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Dynamic Magnetic and Optical Insight into a High Performance Pentagonal Bipyramidal Dy(III) Single-Ion Magnet

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Abstract: The pentagonal bipyramidal single-ion magnets (SIMs) are among the most attractive prototypes of high performance single-molecule magnets (SMMs). Here we introduced the fluorescence-active phosphine oxide ligand, combining the dynamic magnetic measurement, optical characterization and *ab initio* calculation, firstly studied the magneto-optical correlation of a high-performance pseudo- D_{5h} Dy(III) SIM with large $U_{\text{eff}} = 508(2)$ K and high magnetic hysteresis temperature of 19 K, $[\text{Dy}(\text{CyPh}_2\text{PO})_2(\text{H}_2\text{O})_5]\text{Br}_3 \cdot 2(\text{CyPh}_2\text{PO}) \cdot \text{EtOH} \cdot 3\text{H}_2\text{O}$ (CyPh_2PO = cyclohexyl(diphenyl)phosphine oxide). This work shall provide a deeper insight into the rational design of promising molecular magnets.

Introduction

Single-molecule magnets have attracted strong interest in recent years, as it can be trapped in one of the bistable magnetic states that separated by an energy barrier. Among the most promising candidates for high-density information storage, quantum processing and spintronics,^[1] a good SMM shall combine the large energy barrier, high blocking temperature and weak quantum tunneling of magnetization (QTM). The early cases of polymetallic clusters of transition metals show highly suppressed QTM,^[2] but the increasing of the energy barrier hits the bottleneck as the difficulty of combining the large spin states and

high anisotropies together. Lanthanide metal ions, especially Tb(III), Dy(III), Ho(III), Er(III)^[3] with intrinsic large spin-orbit interactions, have been extensively studied recently for SMMs, and the radical-bridged dilanthanide complexes achieved a high blocking temperature up to 14 K.^[4]

For most of the high performance monometallic lanthanide SMMs, a smart strategy is to stabilize some typical coordination symmetries for reducing the off-diagonal elements of orbit-lattice interaction matrix and therefore, suppressing QTM.^[3b, 3e, 5] Based on crystal-field theory,^[6] besides square antiprismatic complexes toward D_{4d} symmetry^[3b, 5e-f, 7] and linear 2-coordinated complexes^[5a-c] / sandwich-type complexes toward $D_{\infty h}$ symmetry,^[3d, 8] pentagonal bipyramidal complexes are on the way of getting close to D_{5h} symmetry that suppresses quantum tunneling and exhibits record anisotropic barriers. These facts illustrate that pseudo- D_{5h} -symmetry SIMs are highly attractive, and it is worth deeply understanding the magnetic dynamics and energy structures.^[9] Although the strict 5-fold symmetry is forbidden in conventional crystals, researchers have been trying to push the local symmetry to the limit.^[9-10]

Recently, we investigated the pentagonal bipyramidal Dy(III) SIMs which comprise a highly symmetrical $[\text{Dy}(\text{Cy}_3\text{PO})_2(\text{H}_2\text{O})_5]^{3+}$ (Cy_3PO = tricyclohexyl phosphine oxide) core.^[9c] The strong axial crystal field provided by the phosphine oxide, combined with the weak equatorial ligands arranged in pseudo-5-fold symmetry, lead to the energy barriers around 500 K and a record-breaking magnetic hysteresis temperature up to 20 K for the Br derivative. To the best of our knowledge, it is also the first report on the Ln-SIMs based on phosphine oxide, which is undoubtedly an emerging and promising family in this area and requires deeper understanding on their multifaceted behaviors and magneto-structural correlations.

Based on the previous study, we introduced the aromatic substituent groups into the system as sensitive antennas to promote the luminescence of Dy(III) ions. Indeed, the fluorescence spectra of rare earth ions can provide extremely valuable information on the energy levels, which is a direct probe and also a solid support to the magnetic studies and the *ab initio* calculations. The title complex here, $[\text{Dy}(\text{CyPh}_2\text{PO})_2(\text{H}_2\text{O})_5]\text{Br}_3 \cdot 2(\text{CyPh}_2\text{PO}) \cdot \text{EtOH} \cdot 3\text{H}_2\text{O}$ (**1**, CyPh_2PO = cyclohexyl(diphenyl)phosphine oxide), comprise a $[\text{Dy}(\text{R}_3\text{PO})_2(\text{H}_2\text{O})_5]^{3+}$ analogue core that retains the nice SMM properties. In addition, the advantages of **1** compare with its predecessor includes: 1) A further increased symmetry by a C_2 axis through the Dy(III) ion; 2) The feasibility to obtain suitable single crystals for micro-SQUID measurement; 3) The introduction of aromatic substituent groups to yield prominent fluorescence spectra for luminescence study. By a

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comprehensive study of crystallography, dc/ac magnetic characterization, single crystal micro-SQUID measurement and luminescence study, we managed to have a deeper understanding of the magneto-structural correlation, the relaxation dynamics and the QTM mechanism of SMMs. These combined with the state-of-the-art *ab initio* calculation shall shed light onto the further study on SMMs.

Results and Discussion

Crystal Structure

The title compound can be synthesized from DyBr_3 and CyPh_2PO in the $\text{H}_2\text{O}/\text{EtOH}$ mixed solution followed by recrystallization in EtOH and slow evaporation in ambient condition (see experimental section for details). Single-crystal crystallography shows that **1** crystallizes in the orthorhombic space group $C22_1$ and each asymmetric unit contains one Dy^{3+} ion, two coordinated CyPh_2PO ligands in the opposite direction, five coordinated waters in the equatorial plane, two free ligand, three Br ions in the outer coordination spheres and two EtOH molecules (Fig. 1). An additional symmetry of a C_2 axis (compared with the Cy_3PO analogue) through the Dy1-O1W provides the crystallographic guarantee for the equal bond lengths and angles for the main anisotropy axis along O1-Dy-O1A , which can be regarded as a step further toward the ideal pentagonal bipyramid geometry. Indeed, the value of Continuous Shape Measures calculations (CShM) is 0.174, which is only slightly larger than the most perfect pseudo- D_{5h} symmetry Ln-SIM (CShM = 0.142, $[\text{Dy}(\text{Cy}_3\text{PO})_2(\text{H}_2\text{O})_5]\text{Br}_3 \cdot 2(\text{Cy}_3\text{PO}) \cdot 2\text{H}_2\text{O} \cdot 2\text{EtOH}$)^[9c] reported to date (Table S2). Nevertheless, the actual symmetry for **1** is only C_2 .

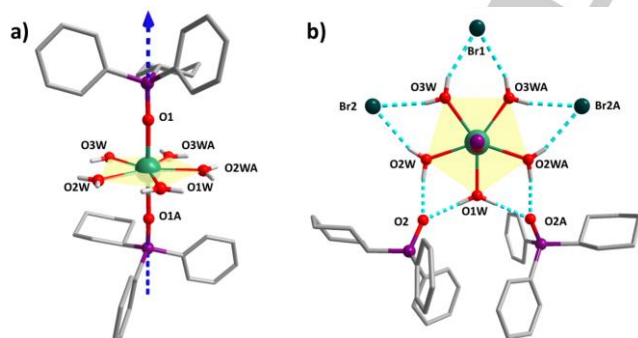


Figure 1. Crystal Structures of **1** emphasizing the pseudo-five-fold symmetry. **a)**: Coordination environment of Dy(III) in **1**. The blue dash line represents the orientation of the main magnetic axes of the lowest Kramers doublets. **b)**: Outer coordination sphere connected with hydrogen bonds in **1**. H atoms of CyPh_2PO are omitted for clarity. Color Codes: Dy, green; P, purple; Br, cyan; O, red; C, gray; H, light gray. Symmetry code (A): 1-x, y, 0.5-z.

For the outer coordination sphere, a five-pointed star is formed by hydrogen bonds between the five coordinated H_2O molecules, three Br ions and two free ligands (Fig. 1b), which helps the stabilization of the pseudo- D_{5h} symmetry of the first coordination sphere. Overall the local environment of Dy^{3+} in **1** is

a highly compressed pentagonal bipyramid with the average Dy-O distances of 2.217 Å (axial) and 2.364 Å (equatorial). Face-to-face and edge-to-face π - π stacking are formed between the benzyl rings of the coordinated ligands and the uncoordinated ones. Between the discrete molecules, the nearest Dy-Dy distance in the crystal structure is as far as 13.8 Å and there is no direct or super-exchange interactions through chemical bonds.

Magnetic Characterization

Variable-temperature magnetic susceptibilities were measured on polycrystalline samples of **1** under a 0.1 T dc field (Fig. 2a). At room temperature, the $\chi_M T$ value is $13.9 \text{ cm}^3 \text{ K mol}^{-1}$, a little lower than the expected value for a free Dy^{3+} ion ($14.17 \text{ cm}^3 \text{ K mol}^{-1}$). Upon cooling, $\chi_M T$ gradually decrease and then suddenly drop below ca. 12 K. Meanwhile, the zero-field-cooled / field-cooled (ZFC-FC) magnetization under a 0.1 T in warm mode (2 K/min) (inset of Fig. 2a) shows a clear divergence below 11 K with a peak at 10.5 K, indicating the process into magnetic blocking. The approaching magnetization values to $5 N\beta$ and the overlapped M vs. H/T curves (Fig. S5) provide an evidence of very far-separated excited doublets for Dy^{3+} .

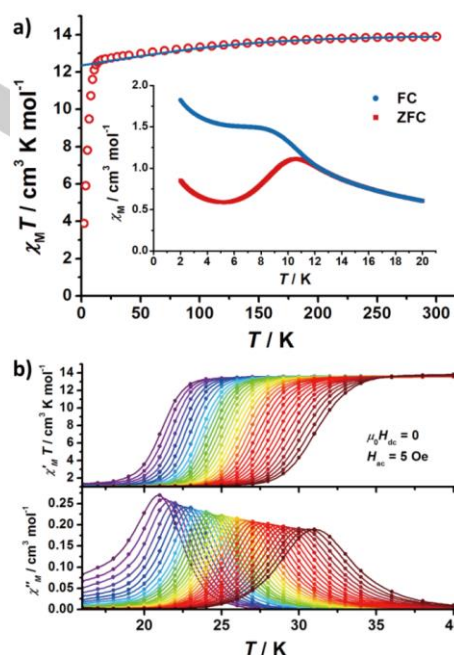


Figure 2. **a)** Temperature dependence of the molar magnetic susceptibility $\chi_M T$ products and the zero-field-cooled / field-cooled (ZFC-FC) magnetic susceptibilities (inset) for **1** under a 0.1 T dc field sweeping at 2 K/min in warming mode. The solid line corresponds to the *ab initio* calculations. **b)** Temperature dependence of the in-phase $\chi'_M T$ product and out-of-phase χ''_M for **1** in zero dc field with the ac frequency of 1-1488 Hz.

To reveal the dynamic magnetic behaviors, variable-temperature and variable-frequency ac magnetic susceptibilities were measured in zero dc field and a 0.1 T dc field. From the temperature-dependent data (Fig. 2b), the patterns are quite

typical for SMMs, with the 1488 Hz peak located at 31 K down to the 1 Hz peak located at 21 K. No overwhelming QTM can be observed on the ac data in the measured window.

For the frequency-dependent measurement (Fig. 3), we pushed the low-frequency limit to 0.1 Hz and the ac peaks still keep shifting towards low frequency region. Moreover, the ac susceptibilities are almost unaffected by the application of a 0.1 T dc field (Fig. 3b). For the high temperature region, the temperature dependence of the relaxation time τ perfectly obeys the Arrhenius law corresponding to the Orbach process, and the best fit gives the $U_{\text{eff}} = 508(2)$ K, $\tau_0 = 8.6 \times 10^{-12}$ s for zero dc field and $U_{\text{eff}} = 503(4)$ K, $\tau_0 = 9.5 \times 10^{-12}$ s for 0.1 T dc field (Fig. 4a).

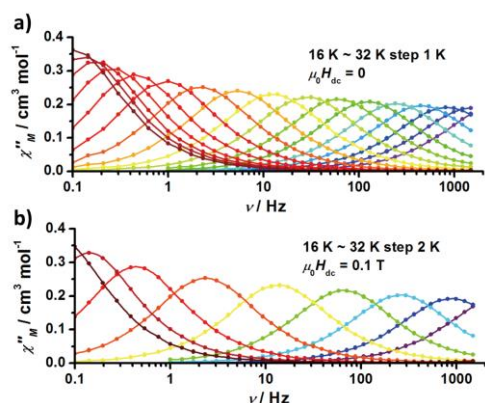


Figure 3. Frequency dependence of the out-of-phase χ''_M in zero dc field (a) and 0.1 T dc field (b) for **1**. Lines are guides to the eyes.

In order to investigate the slow relaxation behavior at lower temperatures, we managed to extend the measured relaxation time using dc magnetization decay. The relaxation dynamics can be fitted with the multiple relaxation equation $\tau^{-1} = \tau_0^{-1} \exp(-U_{\text{eff}}/k_B T) + C T^n + \tau_{\text{QTM}}^{-1}$ involving a Raman process with $n = 4.0(2)$ and slowly goes to a platform with $\tau_{\text{QTM}} = 19.5(8)$ s for zero dc field. When the final dc field is set as 0.1 T, the relaxation times keeps rising with $n = 5.61(4)$ and goes up to 1775 s at 5 K, showing the suppression of QTM (Fig. 4a).

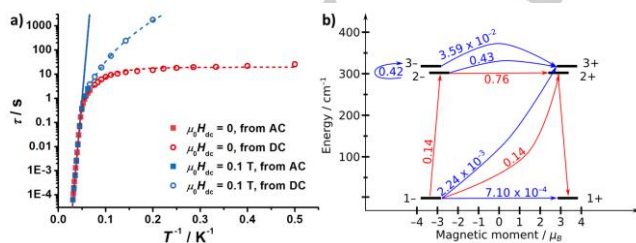


Figure 4. a) Temperature dependence of the relaxation time τ in zero dc field (red) and a 0.1 T Oe dc field (blue) for **1**. The solid lines are the best fits to the Arrhenius law, while the dashed lines are the best fit to the multiple relaxation equation. b) Magnetization blocking barrier of **1**. The lowest spin-orbit states are arranged according to the magnitude of their magnetic moments on the horizontal axis. The numbers next to arrows connecting two states display the

average transition magnetic moment matrix element between the respective states. The most probable relaxation route is indicated by red arrows and takes place via the first excited Kramers doublet.

Magnetic hysteresis loops can be observed as the extremely slow magnetic relaxation rate at low temperatures, even on a conventional PPMS-VSM (Fig. 5). The butterfly shape is typical, due to the faster relaxation around zero field and the major reasons are the unsuppressed QTM from: 1) the deviation of the symmetry; 2) the hyperfine interaction and 3) the dipolar interactions. The hysteresis loops become narrower with the increase of temperature, and the remanent magnetization (M_r) and coercive field (H_c) also become smaller (Fig. S7). Conservatively, we consider the hysteresis temperature to be 19 K with the field ramping speed of 0.02 T/s (In higher temperatures the measurement falls in instrumental error when $H_c < 0.01$ T.), which is comparable with the Cy_3PO derivative^[9c] and higher than that of $[\text{Dy}(\text{bbpen})\text{Br}]$.^[9d] It should be noted that the hysteresis temperature is usually different from the peak temperature of ZFC susceptibilities. Indeed, there will not be a strict definition of blocking temperatures for SMMs but depends on different measurements, as the slow relaxation of magnetization is not a real “blocking” like those in the phase changing materials. Here we also notice that some of the extremely high energy barriers do not guarantee the proportional increase of the relaxation times,^[9d, 10b-c] as the relaxation is already dominated by Raman process in lower temperature region.

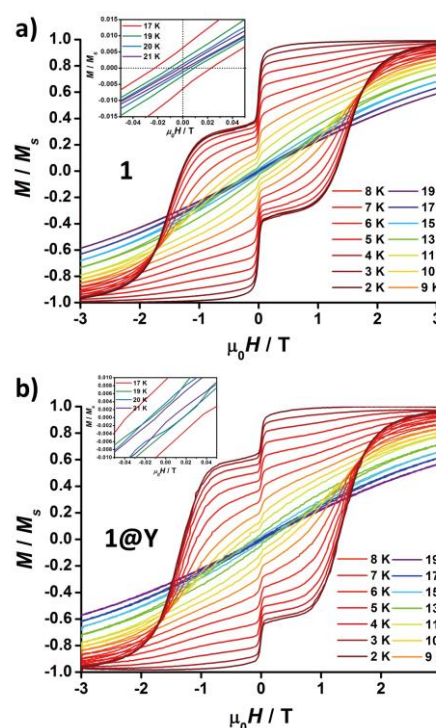


Figure 5. Normalized magnetic hysteresis loops for **1** (a) and **1@Y** (b). The data were continuously collected at intervals of 1 s with the field ramping speed of 0.02 T/s at various temperatures.

Further study of the diluted sample **1@Y** does not show much difference from **1**. Fitting the ac susceptibilities yield the consistent $U_{\text{eff}} = 467(5)$ K, $\tau_0 = 2.4 \times 10^{-11}$ s for zero dc field and $U_{\text{eff}} = 465(5)$ K, $\tau_0 = 2.4 \times 10^{-11}$ s for a 0.1 T dc field (Fig. S9-11). The divergence of ZFC-FC curves is also observed below 11 K with a peak at 10.5 K (Fig. S12). The remanent magnetization (M_r) becomes larger after the dilution (Fig. 5b), which be attributed to the weakening of intermetallic magnetic interactions, though the magnetic hysteresis loops still keep open up to 19 K.

Micro-SQUID Measurement

To extend the characterization down to ultra-low temperature, micro-SQUID measurements were performed on a single-crystal of **1**. The hysteresis loops at 4 K show the similar shapes as the powder sample (Fig. 6a), which become even more prominent at 0.03 K (Fig. 6b). For the decay of the DC magnetization at zero applied field, the single-crystal measurements should be comparable to the powder measurements and the relaxation time obtained from these two measurements should be converged in the same temperature regime. As illustrated in the plots (Fig. 6d), although the two datasets follow in the same trend, they could be converged only with a multiplying factor of 1.7. Such a difference could be due to the fact that there is no way to set the field exactly as zero in conventional SQUID and the remanent field always exists.

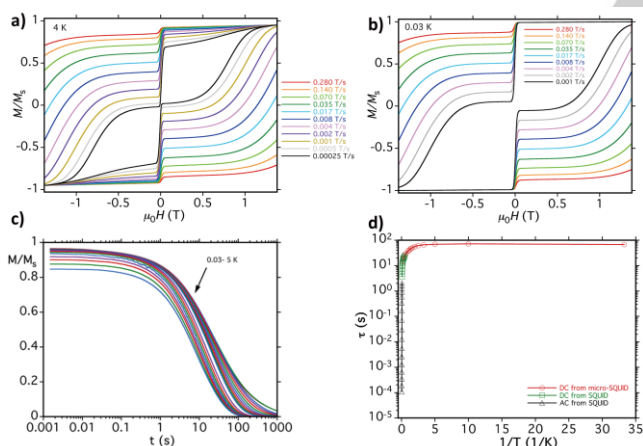


Figure 6. Dynamic magnetic properties for **1** measured on micro-SQUID: **a)** Normalized magnetic hysteresis loops at 4 K; **b)** Normalized magnetic hysteresis loops at 0.03 K; **c)** The decay of the normalized DC magnetization at various temperatures. **d)** Temperature dependence of the relaxation time τ compared with the AC (black) and DC (green) data from SQUID.

Luminescence Study

The introduction of aromatic substituent groups in **1** leads to nice fluorescence spectra both at room temperature and at low temperature, which provides much more valuable information on the energy levels. In general, the emission spectrum of **1** shows the characteristic peaks for Dy(III) with three groups of sharp peaks corresponding to the f-f transition around 475 nm (21000 cm^{-1} , ${}^4F_{9/2} \rightarrow {}^6H_{15/2}$), 575 nm (17400 cm^{-1} , ${}^4F_{9/2} \rightarrow {}^6H_{13/2}$) and 670 nm (15000 cm^{-1} , ${}^4F_{9/2} \rightarrow {}^6H_{11/2}$). (Fig. S13). Here we focus on the

${}^4F_{9/2} \leftrightarrow {}^6H_{15/2}$ transitions around 21000 cm^{-1} involving the lowest-lying energy levels that are crucial to the magnetic dynamics (Fig. 7).

At room temperature, it seems difficult to recognize the multiple peaks. Then, on lowering the temperature, the ${}^6H_{15/2} \rightarrow {}^4F_{9/2}$ absorption and the ${}^6H_{15/2} \rightarrow {}^4F_{9/2} \rightarrow {}^6H_{13/2}$ excitation spectra become sharper, and we noticed that the peaks at 20983 cm^{-1} (No. 2) and 20914 cm^{-1} (No. 3) disappear, indicating the depopulation of excited states. At 10 K, the last peak of the absorption and excitation spectra overlap perfectly with the first peak of the emission spectra at 21308 cm^{-1} (No. 1), which is assigned as the ground-to-ground transition of ${}^4F_{9/2} \leftrightarrow {}^6H_{15/2}$ (red arrow in Fig. 7b). Then, the eight Kramers doublets from the splitting of ${}^6H_{15/2}$ multiplet can be successively determined from the fine structure of the ${}^4F_{9/2} \rightarrow {}^6H_{15/2}$ emission. Meanwhile, we believe the huge intensity difference of the emission peaks also correspond to the extent of mixing of the wave functions, as these transitions are intrinsically forbidden while the selection rules can be broken by lanthanide-crystal field interactions.

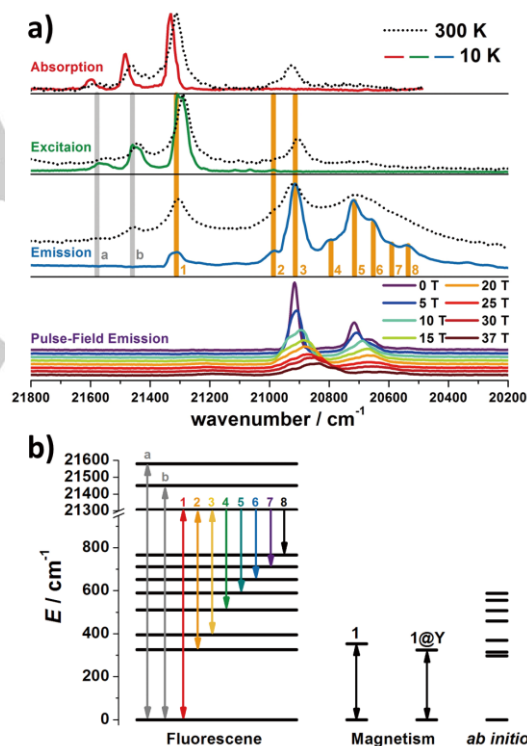


Figure 7. a) Fluorescence Spectra of **1** showing the ${}^4F_{9/2} \leftrightarrow {}^6H_{15/2}$ peaks. The absorption, excitation ($\lambda_{\text{em}} = 574 \text{ nm}$) and emission ($\lambda_{\text{ex}} = 361 \text{ nm}$) spectra were measured at 300 K (black dotted lines) and 10 K (color solid lines), and the pulse-field emission were measured at 5 K. Vertical lines indicate the optical transitions between the microstates of ${}^4F_{9/2}$ and ${}^6H_{15/2}$ and the emission spectra are plotted in logarithmic intensity for clarity. **b)** Energy levels determined by the fluorescence Spectra of **1** compared with the magnetic blocking barrier and ab initio calculation results. Note the higher energy transitions (grey arrows) involve the excited states of the ${}^4F_{9/2}$ state.

We also performed the photoluminescence measurement under a pulsed high magnetic field up to 37 T to probe its magnetic response.^[12] Although several weak peaks cannot be observed, the pulse-field emission still matches well with the conventional spectra. Significant effect of the Zeeman splitting is observed with the general shift towards lower wavenumber (owing to the ground-doublet splitting of the $^4F_{9/2}$ state) and the further splitting into pairs of peaks within the $^6H_{15/2}$ state. We notice that the asymmetric character around 20900 cm^{-1} might indicate the overlapping of multiple peaks with different center positions and Zeeman splitting, which is agree with the close lying of the first and second excited state (No.2-3). Also, the small splitting of the strongest peak at 20914 cm^{-1} (No. 3) rules out its possibility to involve the ground doublet which is with the largest g_z (see calculation for details). Here, the energy gaps from the ground doublet to these two excited doublets are 322 cm^{-1} and 396 cm^{-1} , respectively, matching well with the spin-reversal barrier of 353 cm^{-1} for **1** and 324 cm^{-1} for **1@Y** determined by the magnetic measurements.

Ab Initio Calculation

Further insight into the electronic structure and magnetic properties of **1** was obtained by quantum chemistry calculations. The geometry of **1** was extracted from the crystal structure and the positions of hydrogen atoms were optimized at DFT level. This was done because the coordinates of the hydrogen atoms refined from the diffraction data are usually not very accurate and the hydrogens of the coordinated water molecules lie close to the Dy ion and may therefore have an appreciable effect on the crystal field. The optimization retains the nearly planar pseudo- D_{5h} arrangement of the five water molecules. The structure used in the DFT and *ab initio* calculations included, in addition to the complex **1**, the outer coordination sphere containing the three bromide ions and the two free ligands hydrogen bonded into the coordinated water molecules (Fig. 1). Calculations were also conducted on various truncated structures but the results show that any simplification of the structure leads to qualitative changes in the energy spectrum and the magnetic properties.

Table 1. Calculated g -tensors of the eight lowest Kramers doublets arising from the crystal field splitting of the $^6H_{15/2}$ manifold

KD	g_x	g_y	g_z
1	0.0002	0.0002	19.8812
2	0.9571	3.4268	16.6964
3	0.0881	0.1268	16.9229
4	3.3031	4.5458	8.7070
5	1.4008	3.7771	6.5247
6	0.2470	0.3874	13.6266
7	0.2963	1.2140	11.2279
8	0.5039	1.0113	13.1702

The calculated g -tensors of the eight lowest Kramers doublet are listed in Table 1. The ground doublet is highly axial ($g_x = 0.0002$, $g_y = 0.0002$, $g_z = 19.8812$) explaining the prominent SMM behavior of **1** at low temperatures. The orientations of the

main anisotropy axes (g_z) of the three lowest Kramers doublets are illustrated in Fig. 1a. The axis of the ground doublet is aligned roughly along the main rotational axis of the pseudo- D_{5h} symmetry around the Dy ion and deviates from the two axial Dy–O bond vectors by an angle of 2.9°. The main axis of the first excited Kramers doublet lies along the C_2 axis of the crystal structure and is near perfectly perpendicular to the ground axis with an angle of 90.0° between them. The third axis lies again almost parallel to the main rotational axis with an angle of 0.3° between it and the ground axis.

To elucidate the effects of the electronic structure on the relaxation of magnetization in **1** the magnetization blocking barrier was also studied using previously proposed methodology.^[13a] The results are summarized in Fig. 4b and show that the most probable relaxation pathway takes place via the first excited Kramers doublet. The local pseudo- D_{5h} symmetry around the Dy ion leads to suppression of the QTM process indicated by the relatively small value of the matrix element connecting the two components of the ground doublet. However, the lower symmetry introduced by the outer coordination sphere leads to rotation of the main magnetic axis of the first excited doublet into an orientation parallel to the true C_2 axis of the crystal structure which lies perpendicular to the ground axis. As previously demonstrated, the large angle between the two Kramers doublets leads to large transverse magnetic matrix elements between the components of the respective doublets.^[13b] This allows the magnetization to traverse through the barrier and leads to lowering of the blocking barrier as compared to an analogous complex in ideal D_{5h} symmetry.

Although the properties of Kramers doublets – such as the g -tensors and transition magnetic moments – are well reproduced at CASSCF level, quantitative comparison of the calculated energy levels with those obtained from the luminescence spectrum require explicit treatment of dynamic correlation. This can be achieved by applying a second order correction to the CASSCF energies using the extended multistate (XMS) CASPT2 method.^[13c] Treatment of the full molecular structure, including the outer coordination sphere, at this level is not feasible in terms of computational costs and simplification of the structure may result into qualitative differences in the calculated energetics. Thus, dynamic correlation effects were treated as a correction to the *ab initio* crystal field potential of the $^6H_{15/2}$ ground multiplet (see the supporting information for detailed description of the procedure) and a corrected crystal field Hamiltonian was diagonalized to obtain the final spectrum. The crystal field corrected energies of the eight lowest Kramers doublets are presented in Table S4 along with the respective values calculated at CASSCF level for the full geometry and at CASSCF and XMS-CASPT2 levels for a truncated geometry where the phenyl and cyclohexyl groups of the ligands have been replaced with methyl groups. The results clearly show that dynamic correlation and the full structure are required to approach the experimental levels.

The energy of the first excited Kramers doublet calculated using the crystal field correction approach is 297 cm^{-1} which is in good agreement with the energy difference of 322 cm^{-1} obtained

from the luminescence spectrum. The value is also in semi-quantitative agreement with the experimental barrier heights 353 cm^{-1} for **1** and 324 cm^{-1} for **1@Y** from magnetism. Agreement with the higher excited states is poorer. This deviation most likely results from the approximate nature of the crystal field correction procedure employed in this study. We expect that enlarging the active space of the CASSCF method by explicitly correlating several ligand type orbitals and/or metal's occupied and virtual orbitals would increase the quality of the reference wave function, leading to an even closer agreement for the excited states. This was recently demonstrated for Er-trensal complex,^[14] for which a larger active space increased the overall agreement for each of the crystal field levels of the ground $J = 15/2$ multiplet, as well as for their properties.

Conclusions

In conclusion, while preserving the large energy barrier and high hysteresis temperature, we successfully performed the single-crystal micro-SQUID measurement at ultra-low temperature and introduced the characteristic lanthanide photoluminescence into the family of pentagonal bipyramidal Dy(III) single-ion magnets based on phosphine oxide. The AC, DC and micro-SQUID relaxation data follow in the same trend and the hysteresis loops become wider when going down to 0.04 K. The fluorescence spectra provide crucial information of the energy levels that correspond well with the magnetic measurements and shows semi-quantitative agreement with the *ab initio* calculation. It should be pointed out that the magnetic blocking in many of these SMMs with high energy barriers is controlled by the Raman process rather than the Orbach process, which requires deeper understanding in the future. Apart from this, we also note that the fast QTM around zero field sets up a limit for the further increase of the relaxation time. Efforts regarding this could be focus on the fine-tuning of the magnetic interactions and explore the lanthanide nuclides of suitable nuclear spins into an SIM/SMM system.

Experimental Section

General Procedure All reagents were commercially available and used as received without further purification. The elemental analyses were performed with an Elementar Vario-EL CHN elemental analyzer. The IR spectra were recorded using a Thermo Nicolet AVATAR 330 FT-IR spectrometer. The powder XRD patterns were recorded on a Bruker D8 X-Ray diffractometer with Cu K α radiation. Thermogravimetric analysis (TGA) was carried out on a NETZSCH TG209F3 thermogravimetric analyzer. The ICP-AES analyses were performed with a TJA IRIS (HR) spectrometer.

Synthesis. $[\text{Dy}(\text{CyPh}_2\text{PO})_2(\text{H}_2\text{O})_5]\text{Br}_3 \cdot 2(\text{CyPh}_2\text{PO}) \cdot \text{EtOH} \cdot 3\text{H}_2\text{O}$ (**1**) are synthesized by dissolving Hydrate DyBr_3 (0.2 mmol) and CyPh_2PO (0.4 mmol) in a $\text{H}_2\text{O}/\text{EtOH}$ (1:9, 4 mL) mixed solution, followed by the slow evaporation in ambient condition to almost dryness. Then 2 mL EtOH was added for recrystallization. Colorless crystals suitable for X-ray analysis began to grow after 3 days, which are then collected by filtration, washed with cold EtOH and then dried in air (yield ca. 40%). The

disordered solvent molecules in the voids were unable to be determined by single-crystal diffraction but confirmed via elemental analysis and thermogravimetric analysis. Analysis (calcd., found) for $\text{C}_{76}\text{H}_{112}\text{P}_4\text{Br}_3\text{DyO}_{14}$ (**1**): C (51.40, 51.47), H (6.35, 6.02), IR(cm^{-1}): 3275br, 2933s, 2852s, 1632m, 1485m, 1439s, 1176s, 1135s, 1120s, 1095s, 999m, 887m, 746s, 725s, 700s, 557s, 536s. The diluted samples **1@Y** was synthesized in the same way with $\text{DyBr}_3:\text{YBr}_3 = 1:19$. The dilution ratios were confirmed by both the magnetization (ca. 4%) and ICP-AES analyses ($5\% \pm 0.5\%$), and the phase purity was checked by the powder XRD. Analysis (calcd., found) for $\text{C}_{76}\text{H}_{112}\text{P}_4\text{Br}_3\text{Y}_{0.95}\text{Dy}_{0.05}\text{O}_{14}$ (**1@Y**): C (53.51, 53.61), H (6.62, 6.21), IR(cm^{-1}): 3256br, 2933s, 2852s, 1633m, 1487m, 1439s, 1144s, 1121s, 1092s, 999m, 889m, 825m, 747s, 727s, 698s, 557s, 534s.

X-ray Crystallography Single-crystal diffraction data was recorded at 150(2) K on a Rigaku R-Axis SPIDER Image Plate diffractometer with Mo K α radiation, solved by direct methods and refined using SHELXTL program.^[15] Crystal Data and Structural Refinement are listed in Supplementary Table S1-S3. CCDC 1519258-1519259 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Magnetic Measurements The dc and ac magnetic measurements were performed on the polycrystalline samples using a Quantum Design MPMS XL-7 SQUID magnetometer. The ZFC-FC (0.1 T, 2 K/min), hysteresis (0.02 T / s) and relaxation were measured on a Quantum Design PPMS with VSM option. The single-crystal measurements down to ultra-low temperature were performed on a micro-SQUID.^[16] Diamagnetic correction was performed based on the Pascal's coefficients. In the lower temperature region, the relaxation rate is determined by measuring the dc magnetization decay after the removal of a saturated field of 3 T.

Optical Measurements The luminescence spectra were recorded on an Edinburgh FLS-980 Fluorescence spectrometer equipped with Xenon light, PMT detector and ALS cryostat down to 10 K. The slit and step were set to 0.1 nm to capture the detailed splitting of the peaks. The pulsed high magnetic field photoluminescence measurement was performed at Wuhan National High Magnetic Center with a laser-fiber spectrometer setup operating at 5 K. The magnitude of magnetic field is detected at the falling side of a magnetic pulse and each luminescence spectrum was captured within 1 ms.

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Keywords: lanthanide • dysprosium • magnetic properties • single-molecule magnets • Fluorescence

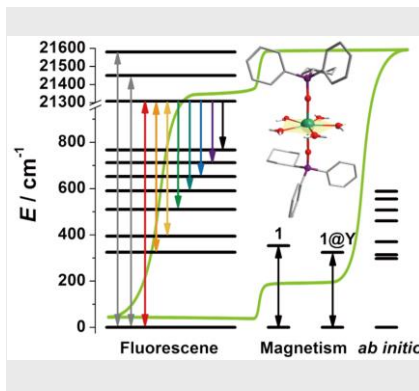
- [1] a) D. Gatteschi, R. Sessoli, J. Villain, *Molecular nanomagnets*, Oxford University Press, **2006**; b) M. N. Leuenberger, D. Loss, *Nature* **2001**, *410*, 789-793; c) W. Wernsdorfer, *Science* **1999**, *284*, 133-135; d) L. Bogani, W. Wernsdorfer, *Nat. Mater.* **2008**, *7*, 179-186; e) M. Mannini, F. Pineider, P. Sainctavit, C. Danieli, E. Otero, C. Sciancalepore, A. M. Talarico, M.-A. Arrio, A. Cornia, D. Gatteschi, R. Sessoli, *Nat. Mater.* **2009**, *8*, 194-197; f) M. Ganzhorn, S. Klyatskaya, M. Ruben, W. Wernsdorfer, *Nat. Nano.* **2013**, *8*, 165-169.

- [2] R. Sessoli, D. Gatteschi, A. Caneschi, M. A. Novak, *Nature* **1993**, 365, 141-143.
- [3] a) S. Osa, T. Kido, N. Matsumoto, N. Re, A. Pochaba, J. Mrozinski, *J. Am. Chem. Soc.* **2003**, 126, 420-421; b) N. Ishikawa, M. Sugita, T. Ishikawa, S.-y. Koshihara, Y. Kaizu, *J. Am. Chem. Soc.* **2003**, 125, 8694-8695; c) D. N. Woodruff, R. E. P. Winpenny, R. A. Layfield, *Chem. Rev.* **2013**, 113, 5110-5148; d) R. J. Blagg, L. Ungur, F. Tuna, J. Speak, P. Comar, D. Collison, W. Wernsdorfer, E. J. L. McInnes, L. F. Chibotaru, R. E. P. Winpenny, *Nat. Chem.* **2013**, 5, 673-678; e) S.-D. Jiang, B.-W. Wang, H.-L. Sun, Z.-M. Wang, S. Gao, *J. Am. Chem. Soc.* **2011**, 133, 4730-4733; f) S. K. Langley, D. P. Wielechowski, V. Vieru, N. F. Chilton, B. Moubaraki, B. F. Abrahams, L. F. Chibotaru, K. S. Murray, *Angew. Chem. Int. Ed.* **2013**, 52, 12014-12019.
- [4] a) J. D. Rinehart, M. Fang, W. J. Evans, J. R. Long, *Nat. Chem.* **2011**, 3, 538-542; b) J. D. Rinehart, M. Fang, W. J. Evans, J. R. Long, *J. Am. Chem. Soc.* **2011**, 133, 14236-14239; c) S. Demir, J. M. Zadrozny, M. Nippe, J. R. Long, *J. Am. Chem. Soc.* **2012**, 134, 18546-18549.
- [5] a) J. M. Zadrozny, D. J. Xiao, M. Atanasov, G. J. Long, F. Grandjean, F. Neese, J. R. Long, *Nat. Chem.* **2013**, 5, 577-581; b) J. M. Zadrozny, J. R. Long, *J. Am. Chem. Soc.* **2011**, 133, 20732-20734; c) R. C. Poulten, M. J. Page, A. G. Algarra, J. J. Le Roy, I. López, E. Carter, A. Llobet, S. A. Macgregor, M. F. Mahon, D. M. Murphy, M. Murugesu, M. K. Whittlesey, *J. Am. Chem. Soc.* **2013**, 135, 13640-13643; d) R. Westerström, J. Dreiser, C. Piamonteze, M. Muntwiler, S. Weyeneth, H. Brune, S. Rusponi, F. Nolting, A. Popov, S. Yang, L. Dunsch, T. Greber, *J. Am. Chem. Soc.* **2012**, 134, 9840-9843; e) S.-D. Jiang, B.-W. Wang, G. Su, Z.-M. Wang, S. Gao, *Angew. Chem. Int. Ed.* **2010**, 49, 7448-7451; f) N. F. Chilton, S. K. Langley, B. Moubaraki, A. Soncini, S. R. Batten, K. S. Murray, *Chem. Sci.* **2013**, 4, 1719-1730.
- [6] C. Görlner-Walrand, K. Binnemans, in *Handbook on the Physics and Chemistry of Rare Earths, Vol. Volume 23* (Eds.: Karl A. Gschneidner, Jr., E. LeRoy), Elsevier, **1996**, pp. 121-283.
- [7] M. A. Aldamen, J. M. Clemente-Juan, E. Coronado, C. Martí-Gastaldo, A. Gaita-Ariño, *J. Am. Chem. Soc.* **2008**, 130, 8874-8875.
- [8] a) K. R. Meihaus, J. R. Long, *J. Am. Chem. Soc.* **2013**, 135, 17952-17957. b) L. Ungur, J. J. Le Roy, I. Korobkov, M. Murugesu, L. F. Chibotaru, *Angew. Chem. Int. Ed.* **2014**, 53, 4413-4417.
- [9] a) J.-L. Liu, Y.-C. Chen, Y.-Z. Zheng, W.-Q. Lin, L. Ungur, W. Wernsdorfer, L. F. Chibotaru, M.-L. Tong, *Chem. Sci.* **2013**, 4, 3310-3316; b) J.-L. Liu, J.-Y. Wu, Y.-C. Chen, V. Mereacre, A. K. Powell, L. Ungur, L. F. Chibotaru, X.-M. Chen, M.-L. Tong, *Angew. Chem. Int. Ed.* **2014**, 53, 12966-12970; c) Y.-C. Chen, J.-L. Liu, L. Ungur, J. Liu, Q.-W. Li, L.-F. Wang, Z.-P. Ni, L. F. Chibotaru, X.-M. Chen, M.-L. Tong, *J. Am. Chem. Soc.* **2016**, 138, 2829-2837; d) J. Liu, Y.-C. Chen, J.-L. Liu, V. Vieru, L. Ungur, J.-H. Jia, L. F. Chibotaru, Y. Lan, W. Wernsdorfer, S. Gao, X.-M. Chen, M.-L. Tong, *J. Am. Chem. Soc.* **2016**, 138, 5441-5450.
- [10] a) S. Cardona-Serra, J. M. Clemente-Juan, E. Coronado, A. Gaita-Ariño, A. Camón, M. Evangelisti, F. Luis, M. J. Martínez-Pérez, J. Sesé, *J. Am. Chem. Soc.* **2012**, 134, 14982-14990; b) S. K. Gupta, T. Rajeshkumar, G. Rajaraman, R. Murugavel, *Chem. Sci.* **2016**, 7, 5181-5191; c) Y.-S. Ding, N. F. Chilton, R. E. P. Winpenny, Y.-Z. Zheng, *Angew. Chem. Int. Ed.* **2016**, DOI: 10.1002/anie.201609685.
- [11] M. Gregson, N. F. Chilton, A.-M. Ariciu, F. Tuna, I. F. Crowe, W. Lewis, A. J. Blake, D. Collison, E. J. L. McInnes, R. E. P. Winpenny, S. T. Liddle, *Chem. Sci.* **2016**, 7, 155-165.
- [12] Y. Bi, C. Chen, Y.-F. Zhao, Y.-Q. Zhang, S.-D. Jiang, B.-W. Wang, J.-B. Han, J.-L. Sun, Z.-Q. Bian, Z.-M. Wang, S. Gao, *Chem. Sci.* **2016**, 7, 5020-5031.
- [13] a) L. Ungur, M. Thewissen, J.-P. Costes, W. Wernsdorfer, L. F. Chibotaru, *Inorg. Chem.* **2013**, 52, 6328-6337; b) L. Ungur, L. F. Chibotaru, *Phys. Chem. Chem. Phys.* **2011**, 13, 20086-20090; c) T. Shiozaki, W. Györfy, P. Celani, H.-J. Werner, *J. Chem. Phys.* **2011**, 135, 081106.
- [14] L. Ungur, L. F. Chibotaru, *Chem. Eur. J.*, **2016**. DOI: 10.1002/chem.201605102
- [15] G. Sheldrick, *Acta Crystallographica Section A* **2008**, 64, 112-122.
- [16] W. Wernsdorfer, *Supercond. Sci. Technol.* **2009**, 22, 064013.

Entry for the Table of Contents

FULL PAPER

Dynamic magnetism and magneto-optical correlation are firstly studied for a pentagonal bipyramidal Dy(III) single-ion magnet based on fluorescence-active phosphine oxide with large $U_{\text{eff}} = 508(2)$ K and high magnetic hysteresis temperature of 19 K.



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Dynamic Magnetic and Optical Insight
into a High Performance Pentagonal
Bipyramidal Dy(III) Single-Ion Magnet