

This is a self-archived version of an original article. This version may differ from the original in pagination and typographic details.

Author(s): Lemma, Tibebe; Wang, Jin; Arstila, Kai; Hytönen, Vesa; Toppari, Jussi

Title: Identifying yeasts using surface enhanced Raman spectroscopy

Year: 2019

Version: Accepted version (Final draft)

Copyright: © 2019 Published by Elsevier B.V.

Rights: CC BY-NC-ND 4.0

Rights url: https://creativecommons.org/licenses/by-nc-nd/4.0/

Please cite the original version:

Lemma, T., Wang, J., Arstila, K., Hytönen, V., & Toppari, J. (2019). Identifying yeasts using surface enhanced Raman spectroscopy. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 218, 299-307. https://doi.org/10.1016/j.saa.2019.04.010

Accepted Manuscript

Identifying yeasts using surface enhanced Raman spectroscopy

Tibebe Lemma, Jin Wang, Kai Arstila, Vesa P. Hytönen, J. Jussi Toppari

PII: S1386-1425(19)30385-3

DOI: https://doi.org/10.1016/j.saa.2019.04.010

Reference: SAA 17018

To appear in: Spectrochimica Acta Part A: Molecular and Biomolecular

Spectroscopy

Received date: 22 January 2019
Revised date: 5 April 2019
Accepted 7 April 2019

date: 7 April 2019

Please cite this article as: T. Lemma, J. Wang, K. Arstila, et al., Identifying yeasts using surface enhanced Raman spectroscopy, Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, https://doi.org/10.1016/j.saa.2019.04.010

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



Identifying yeasts using surface enhanced Raman spectroscopy

Tibebe Lemma ^{1,*}, Jin Wang ², Kai Arstila ³, Vesa P. Hytönen ^{4,} and J. Jussi Toppari ³

- ¹ Faculdade de Clências e Tecnologia (FCT)-Universidade Estadual Paulista (UNESP)-Presidente Prudente, SP,19060-900, Brazil
- ² Institute of Intelligent Machines, Chinese Academy of Sciences, Hefei, Anhui 230031, P. R. China
- ³ NanoScience Center, Department of Physics, P.O. Box 35 (YN), FI-40014, University of Jyväskylä, Finland
- ⁴ Faculty of Medicine and Health Technology and BioMediTech, Tampere University, Arvo Ylpön katu 34, FI-33520 Tampere, Finland and Fimlab Laboratories, Biokatu 4, FI-33520 Tampere, Finland
- * Correspondence: tlemma@gmail.com (T.L.)

Abstract: The molecular fingerprints of yeasts Saccharomyces cerevisiae, Dekkera bruxellensis, and Wickerhamomyces anomalus (former name Pichia anomala) have been examined using surface-enhanced Raman spectroscopy (SERS) and helium ion microscopy (HIM). The SERS spectra obtained from cell cultures (lysate and non-treated cells) distinguish between these very closely related fungal species. Highly SERS active silver nano-particles suitable for detecting complex biomolecules were fabricated using a simple synthesis route. The yeast samples mixed with aggregated Ag nanoparticles yielded highly enhanced and reproducible Raman signal owing to the high density of the hot spots at the junctions of two or more Ag nanoparticles and enabled to differentiate the three species based on their unique features (spectral fingerprint). We also collected SERS spectra of the three yeast species in beer medium to demonstrate the potential of the method for industrial application. These findings demonstrate the great potential of SERS for detection and identification of fungi species based on the biochemical compositions, even in a chemically complex sample.

Keywords: Silver nanoparticle; aggregate; SERS; yeast; helium ion microscopy

1. Introduction

Yeast is the most important part of the brewing fermentation process. Yeasts, such as *Saccharomyces cerevisiae* (baker's yeast or brewer's yeast) and *Dekkera bruxellensis* (also known as *Brettanomyces bruxellensis*) are commonly used in food production, baking, wine, and beer making. Preparations of baker's yeasts are also used as dietary, nutrient supplements because of their high contents of B vitamins, proteins, peptides, amino acids, and trace minerals. In addition to fermenting and flavoring foods, baker's yeast also plays an important role as a model organism in the field of biochemistry, genetics and molecular biology, and it is important commercially and industrially [1]. Yeasts can also be used as an oral, biotherapeutic agent to treat patients with severe cases of diarrhea and other gastrointestinal

disorders [2]. However, yeasts are also responsible for the majority of food and beverage spoilage, which brings on their enormous commercial and economic significance. While there is significant diversity in the yeast species associated with food and beverage spoilage, *Saccharomyces cerevisiae* (wild yeast), *Dekkera bruxellensis and Wickerhamomyces anomalus* (former name *Pichia anomala*) are responsible for the vast majority of spoilage in the winery in cases not intentionally used in the fermentation process [3].

Most of the organisms in the fungi kingdom are multicellular, however, yeast, a typical fungus, is a single-celled eukaryote organism, and thus it contains membrane-bound organelles, such as a nucleus, endomembrane system, and mitochondria. A yeast cell is surrounded by a rigid wall which is ~250 nm thick and constituting about 15-30% of the total dry mass of the cell [4, 5]. The cell wall of yeast is an important component which is responsible for mechanical strength (barrier) of the cell. The wall is composed of mannoprotein [1, 6], glucan, beta-glucan, and chitin (a long-chain polymer of an *N*-acetylglucosamine), which contributes to the insolubility of the fibers. In addition, some proteins are disulfide-bonded to other cell wall proteins [6] which make yeast cell highly diverse in their protein and other components. A single unbudded yeast cell is about 5-10 µm in diameter and is usually oval, cylindrical or spherical in shape.

Raman spectroscopy is a powerful light scattering technique used to investigate a very wide range of samples types (organic and inorganic), including biomolecules and microorganisms. This technique is especially attractive because it demonstrates the ability to perform real-time non-invasive (minimally invasive) imaging without the use of biomarkers, its relativity low cost, its portability and the capability of acquired a lot of data in a short time. Samples for conventional Raman spectroscopy measurement can be either in bulk or in microscopic amounts and no special sample preparation required. The technique has been used extensively in recent years for the rapid and accurate identification of clinically relevant microorganisms both remotely and *in situ*, thus appears to be a convenient technique for the study of microbes. Despite such advantages, the Raman effect is inherently very weak where approximately one in ~10⁵ photons are inelastically scattered from a sample. Compared with luminescence-based processes, normal Raman spectroscopy has a small cross-section (~10 -30 cm² molecule⁻¹) [7], thus precluding the possibility of biomolecules detection at extremely low concentrations and difficult to disentangle the complex overlapping bands from biological macromolecules. Therefore, many Raman based techniques have been developed to deal with the short-coming mentioned above. Surface Enhanced Raman Spectroscopy (SERS), Resonance Raman Spectroscopy (RRS) are a few such techniques that have been

developed for a wide range of applications, including microbes and cellular detection, *in situ* monitoring of biological macromolecules at a physiological concentration.

Surface-enhanced Raman spectroscopy (SERS) is a modern spectroscopy technique where a weak Raman signal of a functional group of a Raman –active molecule is greatly enhanced when the molecule is adsorbed on the surface of specially prepared coinage metals, in particular, Ag, Au, or Cu [8, 9]. Two different phenomena are responsible for the SERS enhancement: firstly and mostly, the classical near-field electromagnetic effect and secondly, the chemical effect [10-15], which allows detecting the local composition of biomolecules at extremely low concentrations (down to a single molecule) [16, 17]. Altogether, the SERS technique provides enhancement up to several orders of magnitude if the excitation source equals with the energy of an electronic transition. Signal enhancements as high as 10¹⁴ times over normal Raman scattering have been reported [16, 18], which gives SERS spectroscopy a huge potential for various applications where sensitivity is needed in molecular detection. Thanks to the ability to provide direct information on chemical composition, the SERS technique has received tremendous attention in various research fields including biomedical, biological sensors and material science.

Molecular discrimination of different microbes is very important when monitoring changes in cellular structure or during classification of cells and bacteria. SERS, which is becoming popular for the chemical characterization of biological species, is an ideal tool to investigate the class of compounds, which constitutes an intracellular and extracellular environment of microbes, especially bacteria and yeast [19-21]. The detection of yeast and bacteria using Raman and SERS has been reported in several publications [19, 20, 22-29], resulting in a wealth of information about the chemical signatures of different biological components associated with DNA, proteins, lipids and RNA and other biomolecules. However, in most cases, the fingerprint peaks have been difficult to interpret due to extremely low physicochemical concentration and adverse influence of many kinds of background impurities, often encountered in nature, resulting in anomalous peaks (multiple peaks) and distortions. In particular, the interaction between the different biomolecules with their surrounding matrix becomes more complicated in lysate cell, since the concentration may decrease lower than those found in the physiological conditions. Thus, in order to avoid the matrix influence, a careful substrate preparation needs to be adapted before the sample is subjected to a SERS measurement. In order to clearly identify species, some studies have combined Raman with techniques like chemometrics [25].

In this contribution, we report the SERS spectra of yeasts (lysate and non-treated cells) mixed with highly SERS active aggregated Ag nanoparticles. In this method, the SERS spectra associated with the

molecular features of the yeast cells give rise to intense bands that provide information about the chemical composition of the fungi and enable us to detect molecular (low concentration) differences between yeast species. Furthermore, this enables us to differentiate yeast species based on their minor molecular variation without any additional methods, and also within a complicated matrix. In addition, high-resolution images of the mixture of fungus and aggregated Ag nanoparticles are taken with helium ion microscopy (HIM). These images provide the cell morphology and the distribution of the Ag nanoparticles on the surface of the yeast cells, which can be attributed to differences in cell adhesion and further to some changes in the observed spectra. This study suggests that such an approach combining of SERS and HIM may provide a better understanding of the structural complexity of yeast, thus enabling development of powerful detection techniques of these industrially relevant species without using any coupling agents. These results open possibilities for developing sensitive and robust biosensors for controlling the food quality and safety.

2. Materials and Methods

2.1 Reagents

Unless stated otherwise, all chemicals, such as silver nitrate, trisodium citrate, sodium chloride, magnesium chloride, sodium hydroxide, 2-propanol and ethanol, were purchased from Sigma–Aldrich (St. Louis, MO, USA). All the chemicals were of analytical grade and used as received.

2.2 Experimental setup

The measurements were made on a Senterra dispersive Raman Microscope (BRUKER, Sweden). Confocal Raman spectroscopy was used to acquire Raman and SERS spectra of bacteria-nanoparticles samples. Raman spectra were obtained using a near-infrared diode laser (785 nm) as an excitation and collecting the Raman scattering in 180° geometry by a Peltier cooled (–70 °C) charge-coupled device (CCD) camera (255×1024 pixels), focusing only within the fingerprint regions (400-2000 cm⁻¹). For all biological samples, a near-IR (785 or 800 nm) laser is a superior light source in order to avoid fluorescence and photodecomposition interference observed with all visible range excitation sources. The spectrometer was equipped with a diffraction grating of 1200 grooves/mm and the slit gave a spectral resolution of 2 cm⁻¹. The laser power at the sample ranged from 1 to 10 mW and the acquisition time ranged from 1 to 5 seconds. Microscope objectives (20, 50 and 100x, NA 0.40, 0.75 and 0.9) were used to focus the laser light on the sample, which consisted of 20 to 50 μL drop of mixed sample on a cleaned microscopic slide. The area of the laser spot on the samples was approximately 1 μm in diameter (when using 100x objective). The system was calibrated using built-in templates and internal Raman standards.

Varying amount of yeast cells (non-treated and lysate) mixed with 250 μ L of aggregated Ag nanoparticles and incubated for 30 min. For the first beer sample, 10 μ L of beer was applied into a 300 μ L solution containing Ag nanoparticles and yeast cells, and for the second sample, 10 μ l of yeast-AgNP-solution was added to 400 μ L of beer. The mixture was further incubated for another 30 min, and then casted on a clean glass slide for the SERS measurement.

The SERS spectra were collected by mapping a small area on the slide ($100 \mu m \times 100 \mu m$). Examples of recorded spectra for each species without any spectral treatments are shown in Supporting Material Figures S2-S4. They indicate high reproducibility of the method. Then the spectra from each collection were averaged and baseline-corrected by rubber band algorithm (using defaults of Origin 8 software). All the measurements were performed in triplicate for each yeast cell.

To determine the morphology of the silver particles, transmission electron microscope (TEM) images were taken on a JEOL-1400 operating at 80 kV. The samples were prepared by depositing and drying 5 µL of the Ag nano-aggregates on 200 TEM mesh copper grids (FORMVAR Carbon Film). Absorption spectra were recorded from the solution using a UV/Vis spectrometer (Lambda 850, PerkinElmer, Massachusetts, USA). Visual inspection of the nanoparticles is performed on the TEM image to determine the shape and the size of the AgNPs.

The yeast samples were imaged with an Orion NanoFab (Carl Zeiss AG, Germany) helium ion microscope operating at 30 keV accelerating voltage and 0.2-0.4 pA beam current. Charge neutralization with a flood gun allowed imaging non-conductive biological samples without using a conductive metal or carbon film on the sample. Charge neutralization was done after each line scan measured with $0.5~\mu s$ dwell time. Individual line scans were averaged 128 times for the final image. The chamber pressure during the imaging was $1\cdot10^{-7}$ mbar, working distance was 9 mm and the pixel resolution of the images were 2048x2048 pixels. The collected images were adjusted for black and white points to fill the full range of 256 grey levels.

2.3 Preparation of SERS-active Ag nano-aggregates

The aggregated silver nanoparticles were synthesized for generating enhanced Raman signals using previously reported methods [30]. All glassware was cleaned in aqua regia (3 parts HCl, 1 part HNO₃), rinsed with Nano-pure H₂O (SG Water USA, LLC), and then oven dried prior to use. In a typical procedure, an aqueous solution of AgNO₃ (1 mM, 200 mL) was brought to reflux while vigorously stirring, and then 4 mL of a 38.8 mM trisodium citrate solution was quickly added, which resulted in a change from colorless to grayish yellow, indicating that the reaction was completed. After the color change, the solution was refluxed for an additional 60 min, allowed to cool down to room temperature and subsequently stored at 4°C. The average diameter of the synthesized mostly spherical AgNPs was

measured using TEM micrographs and its value was found to be 100 ± 5 nm. After six months of storage, the solution contained an aggregated mixture of spherical, wire and rod-shaped nanoparticles as observed from the obtained TEM images (see example in Figure 1). For a time course study of stability, a small amount of AgNP stock solution was diluted with water and the UV-Vis absorbance measurement recorded every two weeks. A typical UV-VIS spectrum is shown in Figure 1.

2.4 Preparation of yeast samples

Saccharomyces cerevisiae (Roberts Ltd. #229264) was grown in Yeast Malt Broth (YM-Broth) (5 g/l peptone, 3 g/l yeast extract, 3 g/l malt extract, 10 g/l dextrose, pH 6.2; Sigma Y3757) at 30°C for ~20 hours with constant shaking (~150 rpm). The cultures were then centrifuged at 5000 g for 20 min at 4 °C, after which the supernatant was discarded and the pellet was suspended in 100 mL of gold dH₂O. This was still repeated to remove the residual growth media, after which cell pellets were resuspended in 50 mL of dH₂O and transferred into smaller containers. To prepare the cellular lysate, the cells resuspended in dH₂O were treated three times with an Emulsiflex C3 homogenizator (AVESTIN Europe GmbH). All the samples were stored in 0.5 mL microcentrifuge tubes in -80°C until the analysis.

Dekkera bruxellensis wild yeast strain was purchased from VTT Technical Research Centre of Finland Ltd. (VTT C-00348). It was cultivated in YM-Broth for ~20 h at +27 °C with constant shaking (~150 rpm). The culture was then treated as described above for *S. cerevisiae*.

Wickerhamomyces anomalus (former name *Pichia anomala*) was purchased from VTT Technical Research Centre of Finland Ltd. (VTT C-02470). It was cultivated in YM-Broth for ~20 h at +25 °C with constant shaking (~150 rpm). The culture was then treated as described above for *S. cerevisiae*.

All the yeast cultures are originating from individual colonies picked from agar plate.

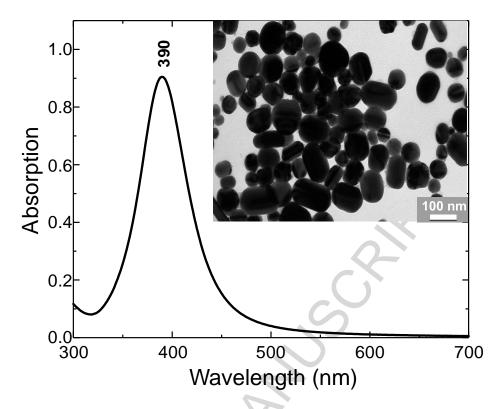


Figure 1. UV-Vis spectrum of aggregated silver nanoparticles. The spectrum shows a clear absorption band at 390 nm. The inset shows a high magnification transmission electron microscope image of a variety of structural features, i.e. hot-spots with plasmonic gap mode forming at the junctions of two or more Ag nanoparticles, that may be responsible for signal enhancement.

3. Results and Discussion

3.1 SERS spectra of yeasts (lysates)

Typical SERS spectra of the lysates from three yeast species, *Saccharomyces cerevisiae*, *Dekkera bruxellensis*, and *Wickerhamomyces anomalus* (former name *Pichia anomala*), mixed with aggregated silver nanoparticles are shown in Figure 2a in the range between 400 to 1800 cm⁻¹. The recorded spectra were highly reproducible as demonstrated by the examples of the non-treated data in Supporting Material figures S2-S4. The full list of peak positions along with the appropriate assignments of the three microbes is presented in Table 1.

The SERS spectra of the microbes show complex features dominated by the functional groups of proteins, lipids, and purines. However, in the spectral window between 500-1000 cm⁻¹ (fingerprint area), we noticed that the SERS peak pattern of *Wickerhamomyces anomalus* is clearly different compared with those from *Saccharomyces cerevisiae* and *Dekkera bruxellensis* spectra, as shown in Figure 2a. There is a great variability in both the band shapes and relative intensities of most of the modes and even a new peak

is observed around 790 cm⁻¹. This can be attributed to the fact that the adsorption behavior of *Wickerhamomyces anomalus* on Ag nanoaggregate is different than those of *Saccharomyces cerevisiae* and *Dekkera bruxellensis* species due to its morphological features (its structural features are found to be different from various groups of fungus) [31]. For this reason, we will discuss briefly the two SERS spectra in more detail in the next paragraph.

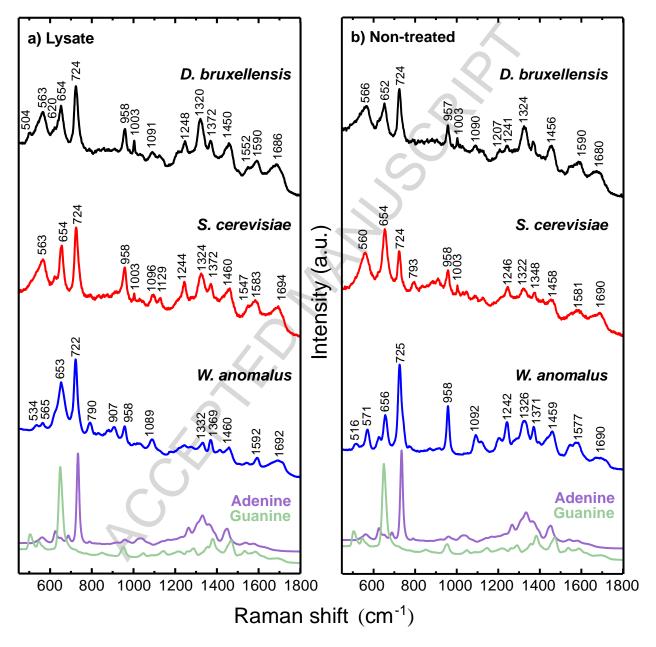


Figure 2. SERS spectra of (a) lysed and (b) non-treated yeast cells: *Dekkera bruxellensis*, *Saccharomyces cerevisiae*, and *Wickerhamomyces anomalus*. For comparison, there are also SERS spectra of Adenine and Guanine measured from aqueous solution (1 mM) mixed with Ag

nanoparticles. Under the experimental conditions described in the methods, SERS spectra were acquired using 785nm a near-infrared diode laser for excitation. The laser power at the sample was 1 to 10mW, acquisition time=1 to 5sec and the laser beam passes through 50 and 100x objective and focuses on the sample. All spectra were baseline corrected by the rubber band algorithm and intensity-normalized.

The SERS spectra of the three microbes exhibit a number of marker modes spreading well over the region between 500-1700 cm⁻¹. However, the three microbes gave similar spectral features, except the obvious minor differences in intensities, positions, and peak sharpness. The major feature (the dominating mode) in all the spectra is the pronounced intense band centered at 724 cm⁻¹ which is assigned to the ring breathing mode of adenine (A) base with some degree of mixing with the other bases as well as purines in general (Figure 2a & Table 1). The ring breathing mode of the guanine (G) and thymine (T) also appears as a weak band at 1374 cm⁻¹. However, these two modes appear to have slightly shifted, i.e., from 724 to 722 cm⁻¹ and 1372 to 1369 cm⁻¹ in the Wickerhamomyces anomalus spectrum. Such spectral shift can be attributed to the differences between the underlying molecular features of the species. The next intense peak occurs at 654 cm⁻¹, which is also related to the ring breathing mode in G and T. In addition, there is a weak shoulder peak observed at 620 cm⁻¹ which originate from the C-C twisted mode of phenylalanine (Figure 2a, Table 1). However, the later band is not well resolved in all three spectra but discernible. It is also worth noting, however, that the SERS spectra obtained from the molecular layers coating the metal surface and the relative intensity of the molecular components are modulated by the orientation of the adsorbate on the surface of the aggregated metal. This phenomenon is manifested over the entire SERS spectra. Moskovits described the selection rule of molecules absorbed (orientation) onto metal surfaces in his famous publication [32]. For the sake of comparison, the powder spectra of A and G are shown in Figure 2.

Table 1. SERS mode assignments of *Saccharomyces cerevisiae*, *Dekkera bruxellensis* and *Wickerhamomyces anomalus* in the lysate and non-treated forms.

Saccharomyces cerevisiae (cm ⁻¹)		Dekkera bruxellensis (cm ⁻¹)		Wickerhamomyces anomalus (cm ⁻¹)		Tentative assignments *
lysate	non- treated	lysate	non- treated	lysate	non- treated	
508	516	504			515	v(S-S) disulfide (amino acid cysteine)
563	559	564	563	565	571	T/C, G

602 621 626 627 626 627 626 627 C-C twisting mode of Phe 654 654 652 653 656 G, T (ring-breathing modes) 724 723 724 724 722 725 Adenine (ring breathing modes of putines) 798 794 791 v(O-P-O) RNA 815 v(O-P-O) RNA 831 834 831 v _s (O-P-O) str., T 856 859 Tyrosine 913 883 882 Phosphodiester, Deoxyribose 920 909 907 Skeletal C-C, α-helix 957 959 959 957 958 957 v(C-O) and (C-N) of proteins 1003 1003 1003 1003 1003 1003 Phe 1017 1017 1017 δ(C-H) of T 1038 1030 1027 1028 1023 1026 δ(CH), Phe (protein assignment) 1127 1126 1124 1172 δ(C-H) of tyrosine and proteins 1203 1204 1211 1205 1198 1205 Amide III 1244 1245 1245 1243 1242 1241 Amide III, v _m (PO ₂) str. 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1458 1452 1461 1456 1458 1460 Nucleic acid modes 1550 1551 1548 Amide II							
Total Tot	602					616	C-C twisting mode of Phe
724	627	621	626	627	626	627	C-C twisting mode of Phe
198 794 791 791 791 792 794 791 795 794 795	654	654	652		653	656	G, T (ring-breathing modes)
815	724	723	724	724	722	725	(ring breathing modes of
831 834 831 V ₄ (O-P-O) str., T 856	798	794			791		ν(O-P-O) RNA
856					815		ν(O-P-O) RNA
874 870 Polysaccharides	831	834			831		v_s (O-P-O) str., T
913 883 882 Phosphodiester, Deoxyribose 920 909 907 Skeletal C-C, α-helix 957 959 959 957 958 957 ν(C-O) and (C-N) of proteins 1003 1003 1003 1003 1003 Phe 1038 1030 1027 1028 1023 1026 δ(C-H), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, v _{as} (PO₂) str. 1239 1234 Amide III, v _{as} (PO₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1458 1452 1461 1456 1458 1460 <t< td=""><td>856</td><td></td><td></td><td></td><td>859</td><td></td><td>Tyrosine</td></t<>	856				859		Tyrosine
920 909 959 957 958 957 v(C-O) and (C-N) of proteins 957 959 959 957 958 957 v(C-O) and (C-N) of proteins 1003 1003 1003 1003 1003 Phe 1017 1017 δ(C-H) of T 1038 1030 1027 1028 1023 1026 δ(C-H), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, vas(PO₂) str. 1239 1234 Amide III, vas(PO₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1458 1452 1461 1456 1458					874	870	Polysaccharides
957 959 959 957 958 957 ν(C-O) and (C-N) of proteins 1003 1003 1003 1003 1003 Phe 1017 1017 1017 δ(C-H) of T 1038 1030 1027 1028 1023 1026 δ(CH), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, ν _{as} (PO₂) str. 1239 1234 Amide III Amide III, ν _{as} (PO₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1458 1452 1461 1456 1458 1460 Nucleic acid modes	913	883			882		Phosphodiester, Deoxyribose
981 980 981 v(C-C) of proteins	920	909			907		Skeletal C-C, α-helix
1003 1003 1003 1003 1003 Phe 1017 1017 δ(C-H) of T 1038 1030 1027 1028 1023 1026 δ(CH), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, vas(PO₂) str. 1203 1204 1211 1205 1198 1205 Amide III, vas(PO₂) str. 1244 1245 1243 1242 1241 Amide III, vas(PO₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) CH₂ sciss (lipids), G,A 1458 1452 1461 1456 1458	957	959	959	957	958	957	ν(C-O) and (C-N) of proteins
1017 1017 1018 1028 1023 1026 δ(CH) of T 1038 1030 1027 1028 1023 1026 δ(CH), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1172 1172 δ(C-H) of tyrosine and proteins 1203 1204 1211 1205 1198 1205 Amide III, ν _{as} (PO₂) str. 1239 1234 Amide III 1244 1245 1245 1243 1242 1241 Amide III, ν _{as} (PO₂) str. 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes			981	980	981		v(C-C) of proteins
1038 1030 1027 1028 1023 1026 δ(CH), Phe (protein assignment) 1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, vas(PO₂) str. 1239 1234 Amide III Amide III Vas(PO₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) CH₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1003	1003	1003	1003	1003	1003	Phe
1092 1089 1091 1092 1089 1092 PO₂ in nucleic acid 1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1172 1172 δ(C-H) of tyrosine and proteins 1203 1204 1211 1205 1198 1205 Amide III, vas(PO₂) str. 1239 1234 Amide III 1244 1245 1245 1243 1242 1241 Amide III, vas(PO₂) str. 1324 1322 1325 1324 1332 1326 G, CH₃CH₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes				1017	1017		δ(C-H) of T
1127 1126 1124 1119 δ(C-H), Phe (protein assignment) 1203 1204 1211 1205 1198 1205 Amide III, ν _{as} (PO ₂) str. 1239 1234 Amide III, ν _{as} (PO ₂) str. 1244 1245 1245 1243 1242 1241 Amide III, ν _{as} (PO ₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1038	1030	1027	1028	1023	1026	. , , , , , , , , , , , , , , , , , , ,
assignment) 1203 1204 1211 1205 1198 1205 Amide III, ν _{as} (PO ₂) str. 1239 1234 Amide III 1244 1245 1245 1243 1242 1241 Amide III, ν _{as} (PO ₂) str. 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1092	1089	1091	1092	1089	1092	PO ₂ in nucleic acid
1203 1204 1211 1205 1198 1205 Amide III, v _{as} (PO ₂ ') str. 1239 1234 Amide III 1244 1245 1245 1243 1242 1241 Amide III, v _{as} (PO ₂ ') str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1127	1126	1124			1119	`
1239 1234 Amide III 1244 1245 1245 1243 1242 1241 Amide III, v _{as} (PO ₂) str. 1289 C 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes				1172	1172		δ (C-H) of tyrosine and proteins
1244 1245 1243 1242 1241 Amide III, ν _{as} (PO ₂) str. 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1203	1204	1211	1205	1198	1205	Amide III, $v_{as}(PO_2^-)$ str.
1289 C 1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes				1239	1234		Amide III
1324 1322 1325 1324 1332 1326 G, CH ₃ CH ₂ deformation 1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1244	1245	1245	1243	1242	1241	Amide III, $v_{as}(PO_2^-)$ str.
1372 1374 1372 1368 1370 1371 T,A,G (ring breathing modes of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes),			1289	С
1395 of purines and pyrimidines) 1395 ρ(CH) 1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1324	1322	1325	1324	1332	1326	G, CH ₃ CH ₂ deformation
1414 CH ₂ sciss (lipids), G,A 1458 1452 1461 1456 1458 1460 Nucleic acid modes	1372	1374	1372	1368	1370	1371	
1458 1452 1461 1456 1458 1460 Nucleic acid modes						1395	ρ(СН)
					1414		CH ₂ sciss (lipids), G,A
1550 1551 1548 Amide II	1458	1452	1461	1456	1458	1460	Nucleic acid modes
	1550		1551		1548		Amide II

1583	1581	1589	1589	1576	1579	C=C bending mode of phenylalanine
1693	1686	1687	1672	1692	1686	Amide I of protein

^{*}All the assignments are from references 12, 16, 18 & 31

Furthermore, a broad peak was observed at 563 cm⁻¹. The characteristic broad and weak bands observed at 563 and 534 cm⁻¹ can be assigned to the T/cytosine (C) and G with some degree of mixing from the C-C twisting mode of Phe (Table 1 and Figure 2a). Interestingly, the peak at 536 cm⁻¹ is extremely weak in *Wickerhamomyces anomalus* cell sample compared with those from *Saccharomyces cerevisiae* and *Dekkera bruxellensis* species, as shown in Figure 2a. Apart from these characteristic peaks, the last one observed at 504 cm⁻¹ is associated with the disulfide (S-S) modes of the cysteine residue in a protein. This mode is clearly indicating that the proteins in the two species contain a plenty of disulfide bonds. However, the absence of this mode in *W. anomalus* species may be ascribed with the low concentration of proteins with disulfide bonds.

The third strongest signal in the SERS spectra of S. cerevisiae and D. bruxellensis (Figure 2a) appears at 958 cm⁻¹ assigned to the stretching mode (C-O), admixed with v(C-N) of proteins. However, this mode is relatively weak and broad in W. anomalus species compared with the two spectra. The different relative intensities of the peaks observed may give some indication of the relative abundances of each component present in each species. Besides this band, in W. anomalus we also observe a peak at 790 cm⁻¹ which is largely absent in the other two species. In addition to this mode, there are two additional peaks at 907 cm⁻¹ attributed to the C-O-C skeletal mode and the weak shoulder at 875 cm⁻¹ which is assigned to v(C-C), respectively. These species-specific modes could serve as a marker to differentiate species of micro-organism owing to the structural uniqueness of the yeast species. Furthermore, it can be seen in Figure 2a that the region between 910 to 1700 cm⁻¹ contains a complex shape of bands (low intensities) and suffer from a lack of high signal-to-noise ratios indicating the presence of several overlapping peaks. This is due to the fact that the cell contains several macromolecules and these constituents co-adsorb on the metal surface, resulting in peak broadening and distortion. In addition, the relative intensity of the molecular components is modulated by the orientation of the adsorbate on the hot spots formed between closely spaced Ag nanoparticles due to a high-intensity plasmonic gap mode. These spectral regions will be analyzed in detail in order to discriminate the three species based on their spectral characterstics.

 $^{^{}e}$ υ, stretching; δ, bending; ρ, rocking; τ, torsion; wagg, wagging; twist, twisting; sciss, scissoring; def, deformation; as, asymmetric; s, symmetric; C, cytosine; T, thymine; A, adenine; G, guanine; Phe, phenylalanine; Tyr, tyrosine

The next medium-intense band is observed at 1324 cm⁻¹ which correspond to guanine and CH₃CH₂ deformation. Another weak band appearing at 1372 cm⁻¹ can be assignable to ring breathing mode of thymine, adenine and guanine (Table 1). These two bands are too weak and broad to be observed in the *W. anomalus* spectrum, indicating the presence of an overlapping band. The C-C skeletal stretching vibration of phenylalanine ring together with C-H in-plane phenylalanine yields the bands at 1003 and 1030 cm⁻¹. However, this band (1003 cm⁻¹) is diminishing in intensity and shifts to 978 cm⁻¹ in *W. anomalus* sample, indicating that the matrix affects the phenylalanine environment. The intensity of the band at 1030 cm⁻¹ also decrease significantly in *W. anomalus* cell. The complex symmetrical stretching mode of PO₂⁻¹ is composed of C-N, C-O, and v(C-C) skeletal of acyl backbone in lipid observed at 1096 and 1129 cm⁻¹ in *S. cerevisiae* and *D. bruxellensis* samples and show a minor spectral shift to 1089 and 1126 cm⁻¹ in *W. anomalus* spectrum.

The amide I, II and III vibrational modes are conformational sensitive and exhibits extremely complex features involving the coupling of other vibrational modes. The amide group involves primarily C=O stretching, in-plane modes of N-H bending, and v (C-C) vibrational modes. In the region between 1100 to 1700 cm⁻¹, very strong amide III mode for α-sheet disordered conformation is observed at 1245 cm⁻¹ for *S. cerevisiae*, at 1248 cm⁻¹ for DB and very weak band for *W. anomalus* which lies at 1241 cm⁻¹. The presence of a shoulder at 1220 cm⁻¹ is also due to amide III and arising from the coupling of C-N stretching and N-H bonding. The amide I SERS band is observed at 1694 (SC), 1686 (*D. bruxellensis*) and 1692 cm⁻¹ in *W. anomalus* species. This band is primarily arising due to C=O stretching mode coupled with the in-plane N-H bending and C-N stretching bands. The high wavenumber observed may be associated with the type of the secondary structure dominating in these species. Thus, the observation of the high wavenumber peaks provides evidence for the existence of β-pleated sheet configuration in all the cells. [33] This result is further confirmed by the observation of the amide III bands at 1245 cm⁻¹.

3.2 SERS spectra of yeasts (non-treated cells)

The vibrational spectra of the *S. cerevisiae*, *D. bruxellensis*, and *W. anomalus* in non-treated form are dominated by proteins, lipids, and purine groups. Figure 2b shows the SERS spectra collected from each cell of the three species absorbed on the aggregated silver particles (Figure 3). As seen in Figure 2b, significant spectral shifts and relative intensity changes are observed when compared with the cells in the lysate form (Figure 2a). We will discuss this rather clear difference between the two forms in detail in the following section. Moreover, the high-resolution HIM images are used to analyze the formation of the aggregates around the fungi cells (Figure 3).

In the region between 500 to 1200 cm⁻¹ shown in Figure 2b, the SERS spectra taken from the non-treated cells is characterized by medium/high-intensity modes observed at 566, 652, 724, 957, 1003 and

1092 cm⁻¹ (as already assigned in Figure 2a) attributed to disulfide bond, adenine ring breathing, vs (CH₃) of protein (α-helix), C-C skeletal of phenylalanine (Phe) and symmetric stretching mode of PO₂⁻ group (Table 1). These bands show a change in peak intensities and a minor spectral change. The peak intensities at 724 and 958 cm⁻¹ increase without a shift in *W. anomalus*, and simultaneously the peak at 654 cm⁻¹ is intensified in *S. cerevisiae* species compared with the other two species, which indicated the adsorption of the purines on the hot-spot. Similarly, the broad peak at 566 cm⁻¹ shifts to 560 cm⁻¹ in *S. cerevisiae*, while it exhibits sharp peak and shifts to 571 cm⁻¹ in *W. anomalus* cell. In addition, a peak at 516 cm⁻¹ was observed in the same spectrum (*W. anomalus*). This mode is assigned to S-S stretching mode within the protein. Lastly, the last two bands observed at 1003 and 1092 cm⁻¹ assigned to C-H in-plane phenylalanine (proteins) and symmetric (PO₂⁻) are very weak for the two species (*D. bruxellensis* and *S. cerevisiae*). The former band is extremely weak in *W. anomalus*, while the 1092 cm⁻¹ mode increases in intensity. However, a strong mix with 1116 cm⁻¹ mode also observed. A similar observation was made in the SERS spectra of the lysate form in *W. anomalus* cells. The selected SERS assignments are presented in Table 1.

The relatively high amount of the purine (adenine, guanine) signals in the SERS spectra of the non-treated cells is a bit surprising, as one would expect more cell wall associated signals arising from the intact cells. However, previous studies revealed that these peaks are commonly detected in SERS experiments carried out for intact yeast and bacterial cells [22, 34-36]. It is possible, that those signals are associated with degradation of nucleic acids and nucleotides due to a cell starvation, which leads into release of the adenine and guanine outside the cells, as described recently for starved bacterial cells by Premasiri *et al.* [35]. Also, traces of nucleic acids could be released from cells broken during the preparation, but we did not observe significant DNA release in the sample preparation. In addition, it is also possible that some of the AgNPs have been able to penetrate into cells and thus yield SERS signal from the purines inside the cells. These particles would not be visible in the HIM images either, since HIM is very surface sensitive and it does not allow detection of the nanoparticles through the cell wall.

The complex shape of the modes between 1200 to 1700 cm⁻¹ is due to the presence of several overlapping modes in phosphate and amide groups of proteins and the ring of purines DNA/RNA. We noticed that the amide III band at 1241 cm⁻¹ is very weak in both *D. bruxellensis* and *S. cerevisiae* spectra, whereas it exhibits a medium and sharp peak at 1242 cm⁻¹ in *W. anomalus* cell. This may indicate a different orientation of the amide II component on the surface of the Ag-aggregates, which is consistent with the fact that the *W. anomalus* species is very different compared with that *D. bruxellensis* and *S. cerevisiae*. In addition, a weak shoulder appears at 1207 cm⁻¹ and its intensity become lower in *S. cerevisiae*, while increases in *W. anomalus* spectrum.

In the amide I region (1590 and 1680 cm⁻¹), the modes observed at 1456, 1590 and 1680 cm⁻¹ are broader (much weaker) and the spectra suffer from a low signal/noise ratio, indicating a very complex spectral pattern due to, the most probable, substancial amount of spectral mixing. For example, the 1680 cm⁻¹ (*D. bruxellensis*) shifts to 1690 cm⁻¹ (*S. cerevisiae* and *W. anomalus*), thus providing evidence to some overlapping with other modes. We also identify a weaker shoulder band around 1547 cm⁻¹ which is possibly due to amide II as suggested in the literature [20]; this band indicates the presence of parallel β-sheet structure in all three species. Besides the weak amide II band, another band is observed at 1590 cm⁻¹, which is considered to be due to C=C bending of Phy and a minor mixing from amide I. This mode is strongly shifted to lower wavenumber 1581 cm⁻¹ (*S. cerevisiae*) and 1577 cm⁻¹ (*W. anomalus*), which indicated strong mixing with other modes.

3.3 Statistical Analysis

The closeness and variability in the SERS spectra of the three yeasts samples can be attributed to the sample matrix, variability in yeasts composition and the effect of Ag NP aggregation that the individual biomolecules oriented within the hot spots. This is assessed using statistical analysis where standard deviation (SD) is calculated on the respective SERS bands appears in all three samples in respect to their spectral shift. It can be seen in Table 1S and 2S that the SD of all the SERS bands in both states deviates from the mean value except for the band at 1003cm^{-1} . For instance, Amide I &III bands (SD =3.2, 6.6), CH₃CH₂ (def) (4.4) and C=C (6.5) bending modes of phenylalanine in lysate state are significantly deviate from their mean values and suggesting that the individual modes are different from each other. Similar trends are also observed with the non-treated species such as the band for v (S-S) (6.1), T/C, G (3.5), T,A,G (ring-breathing modes of purines and pyrimidines) (3.0), C=C bending mode of phenylalanine (5.3) and Amide I of protein (8.1) show significant matrix and compositional dependency between the three species.

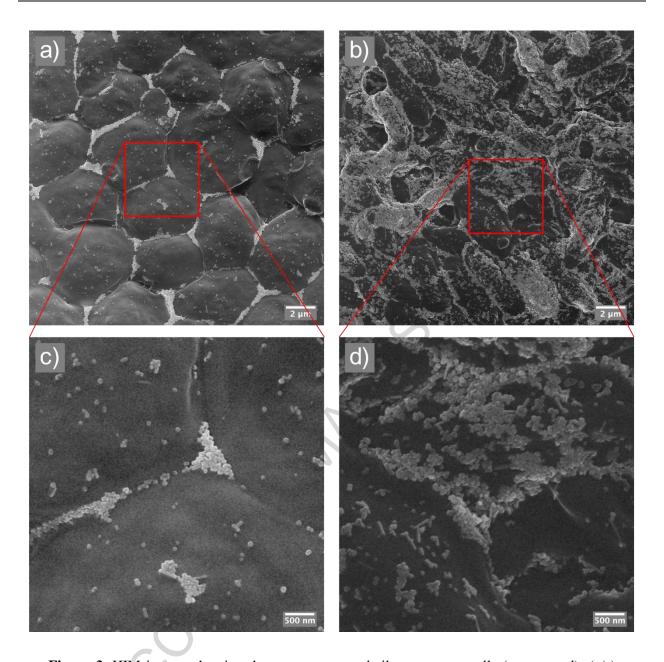


Figure 3. HIM images showing the nano-aggregated silver on yeast cells (non-treated): (a/c) *Saccharomyces cerevisiae* (b/d) *Dekkera bruxellensis*. The image size of (a) and (b) is $20x20 \mu m^2$, and (c) and (d) is $5x5 \mu m^2$.

3.4 HIM imaging of the yeasts

High resolution HIM images of the yeast cells after incubated with the nano-aggregated Ag particles are shown in Figure 3. The images of different yeast species are well resolved allowing the identification of single cells, cell morphology and the way the nanoparticles are distributed around the cells. Each cell type shows a considerable different morphology and selective attachment of the nanoparticles, which is manifested in their SERS spectra. The surface of *D. bruxellensis* is entrapped in the aggregate (complete coverage), whereas the *S. cerevisiae* is only partially covered by nanoparticles which suggests that the

interaction between the cell surface with nanoparticles is mostly governed by the morphology and the adherent behavior of the cells (Figure 3). Despite that the SERS spectra of all the cells show a similar signal-to-noise ratio which is sufficient to detect the very weak modes associated with the microcomponents of the cell. This suggests that the NP strongly adsorb onto the surface of the cell or penetrate into the inner cell structure possibly taken up by the cell through endocytosis [37], leading to the agglomeration of the nanoparticles inside the cell. Such phenomenon would lead to the synergetic effect of the inside and outside aggregation of nanoparticles which may form many hot-spots created by Ag-Ag junction where the cell components could be confined and exposed to extremely high local electromagentic field and therefore generate high SERS signal.

Finally, it is worth noting that the SERS spectra of the non-treated cells suffer from the low spectral resolution in all species compared with the spectra collected from lysed cells. This can be rationalized by the constrains of the cell components by the cell wall, resulting in that the macromolecules do not come in a direct contact with the hot-spots of the aggregated Ag particles where the enhancement of scattering occurs. Figure 3 provides some insights into this question and reveals the size mismatch between yeast cells and silver nanoparticles.

3.5 SERS detection of yeasts (non-treated cells) from beer

We also investigated the SERS spectra of the three yeast species suspended in beer and thus demonstrated the potential and the high sensitivity of this technique in a complex matrix. Figure 4 shows the SERS spectra of the three types of cells incubated with the silver nanoparticles and then mixed into beer. The observed SERS spectra show a good signal-to-noise ratio compared with those observed for Ag solution without the beer. Similar to the pure SERS spectra, the SERS spectra of the cells in beer is characterized by proteins, lipids, purines and show a similar level of enhancement as those shown in Figure 3. We nevertheless noticed that the SERS spectra of D. bruxellensis suffer from a low signal-tonoise ratio compared to the other two SERS spectra (Figure 4). Besides a clear indication of the spectral noise, some bands show an increase/decrease a relative intensity and a spectral shift is observed in peak positions as compared with the pure cells. For instance, the relative intensities of the two bands at 654 and 728 cm⁻¹ were switched and showed a minor spectral shift (2 to 4 cm⁻¹). This behavior is probably attributed to the surface properties of the cell suspended in distilled water vs. cell suspended in beer, which provides chemically less demanding condition for the yeast cell. This may lead to changes in accessibility of cellular components for the nanoparticles. Further, the intensity of the previously assigned band centered at the 1241 cm⁻¹ increased and shifted to higher energy mode (1250 cm⁻¹), which means that more of the (PO₂) moiety may be trapped in the hot-spot in beer-suspended samples. However, the

remaining modes associated principally with nucleic acid, amide I and phenylalanine show no significant spectral changes.

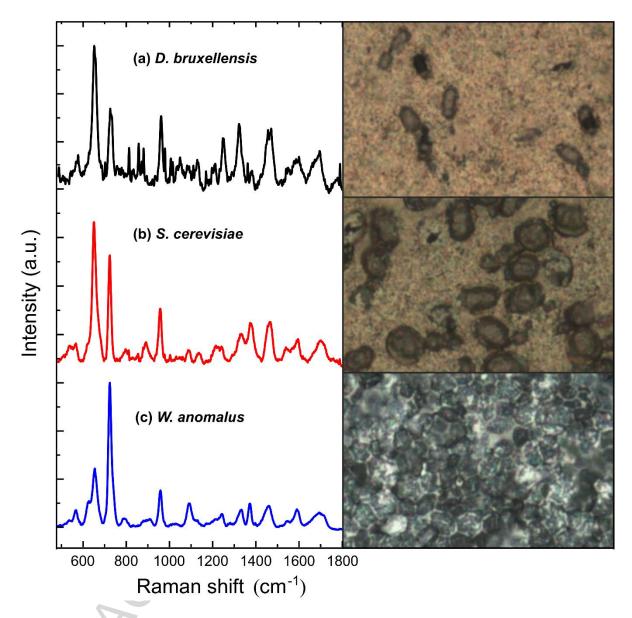


Figure 4. SERS spectra of (a) *D. bruxellensis*, (b) *S. cerevisiae* and (c) *W. anomalus* in beer. The corresponding white light images collected using a microscope are shown. Experimental conditions: Laser power = 1 to 10mw, acquisition time= 1 to 5sec and the laser beam passes through 50 and 100x objective and focuses on the sample.

As with the SERS spectra of *S. cerevisiae*, no significant spectral changes (all modes stay within 1 to 3 cm⁻¹) were observed, except in the 1200-1500 cm⁻¹ spectral region, which revealed increased intensity. As already mentioned above, the enhancement of the SERS intensities depends on the orientation of these modes in the junction. Apart from that, the remaining modes are identical with those of the pure SERS

spectra. Lastly, we have not observed any difference (a major spectral shift or intensities) between the SERS spectra of *W. anomalus* suspended in water or in beer.

To explore the robustness of the technique, we further performed experiments by diluting the three yeast cells in the beer matrix in a higher dilution of 1:40. The observed spectra of the three species still clearly show all the important features of the cells similar to the one seen in Fig. 2 & 4., though the spectra suffer from a low signal-to-ratios, as expected taken into consideration the dilution factor. The spectra are shown in Supporting Material Figure S1. As a control experiment we also mixed bare AgNPs with beer. Expectedly, the obtained spectra showed no features of the yeasts, as depicted in the Supporting Material Figure S5. Thus the result reaffirms the robustness of the SERS techniqe by being able to discern detection and identification of species regardless of a high background due to the complex chemical environment.

4. Conclusions

We used highly SERS-active aggregated silver nanoparticles substrate to discriminate between the three yeast species as lysates and in non-treated cell samples. The SERS spectra of *S. cerevisiae* and *D. bruxellensis* were almost identical except a slight spectral shift and change in relative intensities of the phosphate group in DNA/RNA and amide III, implying similar molecular compositions present in these microbes. However, some part of the observed differences in spectra and band intensities may also be attributable to the differences in the accessibility of the chemical structures of the yeasts and in the characteristics of the interactions between yeast molecules and the metal particle surface.

Importantly, the spectral pattern of the lysate and non-treated cell sample show remarkable differences. While cellular components are well accessible in lysate cell, the constricted cell morphology of the non-treated cells leads to relatively weak SERS spectra. This may be due to steric constrains and a macromolecular crowding, preventing access to the hot-spots within the metal particle aggregates (see Figure 3). Additionally, we observed significant spectral differences in the spectrum of *Wickerhamomyces anomalus* as compared with *Saccharomyces cerevisiae* and *Dekkera bruxellensis*, reflecting the obvious molecular composition diversity within the yeast species. Beside a clear indication of the decrease/increase of relative intensities, most of the SERS modes in *W. anomalus* show a shift in peak position to higher/lower wavenumber compared with the *S. cerevisiae* and *D. bruxellensis* spectra, especially in 1200-1700 cm⁻¹ region. Moreover, it is worth noting that all the three amide groups (I, II & III) were detected in the SERS spectra present in the species and the broader peaks with high

wavenumbers 1680 cm^{-1} (*D. bruxellensis*), 1690 cm^{-1} (*S. cerevisiae*) and 1690 cm^{-1} (*W. anomalus*) indicate the presence of parallel β -sheet structure accessible on the surface of non-treated cells. We also collected the SERS spectra of the same cells in beer medium and analyzed the spectral shift and intensity changes by comparing to spectra obtained in water-suspended cells. The results indicate the extreme sensitivity of the SERS technique that is capable of simultaneously detecting and distinguishing yeast species in a complex media with reasonably high signal-to-noise ratio and spectral reproducibility.

In conclusion, SERS provides a unique platform to detect specific biochemical fingerprints with high sensitivity and specificity at high spatial resolution, which makes the technique suitable for complex mixtures of microbial cultures, and opens possibilities for the development of robust biosensors for monitoring food quality and safety.

Funding: This research was funded by Tekes (project #1185/31/2013). Academy of Finland is acknowledged for funding for VPH (136288) and JJT (263526). This work also supported by grants from FAPESP (2016/06424-5, 2013/14262-7).

Acknowledgments: Support from BioNavis, Fimlab Laboratories, PBL Brewing Laboratory (VTT) and Orion Diagnostica are gratefully acknowledged. Ulla Kiiskinen (U. Tampere) is acknowledged for technical support in sample preparation. Biocenter Finland is acknowledged for infrastructure support.

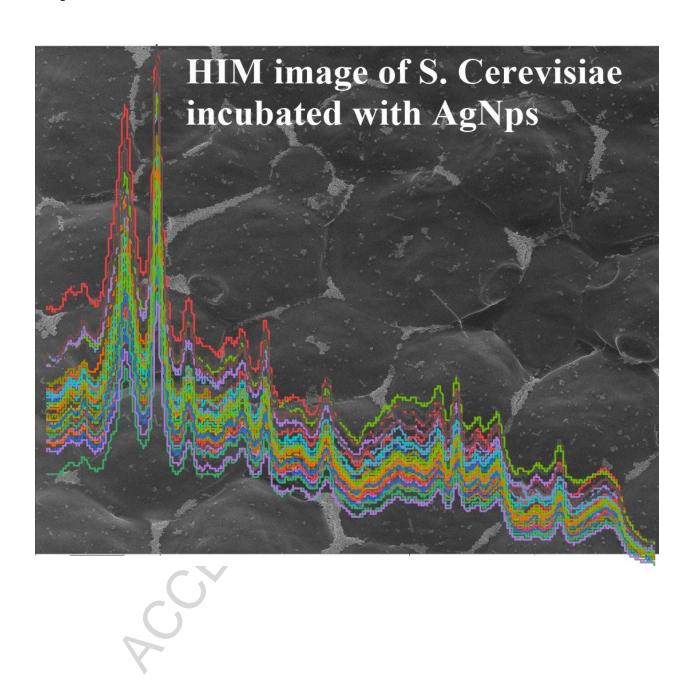
Conflicts of Interest: The authors declare no conflict of interest.

References

- [1] E. Nevoigt, Progress in metabolic engineering of Saccharomyces cerevisiae, Microbiol Mol Biol Rev, 72 (2008) 379-412.
- [2] L.V. McFarland, P. Bernasconi, Saccharomyces boulardii. A Review of an Innovative Biotherapeutic Agent, Microbial Ecology in Health and Disease 6(1993) 157-171.
- [3] G. Naja, P. Bouvrette, S. Hrapovic, J.H.T. Luong, Raman-based detection of bacteria using silver nanoparticles conjugated with antibodies, Analyst 132 (2007) 679-686.
- [4] Fleet.G.H., Cell walls., in: A.H.a.H. Rose, S. (Ed.) The Yeasts, Academic Press, London, London, 1991, pp. 199-277.
- [5] F.M. Klis, Cell wall assembly in yeast, Yeast, 10 (1994) 851-869.
- [6] P. Orlean, H. Ammer, M. Watzele, W. Tanner, Synthesis of an O-glycosylated cell surface protein induced in yeast by alpha factor, Proc Natl Acad Sci U S A., 17 (1986) 6263-6266.
- [7] H.W. Schrotter, H.W. Klockner, Raman scattering cross sections in gases and liquids, in: A. Weber (Ed.) Raman Spectroscopy of Gases and Liquids, Springer-Verlag, Berlin, 1979, pp. 123-166.
- [8] S. Bhavya, R.F. Renee, H. Anne-Isabelle, R. Emilie, P.V.D. Richard SERS: Materials, applications, and the future, Materials, 15 (2012) 16-25.
- [9] T. Zhong-Qun, Y. Zhi-Lin, R. Bin, W. De-Yin, SERS From Transition Metals and Excited by Ultraviolet Light, in: K. Katrin, K. Harald, M. Martin (Eds.) Surface-Enhanced Raman Scattering Physics and Applications, Springer-Verlag Berlin Heidelberg Germany, 2006, pp. 125-147.
- [10] M. Moskovits, Surface-enhanced spectroscopy, Rev. Mod. Phys, 57 (1985) 783-826.
- [11] G.C. Schatz, Theoretical-studies of surface enhanced Raman-scattering, Acc. Chem. Res., 17 (1984) 370-376.
- [12] M. Kerker, Electromagnetic model for surface-enhanced Raman-scattering (SERS) on metal colloids, Acc. Chem. Res., 17 (1984) 271-277.
- [13] A. Otto, I. Mrozek, H. Grabhorn, W. and Akemann, Surface-enhanced Raman-scattering, J. Phys.-Condens. Matter, 4 (1992) 1143-1212.
- [14] P. Kambhampati, C.M. Child, M.C. Foster, A. Campion, On the chemical mechanism of surface enhanced Raman scattering: experiment and theory, J. Chem. Phys., 108 (1998) 5013.
- [15] A. Campion, P. Kambhampati, Surface-Enhanced Raman Scattering Chem. Soc. Rev. , 27 (1998) 241–250.
- [16] K. Katrin, W. Yang, Harald Kneipp, L.T. Perelman., I. Irving, R.D. Ramachandra, S.F. Michael, Single Molecule Detection Using Surface-Enhanced Raman Scattering (SERS), Phys. Rev. Lett., 78 (1997) 1667.
- [17] C.J. Constantino, T. Lemma, P.A. Antunes, R. Aroca, Single-molecule detection using surface-enhanced resonance Raman scattering and Langmuir-Blodgett monolayers., Anal Chem., 73 (2001) 3674-3678.
- [18] K. Katrin, K. Harald, D. Geurt, I. Irving, R.D. Ramachandra, S.F. Michael, Single-Molecule Detection of a Cyanine Dye in Silver Colloidal Solution Using Near-Infrared Surface-Enhanced Raman Scattering Applied Spectroscopy 52 (1998) 175-178.
- [19] A. Sujith, T. Itoh, H. Abe, K. Yoshida, M.S. Kiran, V. Biju, M. Ishikawa, Imaging the cell wall of living single yeast cells using surface-enhanced Raman spectroscopy, Anal Bioanal Chem., 394 (2009) 1803-1809.
- [20] L. Tibebe, S. Alex, H. Ville, P.H. Vesa, J.J. Toppari., SERS detection of cell surface and intracellular components of microorganisms using nano-aggregated Ag substrate, Vibrational Spectroscopy 83 (2016) 36–45.
- [21] D.I.R. José Maria, P.M. Martin-Sanchez., S.-C. Santiago, B. Hermosin., K. Heike, S.-J. Cesareo, Structure of melanins from the fungi Ochroconis lascauxensis and Ochroconis anomala contaminating rock art in the Lascaux Cave, Sci Rep, 7 (2017) 13441.
- [22] I. Sayin, M. Kahraman, F. Sahin, D. Yurdakul, M. Culha, Characterization of yeast species using surface-enhanced Raman scattering, Appl Spectrosc, 63 (2009) 1276-1282.

- [23] L. Boyan, M.S.S. Narayana, H.R. Bryan, G.R. Alan, Using surface-enhanced Raman scattering (SERS) and fluorescence spectroscopy for screening yeast extracts, a complex component of cell culture media, Journal of Raman Spectroscopy 43 (2012) 1074–1082
- [24] A. Sujith, T. Itoh, H. Abe, A.A. Anas, K. Yoshida, V. Biju, M. Ishikawa, Surface enhanced Raman scattering analyses of individual silver nanoaggregates on living single yeast cell wall, Applied Physics Letters, 92 (2008).
- [25] P. Marek, SERS Investigations of Cells, Viruses and Microorganisms, in: G. Elias (Ed.) Surface-Enhanced Raman Spectroscopy: Bioanalytical, Biomolecular and Medical Applications, Springer International Publishing Switzerland, 2016, pp. 127-148.
- [26] S.B. Rodriguez, M.A. Thornton., R.J. Thornton., Raman Spectroscopy and Chemometrics for Identification and Strain Discrimination of the Wine Spoilage Yeasts *Saccharomyces cerevisiae*, *Zygosaccharomyces bailii*, and *Brettanomyces bruxellensis*, Appl Environ Microbiol 79 (2013) 6264-6270.
- [27] C. Wei, M. Li, X. Zhao, Surface-Enhanced Raman Scattering (SERS) With Silver Nano Substrates Synthesized by Microwave for Rapid Detection of Foodborne Pathogens, Front Microbiol, 9 (2018) 2857.
- [28] W. Ruben, P. Márton, W. Michael, N. Reinhard, E. Martin, S. Michael, P.I. Natalia, Surface-enhanced Raman spectroscopy of microorganisms: limitations and applicability on the single-cell level Analyst, 144 (2019) 943-953.
- [29] M. Chisanga, H. Muhamadali, D.I. Ellis, R. Goodacre, Surface-Enhanced Raman Scattering (SERS) in Microbiology: Illumination and Enhancement of the Microbial World, Appl Spectrosc, 72 (2018) 987-1000
- [30] P.C. Lee, D. Meisel, Adsorption and surface-enhanced Raman scattering of dyes on silver and gold sols, J.Phys.Chem, 86 (1982) 3391-3395.
- [31] D. Tibor, Classification of Yeasts, CRC Press, Taylor & Francis Group, New York, 2008.
- [32] M. Moskovits, Surface selection rules J. Chem. Phys, 77 (1982) 4408.
- [33] E. Thaillandier, J. Liquer, A. Taboury, Advances in Infrared and Raman Spectroscopy, in: C. RJH, H. RE (Eds.), Wiley, New York, 1985, pp. 65-81.
- [34] P. W. Ranjith, G. Yoseph, Z. Lawrence D., On the difference between SERS spectra of cell growth media and whole bacterial cells, Appl Spectrosc, 65 (2011) 493–499.
- [35] P. W. Ranjith, L. Jean C., S.-B. Alexis, T. Roger, C. Catherine E., Z. Lawrence D., The biochemical origins of the surface enhanced Raman spectra of bacteria: metabolomics profiling by SERS, Anal Bioanal Chem, 408 (2016) 4631–4647.
- [36] W.R. Premasiri, D.T. Moir, L.D. Ziegler, Vibrational fingerprinting of bacterial pathogens by surface enhanced Raman scattering (SERS), The International Society for Optical Engineering, 5795 (2005) 19-29.
- [37] Z. Sulin, G. Huajian, B. Gang, Physical Principles of Nanoparticle Cellular Endocytosis, ACS Nano, 9 (2017) 8655-8671.

Graphical abstract



Identifying yeasts using surface enhanced Raman spectroscopy

Tibebe Lemma ^{1,*}, Jin Wang ², Kai Arstila ³, Vesa P. Hytönen ^{4,} and J. Jussi Toppari ³

- ¹ Faculdade de Clências e Tecnologia (FCT)-Universidade Estadual Paulista (UNESP)-Presidente Prudente, SP,19060-900, Brazil
- ² Institute of Intelligent Machines, Chinese Academy of Sciences, Hefei, Anhui 230031, P. R. China
- ³ NanoScience Center, Department of Physics, P.O. Box 35 (YN), FI-40014, University of Jyväskylä, Finland
- ⁴ Faculty of Medicine and Health Technology and BioMediTech, Tampere University, Arvo Ylpön katu 34, FI-33520 Tampere, Finland and Fimlab Laboratories, Biokatu 4, FI-33520 Tampere, Finland
- * Correspondence: tlemma@gmail.com (T.L.)

Highlights:

- High quality, i.e. sharp, reproducible, and excellent signal to noise ratio, SERS spectra of yeast microbes ((Saccharomyces cerevisiae (wild yeast), Dekkera bruxellensis and Wickerhamomyces anomalus obtained.
- SERS detection from very closely related fungal species (lysed and non-treated)
- Detection of the three species in complex beer medium
- High-resolution helium ion microscope (HIM) images give insight into the surface morphology of the yeast cells attached to AgNP
- SERS to be used for reliable detection and identification of the species in biosensing applications.