deprotected with K_2CO_3 in water and methanol to give the final products **25**, **27**, **28** and **29** in 39%, 53%, 76% and 40% yield, respectively.

As the building block B is the unmodified half of **Nag 26**, it is a useful intermediate also in the synthesis of **Nag 26**, which was synthesized from **20** in two steps with 30% yield. First, **20** was reacted with 5-bromo-2-methoxybenzenesulfonylchloride to give the sulfonamide **30**. Dimethylaminocarbonyl benzene moiety was coupled to **30** under Suzuki reaction conditions with simultaneous deprotection of the amine to obtain **Nag 26**. This route reduces the required synthetic steps from eleven to six compared to the original route for **Nag 26**. Pharmacological characterization of the resulting **Nag 26** has been reported previously.²²

 OX_1 and OX_2 receptor activation was evaluated first for four azulenebased analogs of **Nag 26** and eight synthesis intermediates by measuring compound-mediated Ca²⁺ elevation at 10 μM compound concentration in three independent experiments (Supporting Information Figure S1). Two compounds induced OX_1 -mediated Ca²⁺ elevation, **27** with 80 \pm



Figure 1. a. Small molecule orexin receptor agonists. b. Azulene derivatives (in blue) were designed to replace the biphenyl structure of Nag 26 (in red). The synthesis of the azulene-based compounds was designed from building blocks A and B.



Scheme 1. Synthesis of azulene-based target agonists via building blocks A and B, and a novel synthetic route for Nag 26. Reagents and conditions: (a) TFAA, Et₃N, DCM, 0 °C, 1 h, 83% (9), 94% (11); (b) i) Phosgene, toluene, rt or 80 °C, 1.5–5.5 h, ii) Dimethylamine, THF, 0 °C, 30 min, 76% (10), 60% (12); (c) i) Sulfur trioxide pyridine complex, benzene, 85 °C, 3.5 h, ii) NaOH, H₂O, EtOH, 30 °C, 2 h, 84% (13), 88% (14), 53% (15), 77% (16); (d) i) Ethylenediamine, 100 °C, 23 h, ii) *m*. Toluoyl chloride, *N*,*N*-diisopropylethylamine (DIPEA), DCM, 0 °C, 2 h, 54%; (e) TFAA, Et₃N, THF, 0 °C, 1 h, 99%; (f) H₂, Pd/C, EtOH, rt, 4.5 h, 51%; (g) i) Building block A, SOCl₂, 67 °C, 15 min, ii) 20, Et₃N, DCM, rt, 19–22 h, 31% (21), 44% (22), 38% (23), 18% (24); (h) K₂CO₃, MeOH, H₂O, rt, 3 h, 39% (25), 53% (27), 76% (28), 40% (29); (i) 5-Bromo-2-methoxybenzenesulfonyl chloride, Et₃N, DCM, rt, 48 h, 57%; (j) 3-(*N*,*N*-Dimethylaminocarbonyl)benzeneboronic acid, Pd(dppf)Cl₂, Na₂CO₃, H₂O, 1,4-dioxane, 100 °C, 48 h, 53%.

7% and **29** with 11 ± 4% of the maximum response to orexin-A (Supporting Information Figure S1), while none of the compounds induced Ca²⁺ response via OX₂ receptors. To further confirm that the observed responses were indeed OX₁-mediated, we repeated the measurements using control CHO cells that do not express OX₁ receptors; no response to **27** and **29** were observed on these cells (Supporting Information Figure S2). Subsequently, we studied the ability of these 12 compounds to block the orexin-A-induced Ca²⁺ signal (Supporting Information Figure S3). Compound **27** was showing OX₁ and OX₂ antagonism corresponding to ca. 4 μ M and 2 μ M K_i values, respectively. However, as **27** also showed efficacy on OX₁, the K_i estimate on OX₁ might be misleading. On OX₂, also **24**, **25** and **29** blocked the response to 0.3 nM orexin-A at levels corresponding to approximately 5–10 μ M K_i . The rest of the compounds had only a minor effect on the orexin-A response indicating $K_i > 10 \ \mu$ M if any.

We continued with **27** assessing the concentration-dependence of the Ca²⁺ elevation on OX₁ receptors; stimulation by **27** led to a similar magnitude of Ca²⁺ elevation as by orexin-A, but with a notably weaker potency (Figure 2). Furthermore, the effect of **27** is approximately 10% of the effect of **Nag 26** on OX₁-receptors; pEC₅₀ for **27** was 5.79 \pm 0.07

(n = 5) and for Nag 26 6.66 \pm 0.06 (as measured previously with identical assay set-up.²² However, OX₂ activation differs dramatically between these two compounds – Nag 26 is also a potent OX₂ agonist (pEC₅₀ = 7.77 \pm 0.07²²), whereas 27 has no OX₂ activity (Supporting Information Figure S1).

To explain further the observed preference in efficacy of **27** towards OX_1 receptor compared to OX_2 , and corresponding structure–activity relationships, we took an advantage of a recently published cryo-EM structure of OX_2 in complex with a G protein and a small molecule agonist **31** (PDB ID: 7L1V), structurally closely related to **Nag 26** (Figure 3).²⁸ OX_2 shares 80% sequence identity with OX_1 in the TM region and has nearly identical ligand binding site (only 2–4 differing amino acid residues, depending on how the binding site is framed). We built a homology model of OX_1 based on the cryo-EM structure of OX_2 and docked **27** to the OX_1 model; the conformation and the interactions formed by the highest scoring binding mode of **27** were similar to those of **31** in the OX_2 structure (Figure 3 a and b).

In the **27** docking pose, the sulfonamide core of **27** forms two hydrogen bonds with the upward-facing $\text{Gln}^{3.32}$ (for the amino acid numbering, see references^{29,30}), while the surrounding ring structures



Figure 2. Ca^{2+} responses to orexin-A (filled triangles), **Nag 26** (filled squares) and **27** (filled circles) on CHO-hOX₁ cells. The responses were normalized to baseline separately for each independent sample before averaging. N = 5. **Nag 26** graph is reproduced from data originally published by Rinne *et al.*²² for easier comparison.

are lined by Pro^{3.29}, Met^{4.64}, Ser^{2.61} and Val^{2.64}. The ability to stabilize Gln^{3.32} in the upward conformation was shown to be crucial for agonist activity on OX₂ receptors.²⁸ Notably, only compounds with a sulfonamide group have been reported as potent small molecule orexin receptor agonists so far.^{16–21} The hydrophobic bottom of the binding cavity is occupied by the amide-linked tolyl, while the amide carbonyl lays within a hydrogen-bonding distance from His^{7.39}. Compounds with the trifuoroacetyl protecting group of the secondary amine (as in **22** and **24**) were inactive, suggesting that limited space is available in the bottom of the binding site.

The azulene moiety of **27** is located near TM2 and TM7, surrounded by residues $\text{His}^{7.39}$, $\text{Tyr}^{7.32}$, $\text{Val}^{2.64}$, and $\text{Ser}^{2.61}$. The methyl group in the 6-position of the azulene occupies a nook between $\text{Ser}^{2.61}$ and $\text{His}^{7.39}$. $\text{Ser}^{2.61}$ is one of the few positions where the ligand binding site of OX₁ differs from that of OX₂; OX₂ has a threonine in this position, thus leaving less room between TM2 and TM7. This likely explains the difference between the orexin receptor subtypes in efficacy of **27**, as the methyl group would not fit between TM2 and TM7 in OX_2 as it does in OX_1 (Figure 3c). Furthermore, when replacing the methyl group in the 6position of azulene with a dimethylamide group (compound **29**), the OX_1 agonist activity decreases dramatically, as there is likely no room for a larger functional group between the helices. Even more drastic drop in the activity is observed when the dimethylamide group in the 3position of azulene was replaced with a trifluoroacetyl group, as both trifluoroacetyl derivatives **25** and **28** were inactive.

Previously reported modifications at a certain position of Nag 26 structure have resulted in either OX2 selective agonists, or agonists with equal effect on both orexin receptors.^{19–21} Identification of OX₁ selectivity has required the modification of the structure of Nag 26 on both ends of the molecule - at first, a tetralin-type dual-orecin receptor agonist 4 was developed,²¹ and further replacement of the biphenyl structure of 4 with anisole or cinnamide derivative has led to compounds preferentially activating OX₁ at moderate potency.²³ Optimization of the compounds has resulted in the first potent OX_1 selective agonist 5. Interestingly, replacing the biphenyl moiety of Nag 26 itself with cinnamide derivative has led to complete loss of both OX1 and OX2 activity, ¹⁹ suggesting that the preference towards OX_1 in compound 5 and its analogs is not dependent solely on the modification of the biphenyl structure. Thus, the azulene-based derivative 27 reported herein is the first analog of Nag 26 with preference towards OX1, which has been reached by modifying only the biphenyl structure. As such, compound 27 contributes to the establishment of structure-activity relationships related to subtype selectivity between the orexin receptors and development of OX₁-selective agonists, although the potency of the compound has yet remained moderate.

3. Conclusions

In conclusion, our results demonstrate that the azulene ring can mimic a biphenyl moiety in orexin receptor agonists. Both types of orexin receptor agonists, those bearing biphenyl as well as those bearing



Figure 3. a. Binding mode of agonist **31** (cyan) in complex with OX_2 (PDB ID: 7L1V). **b.** Docking pose of **27** (bright yellow) in complex with OX_1 homology model. **c.** Superimposition of **27**- OX_1 complex to OX_2 structure. Red dashes mark the possible clash between **27** and OX_2 . Receptors are shown as cartoon and the binding site residues as sticks (white and pink for OX_1 and OX_2 , respectively). Color code is as follows: blue, nitrogen; red, oxygen; yellow, sulfur; pink (OX_2) / white (OX_1) / cyan (**31**) / bright yellow (**27**), carbon. Gray dashes indicate possible hydrogen bonds. **d.** 2D comparison of **Nag 26** (**1**), **27** and **31**. Analogous parts of the compounds are marked with same colors.

azulene scaffold, have been shown to exhibit good efficacy on orexin receptors. While the efficacy of the azulene-based OX₁ agonist **27** is close to that of orexin-A, the potency of **27** remains moderate. However, this study was based on a limited number of compounds, and thus further compound optimization could improve the potency. As the majority of the currently known small molecule orexin receptor agonists are OX₂-selective, **27** poses a promising starting point for developing OX₁-selective tool compounds for pharmacological research to validate the significance of OX₁ modulation *in vivo*. Understanding the structure–activity relationships for OX₁-selective agonists could also open other therapeutic opportunities in conditions such as colon cancer.

4. Experimental section

4.1. Chemistry

General information: All compounds were characterized by NMR spectroscopy using a Bruker Avance III 400 MHz spectrometer or a Varian Mercury 300 MHz spectrometer. Chemical shifts are reported in parts per million (ppm) relative to the residual solvent signals: $CDCl_3$ 7.26 and 77.16 ppm, DMSO- d_6 2.50 and 39.52 ppm, acetone- d_6 2.05 and 29.84 ppm, CD_3OD 3.31 and 49.00 ppm for ¹H and ¹³C NMR, respectively. The progress of the reactions was monitored by thin-layer chromatography on silica gel 60-F₂₅₄ plates. When a product was purified by flash chromatography, silica gel (SiO₂) 60 (230–400 mesh) or silica gel with a Biotage SP1 purification system (SNAP 10, 25, or 50 g cartridges) was used. Mass spectrometer using electrospray ionization (ESI). The purity was determined by UPLC-MS with diode-array detector. The purity of all biologically tested compounds was 95% or higher.

N,N-Dimethylazulene-6-carboxamide (8). Azulene-6-carboxylic acid (0.25 g, 1.4 mmol) was dissolved in DMF (5.5 mL). The blue solution was cooled to 0 $^\circ$ C and EDC·HCl (0.30 g, 1.6 mmol) and HOBt·H₂O (0.24 g, 1.6 mmol) were added. The reaction mixture was stirred at 0 $^{\circ}$ C for 15 min and at room temperature for 1 h. A 2 M solution of dimethylamine in THF (1.8 mL, 3.6 mmol) was added and the stirring was continued at room temperature for 1 h. H₂O (40 mL) was added to the reaction mixture, and it was extracted with EtOAc (80 mL). The organic phase was washed with a 1 M aqueous solution of HCl (40 mL) and a saturated aqueous solution of NaHCO₃ (40 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a green solid. The crude product was purified by flash chromatography (EtOAc) to give 8 as a green, amorphous solid (0.23 g, 81%). ¹H NMR (400 MHz, CDCl₃) δ 8.39-8.36 (m, 2H), 7.97-7.96 (m, 1H), 7.45-7.44 (m, 2H), 7.15-7.12 (m, 2H), 3.16 (s, 3H), 2.94 (s, 3H). 13 C NMR (101 MHz, CDCl₃) δ 172.9, 144.9, 140.1, 138.5, 136.3, 120.8, 119.4, 39.5, 35.4. HRMS (ESI-QTOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₄NO 200.1075; Found 200.1077.

2,2,2-Trifluoro-1-(6-methylazulen-1-yl)ethan-1-one (9). TFAA (0.46 mL, 3.3 mmol) was added dropwise to a solution of 6-methylazulene (0.43 g, 3.0 mmol) and Et₃N (0.84 mL, 6.0 mmol) in anhydrous DCM (8.0 mL) under argon on ice bath. The resulting mixture was stirred at 0 °C for 1 h. Then H₂O (60 mL) was added and stirring was continued for an additional 5 min. DCM (60 mL) was added and phases were separated. The organic phase was washed with H₂O (3 \times 40 mL) and brine (40 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a red solid. The crude product was purified by flash chromatography (n-heptane/EtOAc 9:1) to give 9 as a red, amorphous solid (0.60 g, 83%). ¹H NMR (300 MHz, CDCl₃) δ 9.72 (d, J = 10.2 Hz, 1H), 8.39 (d, *J* = 10.2 Hz, 1H), 8.26–8.23 (m, 1H), 7.67 (d, *J* = 10.5 Hz, 1H), 7.57 (d, J = 10.2 Hz, 1H), 7.23 (d, J = 4.2 Hz, 1H), 2.78 (s, 3H). ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3) \delta 176.0 \text{ (q}, J = 33.5 \text{ Hz}), 153.6, 146.0, 142.7, 139.8 \text{ (q}, J = 33.5 \text{ Hz}), 153.6, 146.0, 1$ J = 3.8 Hz), 138.9, 138.2, 133.0, 131.9, 119.6, 117.6, 117.6 (q, J = 290.2 Hz), 28.3. HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for C13H10OF3 239.0684; Found 239.0690.

N,N,6-Trimethylazulene-1-carboxamide (10). A 20% solution of

phosgene in toluene (9.3 mL, 18 mmol) was added to a solution of 6methylazulene (0.50 g, 3.5 mmol) in anhydrous toluene (0.90 mL) under argon. The resulting dark blue solution was stirred at room temperature for 1.5 h and then solvents were evaporated. The purple residue was dissolved in anhydous THF (6.0 mL) under argon. The purple solution was cooled to 0 °C and a 2 M solution of dimethylamine in THF (8.8 mL) was slowly added. The reaction mixture was stirred at 0 °C for 30 min. The mixture was diluted with DCM (60 mL) and washed with a 1 M aqueous solution of HCl (30 mL). The aqueous phase was extracted with DCM (3 \times 60 mL + 1 \times 30 mL). The organic phases were combined, washed with a saturated aqueous solution of NaHCO3 (100 mL) and H2O (100 mL), dried over anhydrous Na2SO4, filtered, and evaporated to provide a purple oil. The crude product was purified by flash chromatography (*n*-heptane/EtOAc 1:4) to give **10** as a purple oil (0.57 g, 76%). ¹H NMR (400 MHz, CDCl₃) δ 8.62 (d, J = 10.0 Hz, 1H), 8.25 (d, J = 10.0 Hz, 1H), 7.82 (d, J = 4.0 Hz, 1H), 7.25–7.19 (m, 3H), 3.15 (s, 6H), 2.67 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 169.4, 150.7, 140.6, 137.0, 136.9, 136.1, 135.5, 126.8, 126.4, 122.8, 116.7, 37.6* 28.3 (*Methyl carbons of amide only visible in HSQC). HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for C14H16NO 214.1232; Found 214.1233.

N,N-Dimethyl-1-(2,2,2-trifluoroacetyl)azulene-6-carboxamide (11). TFAA (0.30 mL, 2.3 mmol) was added dropwise to a solution of 8 (0.15 g, 0.75 mmol) and Et₃N (0.32 mL, 2.3 mmol) in anhydrous DCM (2.1 mL) under argon on ice bath. The resulting mixture was stirred at 0 °C for 1 h. Then H₂O (15 mL) was added and stirring was continued for an additional 5 min. DCM (45 mL) was added and phases were separated. The organic phase was washed with H₂O (2 \times 30 mL) and brine (30 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a purple solid. The crude product was purified by flash chromatography (n-heptane/EtOAc 2:3) to give 11 as purple, amorphous solid (0.21 g, 94%). ¹H NMR (400 MHz, CDCl₃) δ 9.87 (d, J = 10.4 Hz, 1H), 8.58 (d, *J* = 10.0 Hz, 1H), 8.42–8.40 (m, 1H), 7.72 (dd, *J* = 10.2 Hz, 1.4 Hz, 1H), 7.63 (dd, *J* = 10.0 Hz, 1.6 Hz, 1H) 7.38 (d, *J* = 4.4 Hz, 1H), 3.20 (s, 3H), 2.95 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 176.2 (q, ²J_{C,F} = 34.3 Hz), 171.0, 148.3, 146.8, 143.6, 142.2 (q, ${}^{3}J_{C,F} = 3.7$ Hz), 139.3, 139.1, 129.1, 128.0, 120.7, 118.3, 117.4 (q, ${}^{1}J_{C,F} = 292.9$ Hz), 39.5, 35.5. HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for C15H13NO2F3 296.0898; Found 296.0901.

 N^1, N^1, N^6, N^6 -Tetramethylazulene-1,6-dicarboxamide (12). A 20% solution of phosgene in toluene (1.7 mL, 3.3 mmol) was added to a solution of 8 (0.13 g, 0.65 mmol) in anhydrous toluene (5.0 mL) under argon. The resulting dark blue solution was heated at 80 °C for 5.5 h and then solvents were evaporated. The purple residue was dissolved in anhydous THF (1.1 mL) under argon. The purple solution was cooled to 0 °C and a 2 M solution of dimethylamine in THF (1.6 mL) was slowly added. The reaction mixture was stirred at 0 °C for 30 min. The mixture was diluted with DCM (30 mL) and washed with a 1 M aqueous solution of HCl (15 mL), a saturated aqueous solution of NaHCO₃ (15 mL) and H₂O (15 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a dark blue solid. The crude product was purified by flash chromatography (manual gradient of EtOAc \rightarrow EtOAc/MeOH 97:3) to give **12** as a dark blue, amorphous solid (0.11 g, 60%). ¹H NMR (400 MHz, CDCl₃) δ 8.74 (d, J = 10.0 Hz, 1H), 8.40 (d, J = 10.0 Hz, 1H), 7.97 (d, J = 3.6 Hz, 1H), 7.35 (d, J = 4.0 Hz, 1H), 7.26–7.24 (m, 2H), 3.16 (m, 9H), 2.92 (s, 3H). $^{13}\mathrm{C}$ NMR (101 MHz, CDCl₃) δ 172.3, 168.8, 146.2, 141.7, 137.9, 137.9, 137.6, 136.6, 124.3, 122.9, 122.7, 118.0, 39.5, $H]^+$ HRMS (ESI-QTOF) m/z: [M + 35.4. Calcd for C₁₆H₁₉N₂O₂ 271.1447; Found 271.1447.

General procedure for the synthesis of compounds 13, 14, 15, 16. A mixture of the appropriate azulene derivative and sulfur trioxide pyridine complex (3 equiv) in anhydrous benzene (1.5–10 mL) under argon was heated at 85 °C for 3.5 h. Solvents were evaporated and the residue was dissolved in EtOH (2.2–15 mL). A 1 M aqueous solution of NaOH (2.2–15 mL) was added and the resulting mixture was stirred at 30 °C for 2 h. *n*-Butanol and brine were added to the reaction mixture. Phases were separated and the aqueous phase was extracted with several

times of *n*-butanol. The organic phases were combined and dried over anhydrous Na₂SO₄, filtered and evaporated to provide a crude product, which was purified by flash chromatography.

Sodium 6-methyl-3-(2,2,2-trifluoroacetyl)azulene-1-sulfonate (13). Compound 9 (0.48 g, 2.0 mmol) gave a red solid, which after flash chromatography (manual gradient of MeCN/EtOH 9:1 → 3:2) yielded 13 as a pink, amorphous solid (0.57 g, 84%). ¹H NMR (400 MHz, CD₃OD) δ 9.79 (d, *J* = 10.4 Hz, 1H), 9.30 (d, *J* = 10.8 Hz, 1H), 8.49–8.48 (m, 1H), 7.97–7.94 (m, 2H), 2.86 (s, 3H). ¹³C NMR (101 MHz, CD₃OD) δ 176.7 (q, *J* = 33.8 Hz), 158.1, 145.0, 141.6, 140.9, 140.0, 139.6 (q, *J* = 3.9 Hz), 136.0, 135.1, 134.3, 118.7 (q, *J* = 292.5 Hz), 114.9, 28.2. HRMS (ESI-TOF) *m*/*z*: [M − Na]⁻ Calcd for C₁₃H₈O₄F₃S 317.0096; Found 317.0099.

Sodium 3-(dimethylcarbamoyl)-6-methylazulene-1-sulfonate (14). Compound 10 (0.54 g, 2.5 mmol) gave a purple solid, which after flash chromatography (manual gradient of MeCN/EtOH 4:1 → 3:2) yielded 14 as a purple, amorphous solid (0.70 g, 88%). ¹H NMR (400 MHz, CD₃OD) δ 9.15 (d, *J* = 10.0 Hz, 1H), 8.59 (d, *J* = 10.4 Hz, 1H), 8.10 (s, 1H), 7.57–7.54 (m, 1H), 7.52–7.49 (m, 1H), 3.15 (s, 6H), 2.75 (s, 3H). ¹³C NMR (101 MHz, CD₃OD) δ 170.6, 154.7, 139.3, 138.6, 138.2, 136.2, 135.8, 131.1, 129.9, 129.7, 120.8, 40.1, 35.8, 28.2. HRMS (ESI-QTOF) *m/z*: [M – Na + 2H]⁺ Calcd for C₁₄H₁₆NO₄S 294.0800; Found 294.0800.

Sodium 6-(dimethylcarbamoyl)-3-(2,2,2-trifluoroacetyl)azulene-1-sulfonate (15). Compound 11 (0.23 g, 0.78 mmol) gave a red solid, which after flash chromatography (manual gradient of MeCN → MeCN/EtOH 4:1) yielded 15 as a red, amorphous solid (0.16 g, 53%). ¹H NMR (400 MHz, CD₃OD) δ 9.98 (d, J = 10.4 Hz, 1H), 9.51 (d, J = 10.4 Hz, 1H), 8.67–8.65 (m, 1H), 7.98–7.93 (m, 2H), 3.19 (s, 3H), 2.97 (s, 3H). HRMS (ESI-QTOF) m/z: [M – Na + 2H]⁺ Calcd for C₁₅H₁₃NO₅F₃S 376.0467; Found 376.0466.

Sodium 3,6-bis(dimethylcarbamoyl)azulene-1-sulfonate (16). Compound **12** (0.099 g, 0.37 mmol) gave a blue solid, which after flash chromatography (MeCN/EtOH 8:2) yielded **16** as a blue, amorphous solid (0.11 g, 77%). ¹H NMR (400 MHz, CD₃OD) δ 9.34 (d, J = 10.0 Hz, 1H), 8.76 (d, J = 10.0 Hz, 1H), 8.27 (s, 1H), 7.55–7.52 (m, 1H), 7.49–7.47 (m, 1H), 3.19–3.14 (m, 9H), 2.96 (s, 3H). ¹³C NMR (101 MHz, CD₃OD) δ 173.5, 170.0, 148.7, 140.3, 139.3, 139.1, 138.3, 137.2, 132.9, 125.8, 125.7, 122.6, 39.7, 35.5. HRMS (ESI-QTOF) *m*/*z*: [M – Na + 2H]⁺ Calcd for C₁₆H₁₉N₂O₅S 351.1015; Found 351.1015.

3-Methyl-N-[2-[(3-nitrophenyl)amino] ethyl] benzamide (18). A mixture of 1-fluoro-3-nitrobenzene (1.1 mL, 10 mmol) and ethylenediamine (7.5 mL) was heated at 100 °C for 23 h under argon. The red reaction mixture was quenched with a saturated aqueous solution of NaHCO₃ (70 mL) and it was extracted with DCM (3 \times 40 mL). The organic phases were combined, washed with brine (2 \times 50 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a red solid. The solid was dissolved in anhydrous DCM (40 mL) under argon and DIPEA (3.2 mL, 19 mmol) was added. The resulting mixture was cooled to 0 °C and *m*-toluoyl chloride (1.2 mL, 9.3 mmol) was added dropwise over 5 min. The reaction mixture was stirred at 0 °C for 2 h and then it was diluted with DCM (40 mL). The organic phase was washed with a saturated aqueous solution of NaHCO₃ (3 \times 40 mL) and brine (30 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide an orange solid. The crude product was purified by flash chromatography (*n*-heptane/(EtOAc:EtOH 3:1) 1:1 \rightarrow 2:3) to give **18** as a yellow, amorphous solid (1.6 g, 54%). ¹H NMR (400 MHz, acetone- d_6) δ 7.96 (s, 1H), 7.70–7.65 (m, 2H), 7.45 (t, J = 2.2 Hz, 1H), 7.42–7.39, (m, 1H), 7.35-7.31 (m, 3H), 7.10-7.07 (m, 1H), 5.94-5.90 (m, 1H), 3.69-3.64 (m, 2H), 3.50-3.45 (m, 2H), 2.36 (s, 3H). ¹³C NMR (101 MHz, acetone-d₆) δ 168.2, 150.9, 150.4, 138.8, 135.8, 132.6, 130.8, 129.1, 128.6, 125.1, 119.1, 111.2, 106.4, 44.0, 39.7, 21.31. HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for $C_{16}H_{18}N_3O_3$ 300.1348; Found 300.1351.

3-Methyl-N-[2-[2,2,2-trifluoro-N-(3-nitrophenyl)acetamido] ethyl] benzamide (19). TFAA (0.75 mL, 5.4 mmol) was added dropwise over 5 min to a solution of **18** (1.6 g, 5.4 mmol) and Et₃N (0.75 mL, 5.4 mmol) in anhydrous THF (36 mL) under argon on ice bath. The resulting mixture was stirred at 0 °C for 1 h. A 1 M aqueous solution of HCl (60 mL) was added and the resulting suspension was stirred at 0 °C for 20 min. The reaction mixture was extracted with DCM (100 mL). The phases were separated and the organic phase was washed with H₂O (60 mL) and brine (60 mL), dried over anhydrous Na₂SO₄, filtered and evaporated to provide a brownish solid. The crude product was purified by flash chromatography (DCM/MeOH 99:1 \rightarrow 9:1) to give **19** as a yellow, amorphous solid (2.1 g, 99%). ¹H NMR (400 MHz, (CDCl₃) δ 8.30 (d, J = 8.4 Hz, 1H), 8.23–8.22 (m, 1H), 7.75–7.73 (m, 1H), 7.67–7.63 (m, 1H), 7.55 (s, 1H), 7.51–7.48 (m, 1H), 7.32–71 (m, 2H), 6.66 (s, 1H), 4.11–4.06 (m, 2H), 3.76–3.75 (m, 2H), 2.39 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 168.4, 157.9 (q, ² $J_{C,F} =$ 36.9 Hz), 148.9, 140.2, 138.8, 134.9, 133.8, 132.8, 130.8, 128.7, 127.8, 124.5, 123.9, 123.6, 116.1 (q, ¹ $J_{C,F} =$ 289.3 Hz), 51.4, 38.0, 21.5. HRMS (ESI-QTOF) m/z: [M + H]⁺ Calcd for C₁₈H₁₇N₃O₄F₃ 396.1171; Found 396.1171.

N- [2- [N-(3-Aminophenyl)-2,2,2-trifluoroacetamido] ethyl] -3methylbenzamide (20). Compound 19 (2.1 g, 5.3 mmol) was dissolved in EtOH (45 mL) and Pd/C (0.21 g) was added. The resulting black mixture was stirred under H₂ at room temperature for 4 h. The reaction mixture was filtered through a small pad of Celite with EtOH. Solvents were evaporated to provide a vellow solid. After two flash chromatographic purifications (*n*-heptane/(EtOAc:EtOH 3:1) 9:1 \rightarrow 2:3 and DCM/ MeOH 99:1 \rightarrow 19:1) **20** was obtained as yellow, amorphous solid (1.0 g, 51%). ¹H NMR (400 MHz, CDCl₃) δ 7.58 (s, 1H), 7.53–7.51 (m, 1H), 7.31–7.30 (m, 2H), 7.17 (t, J = 8.0 Hz, 1H), 6.80–6.78 (m, 1H), 6.70-6.68 (m, 1H), 6.61-6.60 (m, 1H), 6.56-6.55 (m, 1H), 4.02-3.99 (m, 2H), 3.74–3.70 (m, 2H), 2.39 (s, 3H). 13 C NMR (101 MHz, CDCl₃) δ 168.1, 158.6 (q, ${}^{2}J_{C,F} = 36.2$ Hz), 147.9, 139.8, 138.6, 134.1, 132.5, 130.5, 128.7, 127.9, 123.9, 117.6, 116.5 (q, ${}^{1}J_{C,F} = 289.1$ Hz), 115.9, 114.2, 51.0, 38.8, 21.5. HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for C18H19N3O2F3 366.1429; Found 366.1429.

General procedure for the synthesis of compounds 21, 22, 23, 24. A mixture of the appropriate azulene derivative (1–2 equiv) and SOCl₂ (0.80–2.0 mL) was heated at 67 °C for 15 min. SOCl₂ was evaporated and the residue was dissolved in anhydrous DCM (1–5 mL) under argon. Compound 20 (1 equiv) and Et₃N (4 equiv) were added, and the resulting mixture was stirred at room temperature for 19–22 h. A 1 M aqueous solution of HCl (5.0–10 mL) was added and then the mixture was stirred for 5 min. The mixture was diluted with DCM (15–50 mL) and the organic phase was washed with a 1 M aqueous solution of HCl (3 × 10–50 mL) and brine (10–50 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a crude product, which was purified by flash chromatography.

3-Methyl-N- [2- [2,2,2-trifluoro-N- [3- [[6-methyl-3-(2,2,2-trifluoroacetyl)azulene] -1-sulfonamido] phenyl] acetamido] ethyl] benzamide (21). Compound **13** (0.20 g, 0.60 mmol) and **20** (0.11 g, 0.30 mmol) gave a dark red solid, which after flash chromatography (DCM/MeOH 99:1) yielded **21** as an orange, amorphous solid (0.062 g, 31%). ¹H NMR (400 MHz, acetone- d_6) δ 9.80 (d, J = 10.4 Hz, 1H), 9.42 (s, 1H), 9.20 (d, J = 10.4 Hz, 1H), 8.45–8.43 (m, 1H), 8.15 (d, J = 10.8 Hz, 1H), 8.09 (d, J = 10.4 Hz, 1H), 7.76–7.72 (m, 1H), 7.57–7.53 (m, 2H), 7.31–7.22 (m, 6H), 3.84 (t, J = 5.6 Hz, 2H), 3.47–3.43 (m, 2H), 2.89 (s, 3H), 2.35 (s, 3H). HRMS (ESI-QTOF) m/z: [M + H]⁺ Calcd for C₃₁H₂₆N₃O₅F₆S 666.1497; Found 666.1497.

N,*N*,6-Trimethyl-3- [*N*- [3- [2,2,2-trifluoro-*N*- [2-(3-methylbenzamido)ethyl] acetamido] phenyl] sulfamoyl] azulene-1-carboxamide (22). Compound 14 (0.32 g, 1.0 mmol) and 20 (0.18 g, 0.50 mmol) gave a brown solid, which after flash chromatography (manual gradient of EtOAc/DCM/Toluene 40:59:1 → EtOAc) yielded 22 as a red solid (0.14 g, 44%). ¹H NMR (400 MHz, CDCl₃) δ 8.91 (d, *J* = 10.4 Hz, 1H), 8.63 (d, *J* = 10.0 Hz, 1H) 8.29 (s, 1H), 7.97 (s, 1H), 7.57 (s, 1H), 7.50 (d, *J* = 6.8 Hz, 1H), 7.42–7.23 (m, 6H), 7.18–7.11 (m, 2H), 6.89–6.88 (m, 2H), 3.73–3.69 (m, 2H), 3.38–3.34 (m, 2H), 3.10–2.97 (m, 6H), 2.66 (s, 3H), 2.35 (s, 3H). HRMS (ESI-QTOF) *m*/*z*: [M + H]⁺ Calcd for C₃₂H₃₂N₄O₅F₃S 641.2045; Found 641.2045.

N,N-Dimethyl-1-[N-[3-[2,2,2-trifluoro-N-[2-(3-

methylbenzamido)ethyl] acetamido] phenyl] sulfamoyl] -3-(2,2,2trifluoroacetyl)azulene-6-carboxamide (23). Compound 15 (0.15 g, 0.38 mmol) and 20 (0.069 g, 0.19 mmol) gave a red solid, which after two flash chromatographic purifications (manual gradient of DCM \rightarrow DCM/MeOH 49:1 and DCM/EtOAc 5:5 \rightarrow 3:7) yielded 23 as a red solid (0.053 g, 38%). ¹H NMR (400 MHz, CDCl₃) δ 9.95 (d, J = 10.4 Hz, 1H), 9.27 (d, J = 10.4 Hz, 1H), 8.59–5.51 (m, 2H), 7.86 (d, J = 10.0 Hz, 1H), 7.74 (d, J = 10.0 Hz, 1H), 7.46 (s, 1H), 7.42–7.40 (m, 1H), 7.26–7.22 (m, 4H), 7.07 (s, 1H), 6.10–6.99 (m, 1H), 6.76 (t, J = 5.0 Hz, 1H), 3.87–3.85 (m, 2H), 3.54–3.52 (m, 2H), 3.18 (s, 3H), 2.86 (s, 3H), 2.32 (s, 3H). $^{13}\mathrm{C}$ NMR (101 MHz, CDCl₃) 176.4 (q, ${}^{2}J_{C,F}$ = 35.3 Hz), 170.0, 168.5, 157.9 (q, ${}^{2}J_{C,F} = 36.3$ Hz), 150.7, 145.3, 143.7 (q, ${}^{3}J_{C,F} = 3.1$ Hz), 141.8, 141.4, 139.6, 139.4, 138.7, 138.2, 133.7, 132.7, 132.0, 131.1, 130.7, 128.7, 127.7, 125.2, 125.0, 124.0, 122.7, 122.0, 116.8 (q, ${}^{1}J_{C,F} = 292.6$ Hz), 116.1 (q, ${}^{1}J_{C,F} = 289.2$ Hz), 115.9, 50.9, 39.3, 38.2, 35.6, 21.4. HRMS (ESI-QTOF) m/z: $[M + H]^+$ Calcd for C₃₃H₂₉N₄O₆F₆S 723.1712; Found 723.1714.

*N*¹,*N*⁶,*N*⁶-Tetramethyl-3-[*N*-[3-[2,2,2-trifluoro-*N*-[2-(3-methylbenzamido)ethyl] acetamido] phenyl] sulfamoyl] azulene-1,6-dicarboxamide (24). Compound 16 (0.097 g, 0.26 mmol) and 20 (0.095 g, 0.26 mmol) gave a blue solid, which after flash chromatog-raphy (manual gradient of DCM → DCM/MeOH 24:1) yielded 24 as a purple solid (0.032 g, 18%). ¹H NMR (400 MHz, CDCl₃) δ 9.10 (d, *J* = 10.0 Hz, 1H), 8.83 (d, *J* = 9.2 Hz, 1H), 8.39 (s, 1H), 8.10 (s, 1H), 7.57 (s, 1H), 7.46–7.39 (m, 3H), 7.29–7.27 (m, 2H), 7.24–7.16 (m, 2H), 7.02–6.92 (m, 3H), 3.76–3.73 (m, 2H), 3.40–3.37 (m, 2H), 3.15–2.86 (m, 12H), 2.37 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 170.9, 168.3, 167.1, 157.5 (q, ²*J*_{C,F} = 36.2 Hz), 148.8, 140.8, 139.6, 139.5, 138.9, 138.7, 138.5, 137.9, 137.4, 134.0, 132.5, 130.3, 128.6, 128.0, 126.8, 126.8, 124.8, 124.1, 122.8, 122.8, 122.2, 121.8, 116.1 (q, ¹*J*_{C,F} = 289.3 Hz), 50.8, 39.4, 37.9, 35.5, 21.5. HRMS (ESI-QTOF) *m/z*: [M + H]⁺ Calcd for C₃₄H₃₅N₅O₆F₃S 698.2260; Found 698.2259.

General procedure for the synthesis of compounds 25, 27, 28, 29. An appropriate azulene derivative was suspended in a mixture of MeOH (5.5–22 mL) and H₂O (0.55–2.2 mL). K₂CO₃ (30 equiv) was added and the resulting mixture was stirred at room temperature for 3 h. The reaction mixture was diluted with EtOAc (60 mL) and washed with H₂O (2×30 mL) and a saturated aqueous solution of NaHCO₃ (30 mL), dried over anhydrous Na₂SO₄, filtered and evaporated to provide a crude product, which was purified by flash chromatography.

3-Methyl-N-[2-[[3-[[6-methyl-3-(2,2,2-trifluoroacetyl)azulene]-1-sulfonamido] phenyl] amino] ethyl] benzamide (25). Compound 21 (0.052 g, 0.078 mmol) gave a brown solid, which after flash chromatography (n-heptane/(EtOAc:EtOH 3:1) 3:2) yielded 25 as a an orange, amorphous solid (0.017 g, 39%). ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.19 (s, 1H), 9.69 (d, J = 10.4 Hz, 1H), 9.17 (d, J = 10.4 Hz, 1H), 8.41 (t, J = 1.8 Hz, 1H), 8.38-8.37 (m, 1H), 8.18-8.15 (m, 1H), 8.11-8.08 (m, 1H), 7.63–7.59 (m, 2H), 7.34–7.32 (m, 2H), 6.83 (t, J = 8.0 Hz, 1H), 6.34 (t, J = 2.0 Hz, 1H), 6.24-6.22 (m, 2H), 5.73 (t, J = 5.6 Hz, 1H), 3.31-3.27 (m, 2H), 3.03-2.99 (m, 2H), 2.84 (s, 3H), 2.35 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 174.5 (q, ${}^2J_{C,F} = 33.5$ Hz), 166.5, 159.2, 149.3, 143.2, 140.4, 140.2 (q, ${}^{3}J_{C,F} = 4.1$ Hz), 139.9, 138.3, 138.2, 137.5, 137.1, 135.9, 134.4, 131.7, 129.5, 128.2, 127.7, 125.0, 124.3, 116.8 (q, ${}^{1}J_{C, F} = 293.6 \text{ Hz}$), 113.4, 107.9, 107.9, 103.9, 42.3, 38.5, 27.8, HRMS (ESI-QTOF) *m/z*: [M $+ H^{+}$ Calcd 20.9. for C₂₉H₂₇N₃O₄F₃S 570.1674; Found 570.1679.

N,N,6-Trimethyl-3-[N-[3-[[2-(3-methylbenzamido)ethyl]

amino] phenyl] sulfamoyl] azulene-1-carboxamide (27). Compound 22 (0.11 g, 0.17 mmol) gave a red solid, which after flash chromatography (manual gradient of DCM → DCM/MeOH 97:3) yielded 27 as a red, amorphous solid (0.049 g, 53%). ¹H NMR (400 MHz, DMSO- d_6) δ 9.97 (s, 1H), 9.04 (d, J = 10.0 Hz, 1H), 8.64 (d, J = 10.4 Hz, 1H), 8.44 (s, 1H), 8.00 (s, 1H), 7.67–7.61 (m, 4H), 7.33 (s, 2H), 6.79 (t, J = 7.8 Hz, 1H), 6.36 (s, 1H), 6.23–6.17 (m, 2H), 5.69–5.67 (m, 1H), 3.34 (m, 2H), 3.02–2.91 (m, 8H), 2.72 (s, 3H), 2.35 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 166.5, 166.1, 155.0, 149.2, 138.9, 138.7, 138.4, 137.5, 136.5, 136.4, 135.1, 134.4, 131.7, 130.5, 130.1, 129.3, 128.2, 127.7, 124.3, 121.1, 120.9, 107.4, 107.1, 103.4, 42.4, 38.6, 27.6, 20.9. HRMS (ESI-QTOF) $m/z:~[M~+~H]^+$ Calcd for $C_{30}H_{33}N_4O_4S$ 545.2222; Found 545.2222.

N,N-Dimethyl-1-[N-[3-[[2-(3-methylbenzamido)ethyl] amino] phenyl] sulfamoyl] -3-(2,2,2-trifluoroacetyl)azulene-6-carboxamide (28). Compound 23 (0.038 g, 0.053 mmol) gave a red solid, which after flash chromatography (manual gradient of DCM \rightarrow DCM/ MeOH 49:1) yielded 28 as a red, amorphous solid (0.025 g, 76%). 1 H NMR (400 MHz, CDCl₃) δ 9.81 (d, J = 10.4 Hz, 1H), 9.07 (d, J = 10.0 Hz, 1H), 8.65-8.64 (m, 1H), 7.98 (s, 1H), 7.71-7.68 (m, 1H), 7.60-7.57 (m, 1H), 7.51 (s, 1H), 7.46–7.44 (m, 1H), 7.24–7.17 (m, 2H), 7.01–6.98 (m, 1H), 6.85 (t, J = 8.0 Hz, 1H), 6.40–6.38 (m, 1H), 6.27–6.24 (m, 2H), 3.50-3.46 (m, 2H), 3.15 (s, 3H), 3.11-3.08 (m, 2H), 2.84 (s, 3H), 2.29 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 176.4 (q, ² $J_{C,F}$ = 35.1 Hz), 170.1, 168.9, 150.23, 148.9, 145.0, 143.6 (q, ${}^{3}J_{C,F} = 3.6$ Hz), 141.6, 139.6, 138.5, 137.5, 134.0, 132.5, 131.7, 130.7, 130.1, 128.5, 127.9, 125.9, 124.2, 116.8 (q, ${}^{1}J_{C, F} = 291.9$ Hz), 115.7, 111.5, 111.5, 110.6, 106.6, 44.0, 39.4, 39.4, 35.5, 21.4. HRMS (ESI-QTOF) m/z: [M + H]⁺ Calcd for C31H30N4O5F3S 627.1889; Found 627.1887.

*N*¹,*N*⁶,*N*⁶-Tetramethyl-3- [*N*-[3-[[2-(3-methylbenzamido) ethyl] amino] phenyl] sulfamoyl] azulene-1,6-dicarboxamide (29). Compound 24 (0.029 g, 0.042 mmol) gave a purple solid, which after flash chromatography (manual gradient of DCM → DCM/MeOH 24:1) yielded 29 as a purple, amorphous solid (0.010 g, 40%). ¹H NMR (400 MHz, CDCl₃) δ 9.02 (d, *J* = 10.4 Hz, 1H), 8.81 (d, *J* = 10.0 Hz, 1H), 8.13 (s, 1H), 7.78–7.74 (m, 1H), 7.64 (s, 1H), 7.59–7.58 (m, 1H), 7.36–7.31 (m, 3H), 7.26–7.23 (m, 2H), 6.83 (t, *J* = 8.0 Hz, 1H), 6.43 (d, *J* = 7.6 Hz, 1H), 6.21 (d, *J* = 8.0 Hz, 1H), 5.89 (s, 1H), 3.28–3.26 (m, 2H), 3.14 (s, 6H), 2.91–2.84 (m, 8H), 2.35 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 171.0, 168.7, 167.3, 148.8, 148.5, 140.8, 139.4, 139.0, 138.4, 138.1, 138.0, 137.6, 134.3, 132.4, 129.9, 128.4, 128.1, 126.4, 124.4, 122.6, 122.6, 111.9, 110.7, 106.3, 44.0, 39.4, 39.3, 35.5, 21.5. HRMS (ESI-QTOF) *m/z*: [M + H]⁺ Calcd for C₃₂H₃₆N₅O₅S 602.2437; Found 602.2437.

N-[2-[*N*-[3-[(5-Bromo-2-methoxyphenyl)sulfonamido] phenyl]-2,2,2-trifluoroacetamido] ethyl]-3-methylbenzamide

(30). 5-Bromo-2-methoxybenzenesulfonyl chloride (0.78 g, 2.7 mmol) was added to a yellow solution of 20 (1.0 g, 2.7 mmol) and Et₃N (0.76 mL, 5.4 mmol) in anhydrous DCM (11 mL) under argon. The resulting mixture was stirred at room temperature for 48 h. A 1 M aqueous solution of HCl (20 mL) was added and the reaction mixture was stirred at room temperature for an additional 15 min. The mixture was extracted with DCM (30 mL). The organic phase was washed with a 1 M aqueous solution of HCl (20 mL) and brine (20 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide an orange oil. The crude product was purified by flash chromatography (n-heptane/(EtOAc:EtOH 3:1) 9:1 \rightarrow 2:3) to give **30** as a yellowish, amorphous solid (0.96 g, 57%). ¹H NMR (300 MHz, CDCl₃) δ 7.91 (d, J = 2.4 Hz, 1H), 7.56–7.48 (m, 4H), 7.32–7.20 (m, 4H), 7.09 (s, 1H), 7.00 (d, J = 7.8 Hz, 1H), 6.84 (d, J = 9.0 Hz, 1H), 6.63-6.60 (m, 1H), 3.96-3.93 (m, 5H), 3.65-3.60 (m, 2H), 2.39 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 168.2, 158.2 (q, J = 35.9Hz), 155.4, 139.6, 138.7, 138.0, 137.9, 134.1, 133.5, 132.6, 130.7, 128.7, 127.8, 127.5, 125.1, 124.0, 122.1, 120.9, 116.2 (q, J = 286.8 Hz), 114.2, 112.9, 56.9, 51.1, 38.3, 21.5. HRMS (ESI-QTOF) *m/z*: [M + H]⁺ Calcd for C₂₅H₂₄N₃O₅BrF₃S 614.0572; Found 614.0576.

4'-Methoxy-*N*,*N*-dimethyl-3'-[*N*-[3-[[2-(3-methylbenzamido) ethyl] amino] phenyl] sulfamoyl] [1,1'-biphenyl] -3-carboxamide (1). A mixture of 30 (0.95 g, 1.6 mmol), 3-(*N*,*N*-dimethylaminocarbonyl)benzene boronic acid (0.36 g, 1.9 mmol), Pd(dppf)Cl₂ (0.057 g, 0.078 mmol) and Na₂CO₃ (0.66 g, 6.2 mmol) in 1,4-dioxane (4.8 mL) and H₂O (4.8 mL) was heated at 100 °C for 48 h. The reaction mixture was filtered through a small pad of Celite with EtOAc (150 mL). The organic phase was washed brine (40 mL), dried over anhydrous Na₂SO₄, filtered, and evaporated to provide a brown oil. After two flash chromatographic purifications (*n*-heptane/(EtOAc:EtOH 3:1) 9:1 → EtOAc: EtOH 3:1 and EtOAc) **1** was obtained as an off-white, amorphous solid (0.48 g, 53%). ¹H NMR (400 MHz, DMSO- d_6) δ 9.79 (s, 1H), 8.44 (t, J = 5.6 Hz, 1H), 7.99 (d, J = 2.4 Hz, 1H), 7.89 (dd, J = 8.8, 2.4 Hz, 1H), 7.67–7.64 (m, 2H), 7.62–7.61 (m, 1H), 7.58–7.57 (m, 1H), 7.48 (t, J = 7.8 Hz, 1H), 7.36–7.32 (m, 3H), 7.27 (d, J = 8.8 Hz, 1H), 6.87 (d, J = 8.0 Hz, 1H), 6.42–6.41 (m, 1H), 6.33–6.31 (m, 1H), 6.22 (dd, J = 8.0, 1.6 Hz, 1H), 5.75 (t, J = 5.8 Hz, 1H), 3.94 (s, 3H), 3.36–3.33 (m, 2H), 3.08–3.03 (m, 2H), 2.99 (s, 3H), 2.90 (s, 3H), 2.34 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 169.8, 166.5, 156.2, 149.2, 138.5, 138.4, 137.5, 137.4, 134.4, 133.0, 131.7, 131.2, 129.4, 129.1, 128.1, 128.0, 127.7, 127.3, 127.1, 125.8, 124.6, 124.3, 113.6, 107.3, 107.3, 103.3, 56.4, 42.4, 39.0, 38.6, 34.7, 20.9. HRMS (ESI-QTOF) *m*/*z*: [M + H]⁺ Calcd for C₃₂H₃₅N₄O₅S 587.2328; Found 587.2328.

4.2. Pharmacological analysis

Materials: Human orexin-A was from NeoMPS (Strasbourg, France), *N*-(2-methyl-6-benzoxazolyl)-*N*'-1,5-naphthyridin-4-yl urea (TCS-1102) and *N*-biphenyl-2-yl-1-[[(1-methyl-1*H*-benzimidazol-2-yl)sulfanyl] acetyl]-L-prolinamide (SB334867) from Tocris Bioscience (Bristol, UK), and probenecid from Sigma–Aldrich (St.Louis, MO, USA).

Cell culture and media: CHO-K1 cells expressing human OX₁ and OX₂ receptors cells (CHO-hOX₁ and CHO-hOX₂, respectively),^{31–32} and control CHO-K1 cells (not expressing orexin receptors,ctrl CHO cells) were cultured in Ham's F12 medium (Gibco/Life Technologies, Paisley, UK) + supplements on plastic cell culture dishes (56 cm² bottom area; Greiner Bio-One GmbH, Frickenhausen, Germany) as previously described.^{33–34} Black, clear-bottom half-area Cellstar µClear 96-well cell culture plates (Greiner Bio-One GmbH; Frickenhausen, Germany), coated with polyethyleneimine (25 µg/mL for 1 h at 37 °C; Sigma-Aldrich, St. Louis, MO, USA), were used for Ca²⁺ measurements. Hepes-buffered medium (HBM) consisting of 137 mM NaCl, 5 mM KCl, 1.2 mM MgCl₂, 0.44 mM KH₂PO₄, 4.2 mM NaHCO₃, 1 mM CaCl₂, 10 mM glucose, 20 mM HEPES and 0.1% (w/v) stripped bovine serum albumin (adjusted to pH 7.4 with NaOH) was used as the basic experimental medium.

 Ca^{2+} elevation: The compound-induced Ca^{2+} elevations were measured as described before.³⁵ In brief, CHO-hOX₁, CHO-hOX₂ or control CHO cells were plated on Cellstar plates, 1.5×10^4 cells per well. After 24-hour incubation, the cell culture medium was exchanged to the loading solution consisting of FLIPR Calcium 4 Assay Kit (Molecular Devices, Sunnyvale, CA, USA) dissolved in and diluted with HBM + 1.0 mM probenecid. After subsequent 60 min incubation at 37 °C, intracellular Ca²⁺ levels were measured at 37 °C using a FlexStation 3 fluorescence plate reader (Molecular Devices, excitation at 485 nm, emission at 525 nm). The wells were measured for 150 s with 30 s of baseline before stimulation and measured approximately every 1.3 s. After the measurement, 0.3 nM orexin-A was added to the wells and the plate was re-measured as described above.

Data analysis: Data were visualized using Graph Pad Prism 8 and analysed using Microsoft Excel software. Concentration–response curves were fit using three-parameter nonlinear regression equation. The data points represent the mean \pm s.e.m. of 3–5 individual experiments (biological replicates) performed typically in triplicates (technical replicates) unless otherwise stated. All presented Ca²⁺ responses were separately normalized to the control maximum orexin-A or ATP responses for each independent experiment before averaging. The background response (vehicle addition) was subtracted from all data values (both in the single-concentration screen and concentration–response curves) prior to the analysis. Crude estimates of the K_i values were calculated from the inhibition-% values by first estimating the IC₅₀values and then converting these to K_i utilizing the Cheng–Prusoff equation with cooperativity coefficients. The rational of this follows the Equation (1), which also lies behind the Cheng–Prusoff equation.

$$(100 - \%inhibition)response_{noinhibitor} = \frac{[orexin - A]^n \times (100 - \%inhibition)}{[orexin - A]^n + EC_{50}^n}$$
$$= response_{withinhibitor}$$
$$= \frac{[orexin - A]^n \times 100}{[orexin - A]^n + EC_{50}^n \left(1 + \frac{|I|}{K_i}\right)}$$
(1)

4.3. Computational studies

Homology modeling: Our goal was to build a model of OX₁ in an active, agonist-bound, conformation suitable for a subsequent docking study with a small molecule agonist. Thus, we used the active-state cryo-EM structure of OX₂ (PDB ID: 7L1V) as a template. The binding sites of OX1 and OX2 are nearly identical as there are only 2-4 conservative differences among the binding site residues, depending on how the site is framed (S/T^{2.61}, S/T^{23.49}, A/T^{3.33}, S/T^{45.48}). Prior to building the models, the ligand and residues within 5 Å of the ligand in the OX_2 structure were refined using Prime minimization³⁶ ³⁷ (Schrödinger Release 2020-4). Based on this minimized OX₂ structure, ten models of OX_1 were built consisting of residues $Trp47^{1.33}$ - $Trp243^{5.66}$ and Glu286^{6.23}-Ala372^{8.57} using MODELLER 10.1³⁸ with default settings. The N- and C-termini and the ICL3 were not modeled in the crvo-EM structure 7L1V, so they were not included in our models. The model used for the docking study was selected based on the similarity of the binding site residue conformers between the template and the model to facilitate the docking.

Molecular docking of 27: The homology model selected for docking was first preprocessed by Protein Preparation Wizard of Schrödinger Maestro (Schrödinger Release 2020–4) with default parameters, and the hydrogen bond assignments were optimized. Control agonist **31** was docked to the refined OX₂ cryo-EM structure and **27** to the OX₁ homology model using Glide SP^{39-41} with default settings and a grid defined by the location of the agonist **31** in the OX₂ complex. The docking poses were analyzed visually in Schrödinger Maestro, and further visualization and figure preparation was carried out with PyMOL 2.3.3 and Gimp 2.10.14. **27** docking poses and homology models are provided as supplementar y files.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.bmc.2023.117325.

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